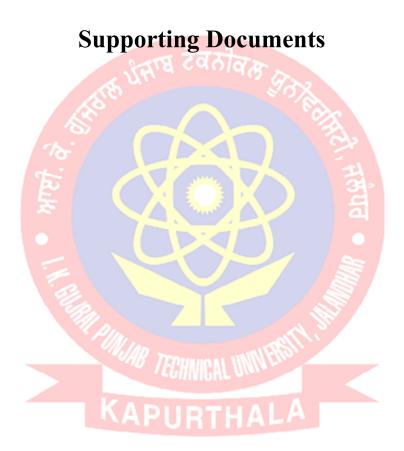
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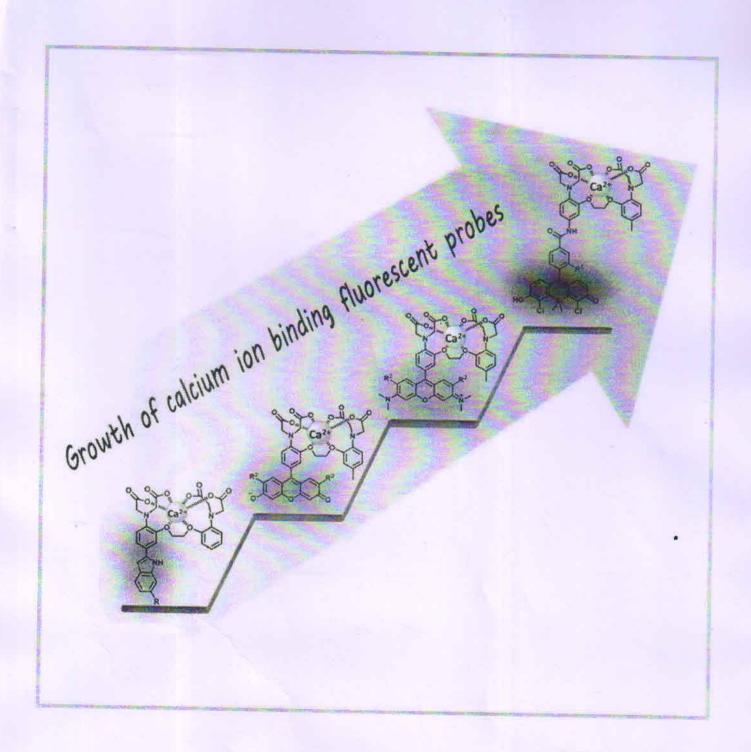
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Design and Applications of Small Molecular Probes for Calcium Detection

Roopa,*[a] Naresh Kumar,[b] Manoj Kumar,[c] and Vandana Bhalla*[c]



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PAPER



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Self-assembled vesicle and rod-like aggregates of functionalized perylene diimide: reaction-based near-IR intracellular fluorescent probe for selective detection of palladium;

Prabhpreet Singh,* Lalit Singh Mittal, Vanita Vanita, Kapil Kumar,* Amandeep Walia, Gaurav Bhargava^c and Subodh Kumar^a

Herein, we report design, synthesis, and self-assembly of perylene climide (PDI) based near-IR intracellular probe (PS-PDI). PS-PDI molecules undergo aggregation to form self-assembled nanospheres and nanorods morphology in THF: H₂O (1:1) and DMSO: H₂O (1:9), respectively. The nanospheres have an open hole on surface reminiscent of vesicle structure (with a diameter of internal void in the range of 20-25 nm) whereas lengths of the nanorods extended up to few um range. The Pd⁰ based depropargulation leads to de-aggregation of these PS-PDI aggregates into smaller sphenical aggregates as evidenced by DLS, SEM, and TEM studies. Interestingly, these aggregates of PS-PDI in solution show highly sensitive behavior in the presence of Pd^0 showing absorbance changes in NIR region ($k_{max} = 710$ nm) and quenching of emission at ž_{em} 630 nm (DMSO : H₂O, 1: 9) with a limit of detection = 6.6 × 10⁻⁹ M or at J_{em} 564 nm (THF H₂O, 1: 1) with limit of detection = 2.1 × 10⁻⁸ M. Time and concentration dependent kinetics profiles of PS-PDI aggregates revealed an impressive rate constant value of 0.4 s⁻¹ and 0.21 s⁻¹ (DMSO-H₂O, 1:9). respectively in fluorescence and UV-Vis spectroscopy using 2 × 10⁻⁵ M concentration of Pa^o. PS-PDI undergoes rapid internalization into HeLa cells with low cytotoxicity and was successfully used as an intracellular imaging reagent for Pd⁰ in live HeLa cells. For practical applications, we exploited these nanoaggregates of PS-PDI for estimation of Pd2+ in the presence of a NaBH4-PPh3 mixture, Pd0 in drug and environmental samples, and Pd2+ in urine samples with excellent selectivity and sensitivity

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Introduction

Perylene diimides (PDIs) are a well documented class of functional organic dyes with excellent electronic, optical, and redox properties.1-5 Due to these properties, PDIs may find applications in dyes and pigments industries,6 optoelectronic devices,7-6 and supramolecular chemistry. 10-20 The aggregation and optoelectronic properties of PDIs can be tailored through modification of its n-molecular structure with chemical groups 1,21,22 either or both at the imide or bay positions of the PDIs core. Due to the presence of a planar x-conjugated core, PDI has propensity for

aggregation in an aqueous medium via π-π stacking, hydrogen bonding, electrostatic, and solvation interactions. Many reports regarding modification of imide position of PDIs to generate selfassembled nanostructures are being published each year. 23,26 However, little attention has been paid to the self-assembly of bay substituted PDIs. 25,26 Probably, the substitutions at a bay position drastically increase the solubility of PDIs and if substituents are bulky in nature, the perylene backbone deviates from planarity to minimize the strain and this hinders the possibility of π - π stacking. Moreover, the application of self-assembled PDIs for molecular recognition of metal ions and anions in water via aggregation or de aggregation of PDIs has been scarcely reported due to the poor solubility of PDIs in aqueous medium. Earlier we reported (ionic) self-assembly of benzimidazolium, amino acids, and nucleobase-functionalized carbon nanotubes (CNTs)27-29 and now we envisage that ease in aggregation of perylene chromophore in aqueous medium could be exploited for molecular recognition by using stimuli-based aggregation-disaggregation of PDIs. In this context, we recently synthesized a PDI-based probe and judiciously utilized its aggregation behavior for detection of perchlorate anion in water and C6 glioma cells. 10

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[&]quot;Department of Chemistry, OGC Centre for Advanced Studies, Guru Nanak Dev University, Amritsur 143 (905, India.

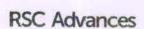
E-mail: prabhpreet chemingadu.ar.in: Tel: +91-84271-01534

Department of Human Genetics, Guru Nanak Dev University, Amrusar 143 005, India

^{*}Department of Applied Sciences, IK Outral Punjab Technical University,

Kapunhala-144601, Punjab, India

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PAPER



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A facile and chemoselective synthesis of 1,4benzodiazepin-2-ones and dienyl thiazolidin-4ones†

Bilash Kuila, *Yogesh Kumar, *Dinesh Mahajan, *Kapil Kumar, *Prabhpreet Singh and Gauray Bhargaya *

An efficient protocol for chemoselective synthesis of previously unknown 1.4-benzodiazepin-2-ones and dienyl thiazolidin-4-one carboxylates in excellent yields by ring transformation reactions of functionally decorated 2-azetidin-3-thiazolidin-4-ones is reported.

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β-Lactams are one of the fascinating heterocyclic systems in organic synthesis. The β-lactam unit is a fundamental structural characteristic in a significant class of antibiotics.1 Numerous lactam based drugs are commercially used for widespread applications. Literature survey clearly reveals that appropriate functionality around lactams diversified their biological activities.3 Apart from substitution dependent biological profile, functionalized β-lactams have been extensively explored as versatile synthons and precursors for various heterocyclic systems.1 The fi-lactam synthon methodology has become an important tool in organic synthesis.5 The key step in B-lactam synthon methodology involves the cleavage of the β-lactam ring. Of the four possible cleavage routes (Fig. 1), the N1-C2 bond cleavage (route-A) has been extensively explored and exploited for the synthesis of biologically relevant molecules such as amino esters, hexahydroisoquinolines, hexahydroisoquinolones, hexahydroisochromones, or thioxo-imidazolidines, imidazolidin-2-ones, piperazine-5,6-diones, dihydroimidazoles, (E) endo-aldonitrones, hexahydroisoquinolin-1-ylidene, and hydantoins, dihydroquino line-4-ones, 6,6,

On the other hand, 1,4-benzodiazepin-2-ones (BZD) are a very successful class of drugs and have been extensively employed as therapeuties since the early 1960s. BZD scaffolds are recurrent templates in the context of their wide range of biological activities such as sedatives, anxiolytics, anticonvulsants or muscle relaxants, anticonvulsants or muscle relaxants, anticumor, and antimalaria factor, and anti-HIV, and antitumor, and antimalaria factor.

activities. There have been significant revelations in the synthesis of C-3 functionalized BZD. 11-12 The substitutions at the C-3 position of BZD are crucial to the possible outcome for the potential biological activities. 10-12 Various BZD with methyl, 12-2 acetyl, naphthalyl, 12-2 and hydrocinnamide, 22-2 groups at its C-3 position have been explored for their applications in many areas of medicinal chemistry; therefore, underlying the importance for the development of new routes for the synthesis of BZD with diverse functionalities at the C-3 position.

As a part of our ongoing interest in the synthesis of a heterocyclic system via lactam synthon methodology, we earlier reported the synthesis and subsequent transformation of functionalized lactams for the synthesis of diaza-bicyclo[3,2,0] heptan-7-one, "amino-pyrrolidine esters" and (2-0x0-4-(styryl) azetidin-3-yl)pyridine. "In continuation of this, we herein document a conceptually simple approach that provides an integrated access to pharmacologically useful 1,4-benzodiazepin-2-one or dienyl-thiazolidin-4-one frameworks by careful chemoselective transformation of a bifunctional hybrid, viz., 3-{2-0x0-1-(aryl)-4-styrylazetidin-3-yl)thiazolidin-4-one by ring opening of 2-azetidinones via [N¹-C²] bond cleavage using sodium methoxide in methanol and subsequent transformation reactions.

3-Amino-2-azetidinones (1a-f) were prepared by reported methods.¹⁵ The variedly substituted 3-amino-2-azetidinones 1a-f used were characterized with the help of analytical data and spectral evidences. These 3-amino-2-azetidinones 1a-f were explored in three component cyclization reactions with variety



Fig. 1 Ring opening of the β-lactam ring.

Department of Applied Sciences, I. K. Gigral Punjab Technical University, Kapurthala, Punjab-144603, India. E-mail: gaurav@ptu.ac.in; Fax: +91-1822-255504; Tel: +91-1822-255504

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Department of Chemistry, Guru Nanak Dev University, Amritsar, Punjab 143005, India

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LETTER



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Cu(i) mediated Kinugasa reactions of α , β -unsaturated nitrones: a facile, diastereoselective route to 3-(hydroxy/bromo)methyl-1-aryl-4-(-styryl)azetidin-2-ones†

Yogesh Kumar, ^a Prabhpreet Singh ^b and Gaurav Bhargava* ^a

The manuscript describes a facile and highly diastereoselective synthesis of cis-3-(hydroxy/bromo)methyl-1-aryl-4-(-styryl)azetidin-2-ones via copper(i) mediated Kinugasa reactions of previously unexplored functionalized α,β-unsaturated nitrones. A variety of functionalized acetylenes and α,β-unsaturated nitrones were studied to yield cis-3-(hydroxy/bromo)methyl-1-aryl-4-(-styryl)azetidin-2-ones in good yields (8,2%).

β-Lactams are wickely recognized as one of the most significant heterocyclic scaffolds1 and have maintained a high level of curiosity, both in academia and in industry. The B-lactam ring is the common structural characteristic of a number of broad spectrum β-lactam antibiotics, such as penicillins, cephalosporins, carbapenents, nocardicins, monobactams, clavulanic acids as well as inhibitors of HIV-1 protease.2 In recent years, renewed interest has been paid to the synthesis and modification of the B-lactam ring to obtain compounds with different pharmacological activities such as cholesterol absorption inhibitory activity, human tryptase, thrombin and chymase inhibitory activity, vasopressin V1a antagonist activity, and antimalarial, antidiabetic, antitubercular, anti-inflammatory, anti-parkinsonian and anti-HIV activity.3.4 \(\beta\)-Lactams are also increasingly being used as useful synthons for the synthesis of a variety of natural products via β-lactam synthon methodology.5 The lactams have also been explored for synthesis of variety of non-protein amino acids, oligopeptides, peptidomimetics, and nitrogen-heterocycles,6 as well as biologically active natural and unnatural products of medicinal interest.8.

The common methodologies for the synthesis of functionalized [3-lactam include ketene-imine [2+2] cycloaddition reactions. 10 metallo-ester enolate-imine cyclocondensations. 11 However, these

traditional synthetic routes are mostly intolerant to acid or base sensitive moieties. After the advent of the Kinugasa reaction, functionalized lactams were synthesized which were previously inaccessible using traditional synthetic methodologies. ¹² The Kinugasa reactions use readily available substrates νiz , nitrones and acetylenes, and are mild and can tolerate different functional groups (Fig. 1). ¹³ The Kinugasa cascade provides access to numerous β -lactam compounds with a distinct configuration and potential biological activity. ¹⁴

An early investigation by different research groups on Kinugasa reactions examined the generation of functionalized β -lactams using aliphatic or aromatic substituted nitrones. These Kinugasa reactions are initiated by [3+2] cycloadditions of nitrones with Cu acetylide. These [3+2] cycloadditions of nitrones with Cu acetylide. These [3+2] cycloadditions (I) usually follow two different pathways: pathway A^{17} and pathway B^{12b} (Fig. 1). Both pathways A & B have revealed the presence of intermediates β -aminoketene (II) & oxaziridinium (III) respectively, which, eventually, collapse to cisitrans- β -lactams. The Kinugasa reactions generally afford β -lactams as mixtures of racemic diastereoisomers; however, various stereoselective variants of such transformations with appropriate chiral auxiliaries or chiral Cu(1) ligands have been reported (Fig. 2). 13,14

Although, in principle, the Kinugasa reaction would provide efficient access to styryl substituted lactam at the C-4 position, to the best of our knowledge, no example of such a process has been

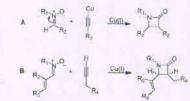


Fig. 1 (A) Previous work on Kinugasa reactions; (B) present work on Kinugasa reactions of $\alpha\beta$ -unsaturated nitrones.

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^{*}I. K. Gujral Punjab Technucal University, Kupurthala, Punjab-144603, India. E-mail: gaurav@pcu.ac.in; Fax: +91-1832-255504; Tel: +91-1822-255504

^bDepartment of Chemisery, Guru Nanak Dev University, Amritsar, Hanjab 143005, India

[†] Electronic supplementary information (ESI) available. See IXH: 10.1039/t6nj01747a

Highly chemo- and diastereo-selective synthesis of 2,6-diazabicyclo[3.2.0]heptan-7-ones, pyrrolidines and perhydroazirino[2,3-c]pyrroles

Yogesh Kumar, Bilash Kulia, Prabhpreet Singh, and Gauray Bhargaya**

^aDepartment of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, Punjab 144 603, India

^bDepartment of Chemistry, Guru Nanak Dev University, Amritsar, Punjab 143 005, India E-mail: gauray@ptn.ac.in

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Abstract

The manuscript describes a simple, convenient and metal-free diastereoselective synthesis of 4-halo-3-aryl/alkyl-6-aryl-2,6-diazabicyclo[3,2,0]heptan-7-ones via intramolecular endo-trig haloamination of 3-aminoazetidin-2-ones and its facile transformations to previously unknown methyl 4-halo-3-arylaminopyrrolidine-2-carboxylates and N-deprotected diazabicyclo[3,1,0]-hexane-2-carboxylic acids in good yields. The synthesis of such heterocyclic system is important in terms of the usefulness as organic synthon as well as their diverse pharmacological profiles.

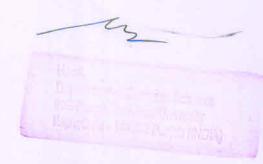
Keywords: Diazabicyclo[3.2.0]heptanones, pyrrolidines, aziridinopyrrolidines, β-lactams, endotrig haloamination

Introduction

Over the past decades, β -lactams have emerged as a useful synthon in organic chemistry. L2 Numerous researchers have explored the synthesis of a variety of novel heterocyclic systems via β -lactam synthon methodology. Ojima and his co-workers have described the crucial role of β -lactam synthon methodology in the synthesis of paclitaxel, docetaxel and new-generation taxoids viz. C-2- and C-3'-modified taxoids, etc. Alcaide et al. have utilized a variety of lactams as organic synthons for the construction of various alkaloid skeletons. Mahajan et al. have explored the β -lactam synthon approach towards the diastereoselective synthesis of functionalized octahydroisoquinolones, pyrroloxazine, the tetra/octahydro-isoquinoline and octahydroindole ring systems. Literature survey clearly reveals that β -lactams are important synthons for the synthesis of a variety of useful aza-heterocyclic systems. Functionalized proline esters, the five-membered azaheterocyclic systems, are important organocatalysts as well as having vital roles in

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REVIEW



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Recent developments in the synthesis of condensed $\beta\text{-lactams}$

Yogesh Kumar, Prabhpreet Singhb and Gaurav Bhargava*a

β-Lactams are important heterocycles with diverse pharmacological profiles and have emerged as useful organic synthons. Different variants of β-lactams condensed with carbo-/heterocycles, which have a broad range of antimicrobial activities and other useful pharmacological activities, justify a renewed interest in their synthetic developments. This review embraces an array of different reports on condensed β-lactams dating back to 2000. The aim of this review is to provide practicing and aspiring β-lactam chemists with a survey and flavor of the types of reactions used to prepare nonconventional condensed β-lactams, and an insight into why certain methodologies are advantageous under particular circumstances.

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Throughout much of the 20th century, β-lactam antibiotics have been a bastion as therapeutics for treating a wide range of bacterial infections. ^{1,2} β-Lactams show a broad spectrum of activities on a variety of pathogens and substantial research has been carried out on the synthesis of new potent 2-azetidinone based antibiotics. ^{3,4} The β-lactam ring is a common structural characteristic of a number of broad spectrum β-lactam antibiotics, *i.e.* penicillins, cephalosporins, carbapenems, nocardicins, monobactams, clavulanic acid, subbactams and

tazobactams, which have been extensively used as chemotherapeutic agents for the treatment of bacterial infections and microbial diseases. ** Most research up to the early 90's was focused on the synthesis and study of 2-azetidinones and their antibacterial activities. β-Lactam antibiotics have played a fundamental role in the fight against pathogenic bacteria and the subsequent rise in quality of life for the world population as a whole.** In recent years, there has been a renewed interest in the synthesis and modification of β-lactam rings for their different pharmacological activities, like cholesterol absorption inhibitory activity, human tryptase, thrombin and chymase inhibitory activity, vasopressin VIa antagonist activity, antimalarial, antidiabetic, antitubercular, anti-inflammatory, antiparkinsonian and anti-HIV activity. ¹²⁻¹⁷

Department of Chemistry, I. K. Cujral Punjab Technical University, Kaprarinda, Punjab-144603, India. E-mail: gastav@ppu.ac.in: Fax: +91-1823-253504; Yei: +919478098058

Thepariment of Chemistry, Guru Nanak Dev University, Avaritsar. Parijab 143045, brilia



Mr. Yogesh Kumar obtained his B.Sc. degree from DAV College, Jalandhar, India in 2008. He joined Guru Nanak Dev University, Amritsar in 2010 to obtain his M.Sc. in Applied Chemistry with specialization in Pharmaceuticals. After one year as an Assistant Professor at DAV Institute of Engineering and Technology, Jalandhar, he started his Ph.D. in Organic Chemistry under the supervision of Dr

Gaurav Bhargava at I. K. G. Punjab Technical University, Kapurthala, His research interests include Organic Chemistry. He has published five research papers in international peer reviewed journals.

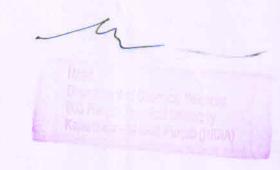


Dr Prabhpreet Singh received his Ph.D in 2006 from Guru Nanak Dev University, Amricsar, India. He spent one year at the Indian Institute of Technology, Kanpur and two years at the University of Strasbourg, France as post-doctoral fellow. He was also an Alexander von Humboldt (AvH) Fellow at Friedrich-Alexander-Universität Erlangen-Nürnberg, Germany. Currently, he is an Assistant Professor at Guru

Nanak Dev University, Amritsar, India. His research interests include Organic. Materials and Supramolecular (Themistry. He has published 48 research papers in international peer reviewed journals.

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COMMUNICATION

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A facile and highly chemoselective synthesis of 1thia-3a,6-diaza-benzo[e]azulen-3-ones by 7-exodig/trig halocyclizations†

Received 21st September 2016 Accepted 15th October 2016 Bilash Kuila, ^a Dinesh Mahajan, *b Prabhpreet Singh^c and Gaurav Bhargava*a

DOI: 10.1039/c6ra23493c

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This manuscript describes a study on relatively unexplored halogen mediated 7-exo-dig/trig cyclization reactions of 2-(2-amino-aryl)-3-prop-2-ynyl/allyl-thiazolidin-4-ones for the formation of thiazole condensed 1,4-benzodiazepines in good yields. The reactions are facile, chemoselective and involve the use of simple substrates leading to synthesis of diversely functionalized 1,4-benzodiazepines. The synthesis of such condensed 1,4-benzodiazepines is important in terms of their usefulness as biological active agents.

Halogen mediated cyclizations are gaining increasing importance for the construction of cyclic compounds having variable complexity including medium and large ring sizes." Iodocyclizations have been efficiently explored for the synthesis of biologically active heterocyclic molecules.2 Significant progress has been made in recent years by application of 5-exo and 6-exo-dig/ trig halocyclizations reactions for the synthesis of five or six membered heterocycles.1-a There are very limited examples in literature involving construction of seven-membered ring systems by 7-endo dig/trig cyclication 4.5 reactions. Recently, Miesch et al. explored a silver-catalyzed 7-exo-dig cyclization.50 Hashmi et al. reported gold catalysed 7-exo-dig cyclizations.3e However, the synthesis of seven membered carbo/heterocycles involving 7-exo digitrig halocyclization received much less attention in past.8 Moreover, the bromoamination reaction for the synthesis of seven membered heterocycles is searcely reported in literature. On the other hand, 1,4-benzodiazepine based structure class holds an important place in drug

discovery because of their diverse pharmacological profiles.⁸
1,4-Benzodiazepines are widely used as analgesic,⁸ antidepressive,⁸ anticonvulsant,⁸ anti-anxiety,⁶ sedative and
hyprotic agents.^{7,8} Olanzapine⁹ (I) is an atypical 1,4-benzodiazepines based anti-psychotic drug with a thieno-benzodiazepinyl
structure. Olanzapine is approved for the treatment of schizophrenia. It is a selective monoaminergic antagonist with high
affinity binding to serotonin 5HT2A/2G, dopamine D14,
muscarinic M1-5 and adrenergic at receptors. Alprazolam (II)
and estazolam (III) have found use as anxiolytic agents.^{100-c} and
artiazolam (IV) is known as an antidepressant ¹⁰⁰ (Fig. 1). In view
of this diverse pharmacological profile of condensed 1,4benzodiazepines, it was planned to explore a new synthetic
route for the formation of 1,4-benzodiazepines having thiazolebenzodiazepinyl structure.

As a part of our ongoing interest in heterocyclic chemistry, ¹² we already reported a β-lactam mediated strategy for a facile, chemoselective, economical, as well as short synthetic route to 1,4-benzodiazepin-2-ones. ¹¹ The present manuscript indicates a facile yet high yielding and chemoselective synthesis of functionalized thiazole condensed benzo[1,4]diazepines by relatively unexplored 7-exo-digitrig halocyclization reactions of 2-(2-amino-aryl)-3-prop-2-ynyl/allyl-thiazolidin-4-ones using halogen and a mild base. The developed synthetic methodology can potentially play an important role in synthesis of structurally diverse fused benzo[1,4]diazepines having diverse

Department of Chemical Sciences, I. K. Gujral Punjab Technical University,
Kapurthala, Punjab-144603, India. E-mail: gattrav@ptu.ac.in; Fax: 191-1822255504. Tel: 191-1822-255504

Drug Discovery Research Centre (DDRC), Translational Health Sciences and Technology Institute (THSTI), Faridabad-121001, India, E-mail: dinesh-mahajan@ thstires in

Department of Chemistry, Guru Nanok Dev University, Amritsur, Punjab 143005, India

† Electronic supplementary information (ESI) available. See DOI: 10.1039/c6m23493c

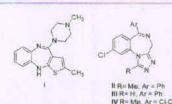


Fig. 1 Some clinically approved condensed 1.4-benzodiazepine based drugs.

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Journal of Materials Chemistry C



PAPER



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Diphenylpyrimidinone-salicylideneamine - new ESIPT based AlEgens with applications in latent fingerprinting†

Prabhpreet Singh,** Harminder Singh,* Rashmi Sharma,* Gaurav Bhargavac and Subodh Kumar**

Diphenylpyrimidinone—salicylideneamine DPPS-1 undergoes concentration dependent self-assembly to form aggregates with rods – mimicking a 'coral reef' superstructure and a spherical morphology evident from dynamic light scattering (DLS), scanning electron microscopy (SEM) and transmission electron microscopy (TEM) DPPS-1 and DPPS-2 exhibit weak excited state intramolecular proton transfer (ESIPT) based emission in CH₃CN and their binary mixtures with water, but in solutions containing >70% water fraction DPPS-1 and DPPS-2 molecules aggregate and ESIPT is facilitated to give a strong green ESIPT emission at 526 nm (ϕ = 0.273, 0.068). The ESIPT process remains active in the solid state and solid DPPS-1 and DPPS-2 appear as fluorescent green under 365 nm light illumination. The AlE-active nature of DPPS-1 and DPPS-2 finds applications in the visualization of latent fingerprints on aluminium, steel and glass surfaces. Even 24 h after the placement of the fingerprints, up to the second level of information vizinger, lake, ridge termination and dots, delta and bifurcation in the fingerprints could be identified.

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1. Introduction

Organic fluorophores possess a high propensity for aggregation in aqueous solution and in the solid state, which could lead to either 'aggregation-caused quenching' (ACQ)^{1,2} or 'aggregation induced emission' (AEE)³⁻⁵ phenomena. For practical applications in optoelectronics, biomedical research, ecosystem monitoring and environmental protection, the majority of organic fluorophores are used as thin films or in the aggregate form, and probably at this juncture, the AIE fluorophores provide an edge over the ACQ fluorophores.¹⁻⁵ Fluorophores with AIE properties remain non-fluorescent in the molecularly dissolved state owing to the dissipation of the absorbed energy through non-tadiative decay channels, whereas in the aggregated form close packing imposes restrictions to non-radiative decay channels and strong fluorescence is observed.⁶⁻¹⁷ In light of the bright prospects of AIE fluorophores in various applications, *researchers

have embarked upon the development of new AIE fluorophores to be included in the AIE family.

The excited state intramolecular proton transfer (ESIPT) process is an intriguing phenomenon4,18 which provides a large Stoke's shift via alteration in the conjugation through proton transfer, ESIPT based fluorophores can show two emission bands termed as normal emission from the enol-form (high energy and a low stokes shift) and ESIPT emission due to the keto-form (low energy and a large stokes shift). However, if the ESIPT process is allowed to happen, proton transfer usually takes place from the hydroxyl group (proton donor) to nitrogen (proton acceptor) through a cyclic transition state resulting in an energy gap much lower than that of a non-ESIPT state and consequently red-shifted emission is observed. Therefore, the fluorescence of ESIPT fluorophores can be quenched in the molecularly dissolved state (via blocking of the hydroxyl group) or can become AIE-active in the aggregated state either due to restriction in the motion process or through isolation from the bulk solvent system due to aggregation (unblocking of the hydroxyl group). In recent years, the combination of AIE and ESIPT has provided new dimensions and functions to bioimaging, self-assembly and disassembly based on the detection of analytes 19-34 in water.

The development of LFPs with a high efficiency and selectivity, and with low background interference is one of the major challenges in the identification of an individual for security and forensic reasons. Fingerprints contain a unique pattern of

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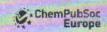
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[&]quot;Department of Chemistry, UCC Centre for Advanced Studies-II, Guru Nanak Der University, Amrician 143 005, Italia. K-masi: prabhpreet.chem@gnds.,ac.in, subodh.chem@gnds.ac.in; Tel: +91-86271-91534, +91-96016-31529

⁸ Department of RK., IK Cajrol Punjab Technical University, Kapurthala-144601, Punjab, India

Depunment of Chemical Sciences, IK Gujral Punjab Technical University, Kapurthalo 141601, Panjab, India

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■ Organic & Supramolecular Chemistry

Nanomolar Cu²⁺ Detection in Water Based on Disassembly of AlEgen: Applications in Blood Serum, Cell Imaging and **Complex Logic Circuits**

Prabhpreet Singh,*[a] Harminder Singh, [b] Vanita Vanita, [b] Rashmi Sharma, [c] Gaurav Bhargava, [d] and Subodh Kumar^[a]

Here in, we report self-assembled aggregates of AlEgen 1, which exhibits excited state inframolecular proton transfer. (ESIPT) coupled aggregation induced emission (AIE) with intense green fluorescence (\(\lambda_{en} = 526 \text{ nm}\) in aqueous solution as well as in solid state. AlEgen 1 behaves as disassembly based fluorescent probe for selective and sensitive detection of Cu1+ in drinking water with lowest limit of detection of 0.261 ppb (4.13 nM) (much lower than recommended by EPA). AlEgen 1 finds applications for the detection of Cu2+ in blood serum, environmental (rap, pond and industrial waste water) samples. Probably the interaction of Cu2 tons at Schiff-base unit of AlEgen 1 via metal chelation disrupts the self-assembled aggregates and results in disassembly of the nanorods into smaller spherical nanostructures as also evidenced in DLS, SEM and TEM studies. Incubation of HeLa cells with Alegen 1 results in fluorescence 'switch-on' showing cell internalization of AlEgen 1 with low cytotoxicity and thus establishes the potential of AlEgen 1 in cell imaging. The reversibility influorescence / absorbance of 1-Cu2- complex with EDTA helps in elaboration of various reconfigurable and superposed logic gates such as NOT (at Fazs, Assel, INH (at Assel, IMP, (at Assel and IMP, (at Fast By observing the output switching in absorbance and fluorescence channels on addition of Cu2+ (In1) and EDTA (In2) as inputs, 'set-reset' flip flop memory element and complementary IMP/INH logic circuits have been.

Introduction

Aggregation induced emission (AIE) is a phenomenon which is associated with abrupt increase in fluorescence intensity on aggregation of organic molecules in water or in solid state and has been exceedingly explored in last few years.[1-6] Organic molecules which are non-emissive in the molecularly dissolved state but strongly emit light in the aggregate form (achieved by addition of poor solvent or analyte complexation), are known as AlEgens. Up to now, AlEgens' family includes siloles,^[2-10] tetraphenylethenes,^[31-12] salicylaldehyde azines,³¹³⁻¹⁶ pyrrole and diketopyrrolopyrroles,[15-16] cyanostilbenes,[17-16] pyrazine, Duidiphenyldibenzofulvenes, 1151 and other derivatives. [21-25] The AIE fluorophores are beneficial in the fields of organic light emitting diode (OLED), cell staining and drug delivery by virtue of high emission in the solid state and in aqueous environment.[1] For practical applications in the domain of molecular recognition, AIE fluorophores provide the liberty of working in solvent systems with water content as high as > 95% and allow analyte detection even in the solid state.[3]

Copper, an essential nutrient, is third most abundant transition metal in the body. The deficiency of copper in diet associated with neurodegenerative diseases viz. Alzheimer's, Minke and Wilson's disease. On the other hand, the high concentrations of copper prove to be toxic to the body and have high environmental hazards.[26-31] Therefore, in order to early diagnosis and treatment of these aforesaid diseases, the development of copper sensors with high sensitivities and low detection limits with ability to work in aqueous conditions is crucial. 196 The detection of copper in water using AIE based chemosensors has been scarcely reported with only limited number of reports in literature[92-35] with lowest detection limit of 160 nM.

On the other hand molecular computing or molecular logic gates could provide alternate of silicon-based computer. Molecular logic gates provide possibility of reconfigurability (wavelength-based) and superposition (functioning of different logic simultaneously) of logic gates. After pioneering work in molecular logic gates by Aviram and then de Silva, researchers are in progress of developing more organic molecule for addressing complex molecular logic gates.[41-43]

[a] Dr. P. Singh, H. Singh, Prof. Dr. S. Kumar Department of Chemistry UGC Centre for advanced studies -II Guru Nanak Dev University Amritsar 143005, India mail: prabhpreet.chem@gndu.acin

1b1 Prof. Dr. V. Vanita Department of Human Genetics Guru Nanak Dev University Amritsar 143005, India

[c] R. Sharma Department of RIC IK Gujral Punjab Technical University Kapurthala 144601, India

[d] Dr. G. Bhargava Department of ChemicalSciences IK Gujral Punjalo Technical University Kapurthala 144601, India

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'To kill many birds with one stone': Addressing half-adder, half-subtractor, demultiplexer, 2-to-4 decoder, comparator, keypad lock with unimolecular system

Prabhpreet Singh a. . Harminder Singh a. Gaurav Bhargava b. Subodh Kumar a

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Keywords Dual cheese Half-Adder/Subtracto Keypad lock

ABSTRACT

Unimolecular system SBU shows multiple output responses on interaction with Cu^{2n}/F ions individually and simultaneously. The complexation of Cu^{2n} with timine (-C = N) and -OH groups results in blue-shift from 332 to 303 nm and de-protonation of —OH group with F ious causes red-shift of the absorption maxima from 332 to 405 nm. The addition of F to SBU∩Cu²⁺ solution or addition of Cu²⁺ to SBU∩F solution shifts the absorption maxima from 303 to 374 nm or 405–303 nm, respectively. Similarly, the complexation of Cu²⁺ tons inhabits the ESIPT emission at 517 nm with concommutant release of emission at 425 nm, whereas F results in quenching of emission at 512 nm up to 10 µM. followed by blue-shift to 425 nm at >10 µM. F ions. The addation of F to SBUINCU² solution causes quenching of emission at 512 and 425 nm, whereas addition of Cu²⁺ to complex SBUINF solution revives the emission at 425 nm considerably. These multiple output response leads to elaboration of YES, NOT, NOR, INH, AND, XOR and XNOR logic gates and Half-adder, Half-subtractor, 2-to-4 decoder, 1-2 densitisplexer, Magnitude Comparator, complimentary OR/NOR circuit and molecular keypad lock logic circuits

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1. Introduction

Information technology (IT) has helped the mankind in developing fast computation techniques that could be used in communication, aeronautical-space science, robotics, medical diagnostics, data storage and even programmed activation of prodrug [1,2]. At present, IT systems have been sculptured from electronic logic gates developed by means of silicon-based-semiconductor materials [3]. However, these are plagued with lack of reconfigurability, size >100 nm, excessive heat generation and load of data [4]. The molecular computing provides promising alternate to silieon computers due to ease in chemical modification in the molecular structure, high level of information storage and bottom-up approaches to control the dimensions and propernes of the molecules [5-10]. The molecular computing or molecular logic gates [7,8] can be elaborated at <1 nm scale.

In parallel to huge array of binary logic gates, which are operational components in computer, Aviran [11] in 1988 first proposed molecular logic and later on de Silva reported molecule based AND logic gate [12]. This was followed by plethors of reports describing molecular logic systems by modulating the input signals such as laser at different wavelengths, chemical stimuli or proton concentration and receiving the outputs in terms of changes in absorption, emission and redox properties [13-16]. Molecular logic gates can be reconfigured [17], by using different output signals at different wavelengths (wavelength-reconfigurable logic gate) and also provide upportunity to look into functioning of different logics simultaneously with a given set of inputs (super-position of logic) [18]. Also, with molecular logic systems, multiple optical signals can be instantaneously transmitted by virtue of an optical fibre cable with high travelling speeds [19]. In literature the execution of complex molecular analogues such as half-adder, half-subtractor, full-adder, full-subtractor, encoder, decoder, multiplexer, demultiplexer, molecular key pad lock and other such complex systems are quite tedious [20-29] and involve simultaneous use of more than one molecule in the solution to read different logic gates or need great attention to choose between positive and negative logic convention [30]. So, there is a need to develop unimolecular systems to address complex molecular logic gates

Dual chemosensor bearing pockets for cation and anion separately can perform supramolecular recognition via colorimetrically or fluorescent-based response through modulation of different signaling mechanisms [31-34]. Such dual chemosensors could be foreseen as potential candidates to develop unimolecular system to address multiple logic gates. In our previous works, we have developed tetrapodal and dipodal systems based molecular switches and logic gates [35,36] Recently, we designed perylenedimide and diphenylpyrin-idinone based chemosensor which shows self-assembled nanorod [37,38] and nanovesicle [39] morphology and utilized its Cu2*/Pd0 based response for mimicking several logic gates [37]. We also reported dipod based dual chemosensor which offered possibility for INH, IMP, AND logic gate via triple signalling mechanism [31].

Email address: prabhpreet chem@gudu ac.in (P. Singh)

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Department of Chemistry, UGC Centre of Advance Studies II, Garu Nanak Dev University, Amrissar 143 005, India Department of Chemical Sciences, IK, Giginal Punjab Technical University, Kaguribala, 144 601, Punjab, India

Corresponding author

Biosensors

Ionic Self-Assembled Platform of Perylenediimide-Sodium Dodecylsulfate for Detection of Spermine in Clinical Samples

Prabhpreet Singh, *[a] Lalit Singh Mittal, [a] Gaurav Bhargava, [b] and Subodh Kumar*[a]

Abstract: The detection and quantification of spermine in clinical practice is important for early diagnosis of many diseases. Chromatographic and immunoassay-based methods are commonly used. However, a fluorescence-based assay could provide real-time detection. Herein, the synthesis and aggregation properties of a dicationic perylene probe (NIdodecyl-N3-(4-phenyl)benzimidazolium-functionalized perylepediimide (DAR-POII) used to develop a fluorescent "hurnon" ensemble for the detection of spermine are discussed. The fluorescence of DAB-PDI (10 μM , Φ = 0.55) is efficiently quenched by negatively charged sodium dodecylsulfate (SDS) through the formation of ionic self-assembled aggregates (charge ratio of negative (N) in SDS to positive (P) in DAB-PDI (N/P) = 9). This negatively charged ionic self-assembly between DAB-PDI and SDS has been characterized by using photophysical, microscopic, dynamic light scattering, isothermal titration calorimetry, and HRMS techniques. The addition of spermine to this ensemble solution results in the breakdown of the DAB-PDI-SDS ensemble owing to strong binding of spermine with SDS and, as a result, the fluorescence of DAB-PDI is recovered. This ensemble exhibits high sensitivity and selectivity for spermine detection in water, urine, and blood serum. The lowest limit of detection is 27.5 nm, which is at least about 36 times lower than that required for early diagnosis of cancer (1 to 10 µm for urinary

Introduction

Spermine, a biogenic aliphatic tetraamine, is found in eukaryotic cells and body fluids. Owing to its polycationic nature, spermine plays an important role in cell growth and proliferation, and is involved in various stages of protein synthesis and stabilization of nucleic acid conformations. [11-3] Elevated spermine concentrations in urine and blood could be related to the presence of malignant tumors or cancer. Thus, the detection of spermine in physiological samples can be regarded as an excellent tool for early diagnosis and to assess the effectiveness of cancer therapy. A Traditional methods, such as chromatographic techniques, immunoassays, and mass spectrometry, are time consuming, require tedious procedures, and expensive instrumentation. (6) On the contrary, fluorescence-based

methods are more selective, approach ultratrace-level sensitivity, and involve simple operation with real-time detection. How ever, to date, few reports on fluorescence-based assays with "turn-on" responses in the nanomolar range for spermine are available in the literature.[7] Therefore, there is a need to develop methods with satisfactory sensitivity and selectivity for the detection of spermine.

Sodium dodecylsulfate (SDS), owing to its amphiphilic nature, is known to form supramolecular aggregates, such as micelles and vesicles in aqueous solution. 181 Owing to extensive industrial and domestic use of SDS, its presence in natural waters has caused serious environmental problems by inhibiting biological activity.¹⁹ It also helps in the transfer of other pollutants, namely, petroleum products, oil, and pesticides, into water because of its foam formation property. Therefore, apart from creating a supramolecular self-assembly, we also need effective fluorescence-based methods that enable the analysis of SDS from water because it can be used as a masking agent in drug abuse and food preparations.

Perylene diimides (PDIs), which form an interesting class of functional organic dyes, have been used as high-performance pigments. (19,11) supramolecular hosts, (12,17) and in optoelectronic devices.[18,19] The functionalization of PDIs at bay-, peri-, ortho-, and N-terminal positions is being performed to increase the solubility of PDIs in organic solvents. However, the poor solubility of PDI derivatives in aqueous media remains one of the most detrimental factors in the development of PDI-based receptors for the recognition of cations and anions in water. In recent years, the groups of Hirsch and Würthner have used the

(a) Dr. P. Singh, L. S. Mittal, Dr. S. Kumar Department of Chemistry, UGC Centre for Advanced Studies Guru Nanak Dev University Amritsar 143005 (Inclia) E-mail: probhprees.chem@gndu.ac.in subodh.chem@gndu.ac.in

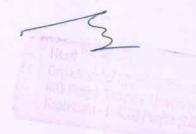
(b) Dr. G. Shargava Department of Chemical Sciences Punjab Technical University Kanuethala-144601 (india)

Supporting information for this article can be found under: http://dx.doi.org/10.1002/asia.201700120. It contains experimental procedures and characterization data for PDI 3 and DAB-PDI; additional micro scopic and spectroscopic data for the DAB-PDI-SDS ensemble; and spec-troscopic and DLS data of thration for spermine with the ensemble.

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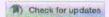


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PAPER

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Cite this: New J. Chem., 2017. 41, 10281 Perylene diimide-based organic π -motif for differentiating CN $^-$ and F $^-$ ions by electron-transfer and desilylation mechanisms: applications to complex logic circuits†

Prabhpreet Singh, ** Latit Singh Mittal, * Harminder Singh, * Gaurav Bhargava* and Suborth Kumar **

In this work, we have designed and synthesized a perylene dilimide based derivative PDI-SiHQ having an O-sitylated-8-hydroxyquinoline group at the bay position. PDI-SiHQ is able to detect and differentiate CNT from FT in the presence of other anions. The electron transfer from cyanide ions to PDI-SiHQ leads to the emergence of panchromatic and NIR absorption bands due to the formation of a PDI-SiHQ* radical anion. In contrast, fluoride induced desitylation of PDI-SiHQ coupled with $\pi-\pi$ aggregation results in a red-shifted absorption band at 730 nm. ¹H NMR and cyclic voltammetry combined with the reversal of the PDI-SiHQ* radical anion to the PDI-SiHQ neutral form in the presence of NOBF₈ or Cu²⁺ ions confirm the formation of the radical anion in PDI-SiHQ. Multi-channel UV-Vis-NIR absorption and emission bands and their reversibility with NOBF₈/Cu²⁺ ions offer the possibility for fabrication of complex logic circuits such as complementary IMP/INIH logic circuits, magnitude comparators and 'set-reset' flip flop memory elements

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1. Introduction

Penylene diimides (PDIs)^{1,2} are a valued class of n-type dye which are highly colored (with broad color range), fluorescent (quantum yield is near unity) and possess a high absorption coefficient and good redox, electron acceptor and light harvesting properties. Moreover, PDIs offer wide freedom of functionalization (at bay-, imide-, peri- and ortho-positions)^{3–5} of their n-backbones for modulation of optical and electronic properties for possible applications in photonic materials⁶ (as chemosensors and molecular logic gates) and optoelectronic materials (as field effect transistors, organic light emitting diodes and dye sensitizers in solar cells).^{7–9} In our endeavor towards exploiting self-assembled materials^{10–13} for applications in biosensing and bioimaging, we have reported a library of PDI based derivatives for molecular recognition of metal ions, anions and biogenic amines in aqueous/organic-aqueous solvents.^{10,11} Within this framework, we recently reported a PDI derivative for the

detection of ClO₄ wia encapsulation in the cavity¹¹ and sodium dodecyl sulphate (SDS) anion via ionic self-assembly¹⁰ in water and live cells. In these cases, the binding between the anion and PDI based receptor relies on non-covalent forces such as hydrogen bonding and hydrophobic and electrostatic interactions.

In recent years, the concept of anion- π interactions, ⁴⁴⁻¹⁶ so called electronic interactions between electron-rich anions and electron-deficient π systems, began to emerge as a novel method to generate radical anions which offer applications in supramolecular and material chemistry combined with multichannel and panchromatic-NIR detection of anions. 17-20 From a series of recent literature, it is clear that the position of charge-transfer bands varied with the nature of electron-donors (viz. nature of the anion and use of reducing agents such as dithionite and ascorbate ions)21-23 and electron-acceptors (arvlenediimide, HAT(CN)6, binary complex of arylenediimide and macrocyclic supramolecular host viz. cucurbit 4 Juril). 17-19.24 Mechanistically, when an acceptor molecule (such as PDI, NDI and HAT(CN)6) having a low lying LUMO level comes in contact with the HOMO of CN , F or OH (strong basic anions), a facile single electron-transfer (SET) mechanism involving HOMO-LUMO rapidly forms a stable radical anion via single electron transfer. Usually, all strong Lewis basic anions (CN , F and OH) trigger thermal electron transfer (ET) processes and thus it is difficult to differentiate between these basic anions. In this context, many efforts are undertaken to develop arylenediimide based derivatives

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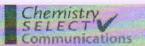
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^{*}Department of Chemistry, 1700 Contre for Advanced Studies, Gura Nanak Dev University, Amritsar 143 005, India. E-mail: prabhprees.chemijygidu.ac.in; Tel: +91-84271-01584

Department of Chemical Sciences, IKG Punjah Technical University,

Kapurthala-144601, Punjab, India

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Catalysis

Copper(I)-Catalyzed Regioselective C-H Amination of N-Pyridyl Imines Using Azidotrimethylsilane and TBHP: A One-Pot, Domino Approach to Substituted Imidazo[4,5-b] pyridines

Nisha, *la] Gaurav Bhargava, [b] and Yogesh Kumar*[b, c]

Dedicated to Prof. M. P. Mahajan on the occasion of his 70th birthday

A novel efficient and one-not multicomponent conner-catalyzed amination of N-pyridyl imines is described. This, one-pot. multicomponent reaction, in which imine acts as a directing group by chelating to the metal core, affords a domino approach for the transformation of the commercial pyridinamines, aldehydes and azide into valuable imidazof4.5-b) pyridines with wide substrate extent and diversity.

Advances in transition-metal catalysis have led to the progress of valuable methods for regioselective carbon-carbon, carbonnitrogen, carbon-oxygen, and carbon-sulfur bond formations. [12] The carbon-nitrogen bond formation has attracted significant interest since nitrogen-containing heterocycles have extensive applications in the biological and medicinal chemistry.^[1,3] Among first-row transition metals (Mn, Fe, Co and Cu), copper is well considered as a catalyst with various amine ligands, additives, and under various conditions for coupling reactions. But, there are very few reports on the coppercatalyzed systems, which are particularly striking because of its high abundance and low toxicity.⁽³⁾ Copper-catalyzed C-H amination is one of the most fundamental and significant reactions as it provide a straightforward and inexpensive protocol to synthesize a wide range of nitrogen heterocycles.

On the other hand, imidazo[4,5-b]pyridines scafold represent a nucleus skeleton of various heterocyclic analogues of purine with biological activities which depend on the substituents on the pyridine ring. These medicinal important Imidazo[4, 5-b]pyridines⁵¹ has been evaluated as an antagonist of different biological receptors, including platelet-activating factor (PAF),^[7] angiotensin-li^{l6} and metabotropic glutamate subtype V. Recently, these are also reported as anticancer, 9 inotropic, not and selective antihistamine (H1) agents[11] aurora kinases^[12] and cyclic phosphodiesterase inhibitors.^[13] But only a few approaches were reported for the regioselective synthesis of imidazol4,5-blpyridines via metal-catalyzed cascade reactions and C-H functionalization/activation reactions. An attempt has been recently made on the expansion of new strategies to build the imidazo(4,5-b)pyridine structural motifs using Pd-catalyze'd amide coupling reaction of chloro-aminopyridines (Scheme 1, a), acid catalyzed cyclization of diamino-

Scheme 1. Main strategies for imidazo[4,5-b]pyridine syntheses

[a] Dr. Nisha Department of Chemistry Faculty of Physical Sciences SGT University Gurugram, Haryana-122505, India E-mail: nlsha8710@gmall.com

[b] Dr. G. Bhargava, Dr. Y. Kumar Department of Chemical Sciences IKG Punjab Technical University Kapurthala, Punjab-144603, India E-malt yogeshsynthesis@gmail.com

Sphaerapharma Pvt. Ltd. Manesar, Fiaryana, India-122051, India

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Invited paper

Fluorometric differential detection of Zn²⁺ and Cu²⁺ by picolylamine appended pyrimidinone-based receptor: Application in mimicking TRANSFER and INH logic gate



Prabhpreet Singha, Harminder Singha, Rashmi Sharmab, Sanjeev Dhawane, Parvesh Singh^c, Gaurav Bhargava^d, Subodh Kumar^a

- Department of Chemistry, UGC-Centre of Advance Studies-II, Guru Nanak Dev University, Amritsar, 143 005, India
- ^b Department of Chemistry, Trinity College, Jalandhar, 144009, Punjab India School of Chemistry and Physics, University of KwaZulu Natal, PiBag X54001, Westville, Durban, 4000, South Africa ^d Department of Chemical Sciences, IK Gujral Punjab Technical University, Kapurkhalo, 144 601, Punjab, India

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Keywonts Pyrimidinone Dual chemose Zn²'/Cu²* Logic gate Picolylamine

ABSTRACT

Herein, we have designed and synthesized pyrimidinone-based dual chemosensor (DPPA) containing picolylamine moiety as binding unit. Investigation on the optical property of DPPA revealed fluorescence enhancement at 435 nm with Zn² ion through binding with picolylamine and carbonyl moieties, while Cu² ion was detected through fluorescence quenching at 425 nm and colorimetric changes at 350 nm due to complexation with picolylamine and deprotonation of amide — NH group. The detection limit of Zn^{2+} and Cu^{2+} was calculated to be 10.04 nM and 34.1 nM, respectively. The Job's plot showed formation of 1:2 and 2:1 binding stoichiometry of DPPA with Cu² and Zn², respectively. ¹H NMR titration in CD₃CN revealed downfield shift in methyl groups of N(CH₃)₂ and methylene (—CH₂) protons of picolylamine chain due to formation of coordinate bonds between Cu²⁺ and DPPA, whereas only downfield shift of methylene (—CH₂) protons of picolylamine were responsible for complexation with Zn²⁺ ions. The complexation mechanisms of DPPA with Cu²⁺ and Zn²⁺ were also supported by detailed theoretical calculations.

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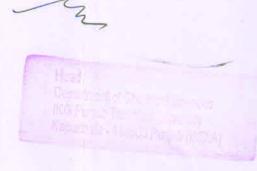
1. Introduction

Dual chemosensors [1-3] are capable of detecting two analytes simultaneously and thus provide an edge over the chemosensors (which can detect only one analyte) in terms of cost-effectiveness and efficient analysis in fields and contaminated sites [4]. Among the different detection approaches reported so far for analyte using dual chemosensors includes either simultaneous detection of two analytes at the same time independent of each other (simultaneous approach) [5]; or detection of second analyte achieved through breaking of complex between chemosensor and first analyte in a sequential manner (displacement approach) [6]. The sequential detection of two metal ions or metal ion and anion can be easily achieved using metal (Zn²*, Cu²*, Hg²*) bound receptor followed by displacement of either metal ion with another metal ion due to higher binding constant or displacement of metal ion with anion via strong binding of these metals with target anions. However, design strategy of dual chemosensor for simultaneous detection of two metal ions or metal ion and anion required incorporation of differential binding domain for both analytes in unimolecular system [7,8]. The binding of these analytes at their respective binding domain would produce differential color or fluorescence response due to change in signalling mechanism such as CHEF (chelation induced fluorescence enhancement), ESIPT (excited state intramolecular proton transfer), CHEQ (chelation induced fluorescence quenching), PET (photoinduced electron transfer), ICT (internal charge transfer), PCT (photoinduced charge transfer) and FRET (fluorescence resonance energy transfer) [9]. Moreover, in last few years we have witnessed that organic molecules, having dimension in nanometer scale and variability in properties through chemical functionalization, are extensively employed for mimicking logic gates and circuits. In this context, dual chemosensor showing multiple outputs in absorbance or fluorescence mode could be used to fabricate logic gates and circuits [10-17].

Zn2* is the second most abundant transition metal ion and plays an important role in many biological processes such as apoptosis,

* Corresponding author E-mail address, prabhpreet, chemiligad u.ac.ia (P. Singh).

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Research Paper

Self-assembled small molecule based fluorescent detection of serum albumin proteins: Clinical detection and cell imaging



Prabhpreet Singh a., Lalit Singh Mittal a, Sandeep Kaurb, Satwinderjit Kaurb, Gaurav Bhargava^c, Subodh Kumar^{a,*}

- Department of Chemistry, Gura Nanak Dev University, Amritsar 143 005, India
 Department of Botanical and Environmental Sciences, Gura Nanak Dev University, Amritsar, India
 Department of Chemical Sciences, IK Gigfal Punjab Fechnical University, Kapur thala 144601, India

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Biosensors Fluorescence Proteins Perylenediimides

ABSTRACT

We report perviewed/imide-benzimid-zolium based fluorescent 'turn-on' probe BIM-PDI for selective detection of human serum albumin (HSA) and bovine serum albumin (BSA) proteins. In HEPES buffer (0.1% DMSO), BIM-PDI self-assembles into aggregates and shows absorption maxima at 500 nm and weak fluorescence centered at 577 nm. The addition of HSA or BSA ($1\times10^{-9}-5\times10^{-8}$ M) to the solution of BIM-PDI results in decrease in the emission intensity at 577 nm. However, further increase in concentration of HSA/BSA results in appearance of new blue shifted emission band at 540 nm. The minimum detection limit for HSA/BSA is $3.01\times10^{-10}\,\mathrm{M}$ at 577 nm and $4.2\times10^{-8}\,\mathrm{M}$ at 540 nm. On addition of BSA to the solution of BIM-PDI, the size of the aggregates decreased from 100 to 250 nm to <10 nm assigned to microencapsulation driven disassembly of BIM-PDI aggregates responsible for 'turn-on' response in fluorescence spectrum. Site-selective experiments using warfarin and diazepam drugs show that BIM-PDI preferably binds at site-I of HSA/BSA proteins. In clinical application of BIM-PDI, we estimated HSA content in blood serum and urine samples. BIM-PDI showed potential application in serum albumin protein tracking and imaging.

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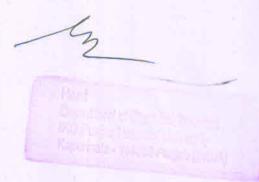
1. Introduction

Perylenediimide (PDI) is an important scaffold and well documented class of functional organic dyes for developing supramolecular nanostructures for applications in molecular recognition [1-8]. The perylenedlimide core can be modified through functionalization at -boy, -peri, -ortho and N-terminal positions [1,9,10]. The functionalization of PDIs at N-terminal positions are being done to increase the solubility of the PDIs in organic solvents without affecting the optical properties of PDIs, whereas bay functionalization dramatically changes the photophysical properties of PDIs. The poor solubility of -bay and N-terminal functionalized PDIs in aqueous medium results in selfaggregation of PDIs through short-range non-covalent forces such as π-π stacking. H-bonding or van der Waal's forces to produce either H-aggregates [11] or J-aggregates [12] and thus greatly hampers the exploitation of PDIs in supramolecular chemistry.

However, if the interaction of a specific stimulant with these aggregates of PDIs in water occurs, it would result in change of fluorescence and morphology of the aggregates to mark the molecular recognition process in real time. In this context, recently we have reported library of 1- or 1,7- substituted perylenediimide derivatives to achieve self-assembled morphology of nano-spheres, vesicles or rods and utilized these supramolecular aggregates to develop fluorescent probes for ClO₄ . Pd⁰/Pd²⁺, sodium dodecyl-sulfate (SDS), spermine, trivalent metal ions in water and live cells; detection of Cu2+ and CN- in organic media [3-5].

Concept of using self-assembled aggregates of small molecules as fluorescent bio-probes for the detection and imaging of protein has attracted much attention in recent years [13-25]. Amongst different thiol-containing proteins, the serum albumins (SA) - human serum albumin (HSA) in human and bovine serum albumin (BSA) in animals, play important roles in maintaining oncotic pressure of the blood, transporting and deposition of various endogenous and exogenous molecules to specific sites, scavenging free radicals and in preventing protein aggregation [26-28]. HSA and BSA are structurally similar and have two major sites for non-covalent binding of figands namely site-I, where binding is primarily driven by hydrophobic interactions, and site-II, where both hydrophobic

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^{*} Corresponding authors.
E-mail addresses: prabhpreet.chem@gndu.ac.in (P. Singh), aubodh.chem@gndu.ac.in (S. Kumar).

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(rods-wires-spheres) and ICT/PET based perylene probes for palladium detection in solution and the solid state†

Kapil Kumar, Gaurav Bhargava, Subodh Kumar and Prabhpreet Singh **

Controllable supramolecular self-assemblies

Control over the self-assembly of bay functionalized perylene dilmides and the response of these self-assembled materials towards analytes in mixed aqueous—organic media is important to consider the potential of this intriguing dije family in supramolecular chemistry. Herein, we have designed and synthesized altylcarbonate-functionalized perylene climinde AC-PDI which shows solvent dependent supramolecular nanostructures and could readily self-assemble into nanorods of tens of μ m in length in CH₃CN, but the length of the rods drammatically decreased to 0.8–1.2 μ m in a H₂O: CH₃CN (1: 1, ν /v) mixture. Interestingly, AC-PDI self-assembles into nanowires of several tens of μ m in length in a 30% H₂O:DMSO mixture and forms only nanospheres in ethyl acetate AC-PDI can be used as a near-infrared (NIR) colorimetric U_{atip} 685 nm) and fluorescent U_{cm} 530/570 nm) probe for selective detection of Pd⁰ (LOD = 39 nM (UV); 45 nM (FI)) Furthernore, TLC strips coated with the AC-PDI probe could detect 0.58 pg cm⁻² of Pd⁰ in the solid state. The mechanism of interaction of AC-PDI with Pd⁰ has been confirmed by ¹H NMR titration, kinetic and morphological studies.

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Introduction

Controlled self-assemblies of π -conjugated molecules have drawn much attention in the last decade due to their potential applications in photovoltaic and optoelectronic devices.1-4 Among various π-conjugated molecules, perylene diimides (PDIs) are one of the most studied classes of functional dyes with excellent electronic, optical and redox properties.2,3 PDIs have been widely utilized as an active component of organic field-effect transistors (n-type), light emitting diodes, solar cells and sensors 4-10 because of their strong absorption combined with high fluorescence quantum yield, low reduction potential, good electron acceptor properties and broad colour range properties.2,3 The aggregation (self-assembly) and optoelectronic properties of PDIs primarily depend on the local arrangement of their x-stacked array, which can be tuned through modification of their π-molecular structure at the bay-, peri-, ortho- and N-imide positions with organic chemical groups.3-4 Many reports regarding modification of the

N-imide position of PDIs to generate self-assembled nanostructures including fibers, ribbons, rods, tubes *etc.* are being published each year. However, little attention has been paid to the self-assembly of bay substituted PDIs in an aqueous medium.¹¹ Probably, substitutions at a bay position drastically increase the solubility of PDIs and if the substituents are bulky in nature, the perylene backbone deviates from planarity to minimize the strain and this hinders the possibility of π - π stacking and so the examples of self-assembled bay functionalized PDIs are less known.

Palladium is one of the most important transition metals, extensively exploited in modern organic chemistry and industrial production. Moreover, the high affinity of Pd species towards thiol containing biomolecules such as amino acids and proteins may cause their denaturation and Pd can also bind effectively to DNA which may disrupt a number of cellular processes in the human body. ¹³⁻¹⁶ Henceforth, it is prerequisite to establish fast and effective techniques for the detection of palladium with high sensitivity and selectivity. Due to the heave cost factor and tedious operational procedures involved in conventional analytical methods, appreciable attention has been paid to the establishment of fluorescent probes for the detection of Pd species (Pd⁰ or Pd^{2*}), ^{5c/17}

In the last decade, successful efforts have been made to explore N-imide functionalized PDIs as photoinduced electron-transfer (PET) based chemosensors for metal ions in mixed

of AC-PDI and AC-PDI + Pdo. See DOI: 10.1039/e7nj03751a

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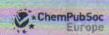
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Department of Chemistry, UGC Centre for Advanced Studies, Guru Nanak Dev University, Amritsar 143 005, Punjah, India. E-mail: prohipport.chem@gideLac.in/ Tel: +91-84271-01534

^{*}Department of Chemical Sciences, IKG Projab Technical University, Kapurthalu-144601, Projab. India

Raparthala:144601, Panjat. India.

† Electronic supplementary Information (ESI) available: Experimental cleraits, and additional photophysical, spectroscopic and microscopic characterization.



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■ Cycloadditions

Transition-Metal-Catalyzed [3+2+2] Cycloaddition Reactions

Bilash Kuila, [a] Maninderjeet Kaur, [a] Prabhpreet Singh, [b] and Gaurav Bhargava*[a]

Abstract: [3+2+2] Cycloaddition reactions are of key importance in organic chemistry for the synthesis of seven-membered mono- and condensed carbocycles. This review article summarises the different reports on transition-metal-catalysed [3+2+2] cycloaddition reactions since 2000. Different transitionmetals, such as Ni, Rh, Pd, Co, or Ru, which have been earlier reported for higher order [3+2+2] cycloaddition reactions are summarized in this article. The [3+2+2] cycloadditions of different functionalized substrates afford cycloheptane derivatives, 6-7,5-7 condensed bicyclic carbocycles, or 5-7-5 or 6-7-5 condensed tricyclic carbocycles. These higher order cycloaddition reactions may provide an easy route for the total synthesis of natural products, or as a preferable key synthetic step leading to final structure

1. Introduction

Cycloaddition reactions are the most powerful method for the construction of a complex structure in a single step.[1] Although two component (4+2) and (3+2) cycloaddition reactions are synthetically the most explored reactions for the synthesis of six- and five-membered carbo-/heterocycles, respectively, but the comparable methodology for higher-membered carbo-/ heterocycles are still in the process of development. The higherorder [m+n+o] cycloaddition reactions are emerging as useful synthetic tools for the synthesis of higher order mono or condensed carbo/heterocyclic ring systems.[2,3] The transitionmetal-catalyzed [m+n+o] reactions provide useful carbo-/ heterocycles from relatively inactive substrates, such as alkenes and alkynes.[4,5]

Of different variants of [m+n+o] cycloadditions reactions, [3+2+2] is an important synthetic tool for the synthesis of functionalized mono- or condensed cycloheptanes. The importance of atom-economical [3+2+2] cycloadditions lies in the fact that cycloheptanes and their analogues have played vital role as functionalized biomolecules in food, pharmaceutical, and natural products. (6-8) Methylenecyclopropanes as а 3л component has been utilized as an important synthon in organic synthesis.^[9–14] Methylene- and vinyl-cyclopropane as organic synthons have been successfully explored for [m+n] and [m+n+o] cycloaddition reactions with a variety of alkenes and alkynes using different metal complexes as catalysts.[15] For example, cyclopropyl ketones, alkylidenecyclopropanes (ACP) and vinylcyclopropanes (VCP) have been shown to participate in several transition-metal-catalyzed [3+2], [4+3], [3+2+2] and [5+2] cycloaddition reactions.[16-19]

Different researchers have successfully explored functionalized cyclopropanes as 3π components with various alkenes, alkvnes, divnes, among others, as 2π components in (3+2+2) cycloaddition reactions using NI, Pd, Rh and Ru metal complexes as catalysts [70-27] The present review summarizes various reports on [3+2+2] cycloaddition reactions with not only products arising from the activity of the key players (different metal complexes) but also those from proton-based chemistry for the synthesis of mono-and/or condensed cycloheptadiene derivatives, since 2000.

2. Nickel-Catalyzed [3+2+2] Cycloaddition Reactions

In this section, we have summarized the various nickel-catalysed inter-and intramolecular [3+2+2] cycloaddition reactions. A majority of the reports on [3+2+2] cycloaddtions involve the use of functionalized cyclopropanes as 3x-components in cycloadditions with a variety of alkenes and alkynes as 2π-components.

2.1 Nickel-Mediated [3+2+2] Cycloadditions of Alkenyl Fischer Carbene Complexes

One of the first reports on [3+2+2] cycloaddition reactions was described by Barluenga et al. it involves the nickel-mediated [3+2+2] cycloaddition reaction of chromium carbene complexes 1 with various terminal alkynes 3 (Scheme 1) to afford a variety of cylcoheptatrienes in fair yields. [28]

Nickel-mediated [3+2+2] cyclizations of Fischer-carbene complexes and alkynes take place with complete regio- and stereoselectivity, with the formation of an endo isomer. Electron-poor alkynes such as methyl propiolate also undergo [3+2+2] cyclizations to afford functionalized cycloheptatrienes In good yields. [28]

A pathway for the formation of 6 and 7 involves the initial chromium-nickel exchange to yield the nickel species 2 (where

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⁽a) Department of Chemical Sciences, IKG Punjob Technical University, Kapurthaln 144603. Punjab, India E-mail: gaurav@ptu.ac.in https://maincompus.ptu.ac.in/faculty/dr-gaurav-bhargava/

⁽b) Department of Chemistry, Guru Nanak Dev University, Amritsar, 143005, Punjab, India

Acetylenic Ester Promoted Tandem Ring Opening of Dienyl Thiazolidin-4-ones and Cyclizations: A Facile and Chemoselective Synthesis of Functionalized Pyridine-2-carboxylates

Bllash Kulla Kapil Kumar^b Dinesh Mahajan Prabhpreet Singh Gauray Bhargaya**

Department of Chemical Sciences, t. K. Gujral Punjab Technical University, Kapurthala, Punjab-144603, India Department of Chemistry, Caru Nanak Dev University, Amritsar,

Department of Chernstry, Card Manax Dev University, Amints. Punjab 143005, India Drug Discovery Research Centre (DDRC), Translational Health Sciences and Technology Institute (THSTI), Fandabad-121001, India

paurav@otu.ar.in

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Abstract Acetylenic ester promoted ring opening of dienvi-thiazoli ones and subsequent electrocyclization affords 5-phenyl-6-arvl pyridine-2-carboxylates in good to excellent yields.

Key words pyridine-2-carboxylate, dienyl-thiazolidin-4-one, acetylenester, cyclization, 5.6-diarylpyridine

Functionalized pyridines having ester substituents at their 2-position i.e., pyridine-2-carboxylates, are prominent in biologically active molecules.¹⁻³ Functionalized pyridine-2-carboxylates have been identified as cholecystokinin (CCK1) receptors, cannabinoid receptor type 1 (CB1), and telomerase inhibitors.4 Similarly, 5,6-diaryl-2-pyridinecarboxamides have been used as urotensin II receptor antagonists and sphingosine-1-phosphate (51P) receptor agonists.5

Traditionally, [4+2] cycloadditions of acyclic 2-azadienes have been used as a versatile method for the formation of functionalized pyridines, dihydropyridines, and tetrahydropyridines.7-11 Barluenga et al. have explored the synthesis of functionalized pyridines utilizing ethoxycarbonyl 2-aza-1,3-butadienes as starting materials. 12 Meurer et al. have reported the synthesis of different 5.6-diarylpyridine carboxylates and carboxamides via cycloaddition of azadiene phosphazene moieties and subsequently studied their human CB1 inverse agonist activity.13 However, most 2-azadienes are found to be quite unstable and their synthesis requires cumbersome experimental procedures. Moreover, synthesis and cycloadditions of conjugated 2azadienes have been little explored.

Thiazolidin-4-ones represent an important class of heterocyclic compounds due to their diverse biological activities.14 Thiazolidin-4-ones have activity profiles, acting as inhibitors of COX-1,15 HIV-RT,16 aldose reductase,17,18 bacterial enzyme MurB and YycG histidine kinase, 19.20 as well as having antidiabetic activity,21 antitubercular, antifungal, and antihelmintic activities.21 Thiazolidin-4-ones have also been explored as useful organic synthons of different heterocycles.²²

There are numerous reports on the acetylenic estermediated synthesis of thiazolidin-4-ones using a variety of thioamides and thioureas.²² However, there are few reports exploring acetylenic ester mediated ring opening/transformations of 4-thiazolidinones. The present manuscript demonstrates acetylenic ester mediated synthetic transformations of dienyl-thiazolidinones21 leading to a facile synthesis of functionalized pyridine-2-carboxylates. In this transformation, the dienyl-thiazolidinones 1a-h behave as masked conjugated 2-azadienes and afford a facile and chemoselective formation of pyridine-2-carboxylates mediated by acetylenic esters in good to excellent yields.

The dienyl thiazolidin-4-ones 1a-h were prepared by amidiolytic ring opening of 2-azetidinones-3-thiazolidin-4ones with sodium alkoxide in the corresponding alcohols (Scheme 1).23 Crystallography data for 1a established the trans conformation of the dienyl thiazolidin-4-one moi-

Scheme 1 Synthesis of dienyl thiazolidin-4-ones²³

We started our investigations by attempting Diels-Alder reactions of dienyl thiazolidin-4-one la with electron-deficient dienophiles (such as methyl acrylate, methyl vinyl



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Quadruple-signaling (PET, ICT, ESIPT, —C=N— rotation) mechanism-based dual chemosensor for detection of Cu²⁺ and Zn²⁺ ions: TRANSFER, INH and complimentary OR/NOR logic circuits



Harminder Singha, Gaurav Bhargavab, Subodh Kumara, Prabhpreet Singha,

Department of Chemistry, UGC-Centre of Advance Studies-II, Guru Nanok Dev University, American-142 005, India to Department of Chemical Sciences, IK Guiral Punjab Technical University, Kapurthala-144 001, Punjab, India

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Complementary logic circuit

ABSTRACT

Herein, we have designed and synthesized o-phenylenediamine-based 4-in-1 dual chemosensor DAU, for simultaneous detection of Cu²⁺ and Zn²⁺ through modulation of PET, ICT, ESIPT and —C=N— rotation mechanisms. The binding of first Cu²⁺ ion with —NEt₂ group causes red shift of absorption band from 372 nm to 420 nm and fluorescence enhancement due to keto- form at 480 nm (~20 fold). When concentration of Cu²⁺ further increased from 1 to 10 equivalents, it results in decrease in absorption intensity at 420 nm and increase in the order of the contrary higher concentrations of Zn²⁺ causes only end shift from 372 to 420 nm however fluorescence remained low due to PET quenching effect. Significantly, when Cu²⁺ is added to DAU-Zn²⁺ (1:5) complex, it results in decrease of absorption intensity at 420 nm and increase in fluorescence at 480 nm followed by a 7-fold increase of the emission intensity at 410 nm. The multiple outputs of DAU can be used for fabrication of INHIBIT, TRANSFER logic devices and complementary OR/NOR logic circuit.

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1. Introduction

Dual chemosensors [1–3], which may or may not possess two separate binding sites, can be useful for detection of two or more analytes through modulation of different signaling mechanisms such as PET (photo-induced electron transfer), ICT (intramolecular charge transfer), FRET (fluorescence resonance energy transfer), AIE (aggregation induced emission), ESIPT (excited state intramolecular proton transfer), blocking of —CH—N— rotation, excimer or exciplex formation, CHEF (chelation enhanced fluorescence effect), TBET (through bond electron transfer) and CHEQ (chelation enhancement quenching effect) [4–9]. In literature, the design of dual chemosensor allows the detection of two analytes at the same time independent of each other (simultaneous approach) [10] or detection of second analyte achieved after breaking of complex between chemosensor and first analyte (displacement approach) [11] (Fig. 1). However, currently the majority of literature reports

are on the dual chemosensors which shows modulation in singlesignaling mechanism for binding of one or two analytes and perhaps, this limits the use of such chemosensors in the fields of multiple ion sensing and molecular computing (fabrication of molecular scale logic devices) [12,13].

In this context, multiple-signaling mechanisms-based dual chemosensor would definitely provide an edge over the single-signaling mechanism-based dual chemosensor in terms of cost effectiveness, efficient analysis in contaminated sites and application in molecular computing [14–16]. Therefore, the control and understanding of relationship between different signaling mechanisms, which are responsible for change in absorbance/emission intensities and wavelengths on coordination with the analyte, would be important in designing the multiple-signaling mechanisms-based dual chemosensor for one or two analytes. Some examples are known where dual signaling mechanism-based chemosensor are used for sensitive and selective detection of one or two analytes [17–23]. These dual-signaling mechanism-based chemo-sensors can be grouped based on the combination of signaling mechanism such as ICT-FRET, PET-FRET, TBET-PET, ICT-PET, ESIPT-AIE or ESIPT-PET, through which they respond to the

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^{*} Corresponding author.

F-mail address: prathipreet.chem@gndu.ac.in (P. Singh).



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Microstructural (self-assembly) and optical based discrimination of Hg2+, CN and Hg(CN)2 ion-pair; Hg2+ promoted-ESIPT assisted guanylation of thiourea



Harminder Singha, Gaurav Bhargavab, Subodh Kumara, Prabhpreet Singha,

ARTICLEINFO

Keywords: Dual chem lon pair ESFPT ntramolecular guenylation Self-assembly

ABSTRACT

A multifunctional self-assembled dual-chemosensor DTS was synthesized and used for optical and micro-structural based discrimination of Hg^{2+} , CN^- ions and $Hg(CN)_2$ ion-pair. The addition of Hg^{2+} ions to the CH₉CN solution of DTS promotes the ESTPT phenomena (emission peak at 458 nm red shifted to 510 nm; increase in absorption between 400 and 480 nm), followed by cyclization (guanylation) reaction with concomitant desulfurization (decrease of emission intensity at 510 am and minor increase at 410 nm; increase and decrease in absorbance intensity at 270 and 325 nm, respectively). In contrast, CN induced deprotonation in DTS results in simultaneous decrease in absorbance at 310 nm and increase in absorbance at 385 nm and 425-475 nm and fluorescence enhancement at λ_{em} 452 nm. On the addition of Hg(CN)₂ ion pair, DTS shows apparent effect of both Hg²⁺ and CN ions. The mechanism of interaction of DTS with Hg²⁺, CN and Hg(CN)₂ has been established by detailed ¹H NMR and ESI-MS studies. DTS in CH₂CN shows flake like morphology which undergoes analyte induced self-assembly to form a bunch of nanorods with Hg²⁺; segregated manorods with curvy ends with CN and spherical structures with Hg(CN)2 ion pair.

1. Introduction

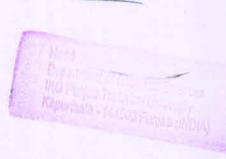
Molecular recognition of toxic metal ions and anions using chromofluorescent methods has received great attention for the past decade because of their involvement in biological, environmental and chemical processes [1,2]. Hg2+ is considered as significant pollutant due to its toxic nature to both animals and plants [3]. The most devastating effect of mercury on aquatic life is due to bacteria mediated conversion of inorganic ${\rm Hg}^{2+}$ into methylmercury (II) which has neurotoxic and neurodegenerative effects on human health [4,5]. The Hg2+ tamination in water and soil even at low concentrations, can lead to problems in digestive system, central nervous system, brain and kidneys [6-8]. Cyanide is a highly toxic anion as its 0.5-3.5 mg/kg body weight is sufficient to cause death in human due to its strong interaction with a heme unit at the active site of cytochrome C and this, interferes with the oxygen supply in the body [9,10]. In recent years, many different strategies for detection of CN have been developed in the literature which include hydrogen bonding interactions, metal coordination (due to strong affinity of CN $^-$ towards ${\rm Hg}^{2+}$ and ${\rm Cu}^{2+}$ etc.) and nucleophilic addition of CN towards -C=C-, -C=N- and -C=O [11-14]. On the other hand, many fluorescent chemosensors that bind Hg²⁺ ions have been developed based on the mechanisms which include Hg2+induced desulfurization followed by guanylation, hydrolysis, ring opening reactions and chelation via S atoms [15-22],

The majority of current literature reports are based on single target of interest such as Hg²⁺ or CN⁻ ions [11–22]. However, simultaneous detection of multiple targets such as Hg2+ and CN ions using chromofluorescent probe is of prime importance in supramolecular chemistry. In this context, dual chemosensors, which possess two different binding sites, could be used for individual or simultaneous detection of two analytes at the same time independent of each other [23-25]. Dual chemosensors, which show responses for multiple targets at different wavelengths through the modulation of multiple-signaling mechanisms are still in demand [26-30]. Moreover, if the interaction of a specific single or multiple analyte with self-assembled aggregates of dual chemosensor occurs, it would also result in change of microstructural features (morphology) of the aggregates to mark the molecular recognition process [31-33].

Recently, our group produced first report of triple-signaling mechanisms-based 3-in-1 multichannel chemosensor for recognition of Cu2+, AcO" and Cu(OAc)2 ion pair [34]; individual and simultaneous detection of Cu2+ and P- ions [35]. The multi outputs were further

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Department of Chemistry, Guru Nanek Dev Uriversity, Amritsa: 143 005, India
 Department of Chemical Sciences, IK Gujrd Punjab Technical University, Kapurthala 144601. India

^{*} Corresponding author.

F-mult address: praishpress chemigriculació (P. Singh).

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Letter

Dissymmetric Bay-Functionalized Perylenediimides

Gauray Bhargaya Suhodh Kumara Prabhpreet Singh***

Department of Chemistry, UGC Centre for Advanced Studies, Curu Nanak Dev University, Amiltser 143 005, India prabhpreet, chemeligndu ac.in Department of Chemical Sciences, IK Cujiral Punjab Technical University, Kapurthala-144601, Punjab, India

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Abstract We report the synthesis of perylenedlimide (PDI)-based donor-acceptor hybrids through dissymmetric boy functionalization of PDIs. The dissymmetric boy functionalized di- and trisubstituted PDIs were characterized by using one- and two-dimensional NMR spectroscopy. Density functional theory calculations revealed (i) an energy gap between the HOMO and LUMO in the range 2.14–2.34 eV, beneficial for charge-transfer properties; and (ii) a twist angle between the two naphthalene units in the range 17-26*, which might be beneficial for disruption of co-facial stacking of the PDI.

Key words perylenediimide, dissymmetric functionalization, boy functionalization, nitration, density functional theory

Perylenediimide1-4 (PDI), an important member of the rylene family, is a promising scaffold owing to the prominent features of PDIs such as their excellent optical, chemical, and thermal robustness; high fluorescence quantum yield; high electron mobility; and high molar absorptivity. Moreover, the optical, electronic, and aggregation properties of PDIs can be modified by judicious selection of functional groups at the N-imide, bay, ortho, or peri positions of the PDI core.2-6 Therefore, PDIs have been explored for various applications in the dye and pigment industries5 and in optoelectronic devices. 6-10 Recently, bay- and N-terminalfunctionalized PDIs have been explored in supramolecular chemistry.11-17 In this context, PDI-based chromophores with attached donor and acceptor substituents (D-A hybrids) might be potential candidate for use as reactionbased probes for producing long-wavelength-absorbing dyes18 and in the field of organic photovoltaic devices.19

In literature reports, basically, either two (1.7-) or four (1,6,7,12-) positions of the bay area of PDIs (A, Scheme 1) have been extensively used for homogeneous functionalization through nucleophilic substitution or Suzuki/Sonogashira coupling reactions of di- or tetrahalogenated derivatives in the presence or absence of a catalyst, to attach various aryl or alkyl groups to the PDI core (B and C; Scheme 1),1-6 PDIs with two different functional groups in the bay positions can be prepared by multiple stepwise substitution reactions of halogenated PDIs, but these syntheses involve cumbersome purification procedures and give the desired product in low to moderate yield (D; Scheme 1).20 Few reports have appeared that discusses halogenated or nonhalogenated routes for the synthesis of dissymmetric bayfunctionalized di- or trisubstituted PDIs (E; Scheme 1). We therefore proposed that PDI-based D-A hybrids might be synthesized by dissymmetric bay functionalization of the PDI core.21

Scheme 1 Ceneral approaches for the synthesis of dissymmetric functionalized PDIs reported in the literature



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AIE+ESIPT based red fluorescent aggregates for visualization of latent fingerprints

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Harminder Singh[®], Rashmi Sharma^b, Gaurav Bhargava^c, Subodh Kumar^{a*} and Prabhpreet Singh^{a*}

fluorescent molecules are extensively employed in (bio)sensing and bioimaging but their applications in visualization of fingerprints are less reported due to their inherited aggregation-caused quenching (ACQ) mechanism in solid state (aggregated form). DPSA undergoes self-assembly to form aggregates with rod-like structures in 90% H₂O:CH₂CN mixture evident from dynamic light scattering (DLS) and scanning electron microscopic (SEM) techniques. Self-assembled diphenylpyrimidinone derivative (DPSA) shows aggregation-induced emission (~24 fold) in >90% H₂O:CH₂CN mixture with orange-red fluorescence (A_{max} 564 nm). The red fluorescent aggregates of DPSA show applications in visualization of latent fingerprints on aluminium, stainless steel and metal currency coin substrates. In addition, we established the individuality of three subjects based on information gathered at level one (pattern of arch, loop or whorl) and level two (minutae details of dots, bifurcation, ridge ending, core and delta).

Introduction

The papillary ridges and furrows present on fingers and palms offer a unique identity to every individual. The papillary ridges are a fine network of minute ridges formed by skin folding whereas depression between adjacent ridges is known as furrow. These papillary ridges have small apertures at their surface which are connected to sweat glands existing underneath the skin and these glands persistently secrete sweat with 99.5% of its volume being water and rest (0.5%) being organic (mainly consists of sebum) and inorganic substances. However, the sebum content of these papillary ridges is escalated if we make contact of fingers on face, neck and hair follicles. Afterwards, if contact of these sebum rich papillary ridges be made with any surface (aluminium, mica, coins, tiles etc.), it could leave a pattern (mirror image) of fine ridges of fingers on that surface. This pattern of fine ridges is called latent (invisible) fingerprints.

Latent fingerprints (LFPs) collected from crime scenes are considered as very useful evidence in forensic investigations. The routinely used procedures for the visualization of latent fingerprints includes powder dusting², vacuum metal deposition³, dye adsorption on nanoparticles⁴⁶, quantum dots (QDs) and staining using chemicals such as iodine, ninhydrin and cyanoacrylate⁸ Presently, development of fingerprints

using powder dusting is widely used method in forensic analysis. However, these techniques have limitations such as destruction of LFPs during washing or cleaning of LFP impressed surfaces, high toxicity of used chemicals and low resolution of the developed LFPs. Recently, spectroscopic techniques such as Raman 9-10, FT-IR 12-12 and mass 13 are also used for visualisation of fingerprints but requirement of heavy instrumentation limits the scope of their use.

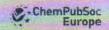
In recent years, combination of aggregation induced emission (AIE) and excited state intramolecular proton transfer (ESIPT) mechanisms provided a new paradigm for developing application based fluorescent materials. Both AIE and ESIPT process are important phenomena and the participation of both in a molecule can provide high fluorescence in solid/aggregate state. Therefore, fluorescence based methods involving use of molecules possessing AIE+ESIPT could be more useful for visualisation of LFPs on wet non-porous surfaces. Probably, these hydrophobic aggregates of AIE+ESIPT fluorophores could bind preferentially to the sebum rich papillary ridges of fingerprints by virtue of hydrophobic interactions. On illumination with 365 nm UV lamp, the hydrophobic aggregates would light up and minute details of papillary ridges can be visualised. 14-19 However, this field is still growing and to the best of our knowledge, only few reports are available in the literature 14,20-21.

In last few years, we explored perylenedimide, peptides and CNTs based self-assembled materials for application in biosensing, bioimaging, molecular computing and catalysis. 14. 23-33 Recently, we have shown that the attachment of salicylideneamine can make the diphenylpyrimidinone (an interesting class of heterocyclic derivative based on pyrimidine nucleobase) be AlE-active, for example, DPPS-1 (Fig. 1). It

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^{*} Department of Chemistry, UGC Centre for Advanced Studies-II, Gurus Nanok Dev Department of Chemistry, Osc. Centre for Advances statues in ourse ratios of University, Amritsar 143 005, India, e-mail: grathipreet.chem.@gndu.gc.in (F. Singh); Tel. +91-84271-01534 e-mail: subodh.chem@gndu.gc.in (S. Kumar).

Deportment of Chemistry, Trinity College, Jaiondhar 144009, Punjah, India Department of Chemical Sciences, it Gujrul Punjab Technical University, Kapurthalu-144601, Punjab, India. (Electronic Supplementary Information (ESI) available: [Synthesis, characterization, Photophysical properties of DPSA] See DOI: 10.1039/s0xx000000



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Organic & Supramolecular Chemistry

ESIPT-Based Dual Chemosensor for Sequential Detection of Cd2+/Zn2+ and Nucleoside Triphosphates in Water: Application in Logic Gates

Harminder Singh, at Rashmi Sharma, bi Gauray Bhargaya, c Subodh Kumar, and Prabhpreet Singh*[a]

Here in, novel excited state intramolecular charge transfer (ESIPT)-modulated fluorophore based on diphenylpyrimidinone DPHQ has been designed and synthesized. DPHQ shows selective 8-fold 'turn-on' fluorescence response with Cd2-1Zn2+ in buffer (containing 1% CH,CN, pH 7.2) with low detection limits of 55:7/2.7 nM accompanied with a fluorescent colour. change from colourless to cyan. DPHQ-Cd² ensemble undergo selective fluorescence quenching with nucleoside triphosphates but it is silent to nucleoside di-and mono-phosphates. It can detect 58.4 nM concentration of adenosine triphosphate (ATP). The binding mode and sensing mechanism of DPHO with Cd2-/Zn2+ was supported by the change in optical behaviors, density functional theory (DFT) calculations and Job's plot. DPHQ was successfully used for construction of 'Set-Reset' memorized logic devices and OR, INHIBIT logic gates. As practical applications of DPHQ, we utilized DPHQ for detection of Cd2+/Zn2+ in blood serum and quantification of Zn2+ in tap

Introduction

Transition metal ions have persistent, non-biodegradable nature and become toxic if accumulated into a living system over the permissible limits.[1] Cd2+ metal ion is carcinogenic in nature and its over-exposure results in renal and respiratory disorders such as reduced lung capacity, emphysema, bone disease, heart disease, [2-3] Zn2- ions function as a cofactor for several metallo-proteins which are vital for immune response, neurotransmission and apoptosis suppression processes.[40] However, excess amounts of Zn2+ ions in the body disrupt the body homeostasis due to the suppression of copper and iron intake in conjunction with other disorders such as Alzheimer's disease and diabetes. So, it is important to detect and monitor the concentration of these toxic metal ions in the environment.

Phosphates are ubiquitously present in living cells and perform vital functions for efficient working of biological processes in the body^[6] such as DNA replication and transcription essential for cell division.[7-8] Nucleoside polyphosphates (NPPs) i.e. mono-, di- and triphosphates of adenosine, guanosine, thymidine, uridine and cytidine constitute a class of biological phosphates which are compulsory for the normal functioning of cellular system. 91 Nucleoside triphosphates have fundamental role in cellular respiration including glycolysis no and Kreb's cycle^[1] and it is also involved in other important processes such as ion channel modulation 112 and signal transduction.[13] The deficiency of nucleoside triphosphates can cause critical illness such as ischemia, hypoglycaemia and Parkinson's disease and thus the monitoring of the nucleoside triphosphates becomes paramount importance.[14] The commonly employed techniques for determination of Nucleoside triphosphates such as adenosine triphosphate (ATP) include luciferin-luciferase bioluminescence assay and sequential enzyme reactions coupled with an electrochemical sensor.[15-16] However fluorescence based methods utilizing small organic molecules for detection of nucleoside triphosphates can provide advantage of fast response and real-time detection.

Many chemosensors based on pyrene, naphthalimide, anthracene, coumarin and other moieties[17-21] with colorimetric and fluorimetric response have been designed up to now for the recognition of phosphate using anion coordination chemistry involving H-bonding and electrostatic interactions. However, the workability of such receptors in the highly competent aqueous medium is difficult due to the strong Hbonding ability of these receptors with water molecules which sometimes supersedes the ability of phosphate to bind to receptor site.[22] Alternatively, in recent years, the use of metal receptor ensemble as a platform for detection of anions has become important paradigm where target anion displaces the metal ion from the receptor pocket due to higher metal anion affinity. Since nucleoside triphosphates show strong binding preference with metal ions such as Zn2+ and Cd2+ ions therefore, metal-ligand ensembles incorporating these ions could be used as tailor made receptors for the detection of nucleoside triphosphates.[23-24]

[[]a] Dr. H. Sinah, Prof. Dr. S. Kumar, Dr. P. Sinah Department of Chemistry, UGC Centre for advanced studies -II, Guru Nanak Dev University, Amritsar 143005, India E-mail: prabhpreet.chem@gndu.ac.in

Department of Chemistry, Trinity College, Jalandhar 144006, India

⁽c) Dr. G. Bhargava Department of Chemical Sciences, IK Gujral Punjab Technical University, Kapurthala 144601, India

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Multifunctional metallo-supramolecular interlocked hexagonal microstructures for the detection of lead and thiols in water?

Prabhpreet Singh, 🕒 ** Lalit Singh Mittal, * Kapil Kumar, * Poonam Sharma, * Gauray Bhargaya^b and Subodh Kumar

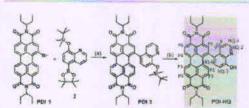
We report Pb2+-mediated interlocked hexagonal self-assembly associated with the near-IR (660 nm) 'turn-on' detection of Pb2+ ions (LOD = 25 nM) in water using a perylene diimide derivative (PDI-HQ). The PDI-HQ—Pb2* complex was further used to develop a fluorescent NIR ensemble for the detection of thiols.

Metallo-supramolecular self-assembly is a term used to describe metal-mediated self-assembly.1 In this area, the emphasis is on the design and synthesis of receptors which possess the correct number and spatial arrangement of coordination (donor) sites, whereas the metal ions which are being used as a template must have the correct coordination number and stereochemical preference. This metal-directed self-assembly has led to the development of fascinating architectures of boxes, triangles, grids and helicates.2.3 The metal-directed self-assembly of bay functionalized perylene diimide (PDI) dyes has been only recently investigated.4 PDIs and their derivatives have been valued for a long time due to their unique properties and case of functionalization at the so called bay-, N-terminal-, ortho- and peri- positions.3 Current available literature has shown that N-terminal substituted PDIs resulted in the formation of self-assembled nanostructures6 including fibers, ribbons, rods and tubes. However, the main challenge is to control the self-assembly of bey substituted PDIs and exploration of the molecular recognition properties of these self-assembled PDIs in water.7

Heavy metal ions (Pb2+, Hg2+ and Cd2+) have non-degradable nature and possess detrimental effects and become a threat to global sustainability if accumulated in the ecosphere (air, water and soil) over the permissible limits. 16,31 Lead (Pb2*) is an environmental pollutant and can cause physical growth impairments, irritability in children and reduced IQ levels 12,13 and can

have adverse effects on human organs. Thiol-containing amino acids such as cysteine (Cys) and homocysteine (HCy) play important roles in biological processes such as protein synthesis, metabolism and growth of cells and tissues. The alteration of Cys levels in living systems can cause hair de-pigmentation, weakness, lethargy, skin lesions and slow growth rates.15-1 Traditional methods^{17,18} involve the usage of ionic solvents, heavy instrumentation, trained operators and lack of portability. Therefore, fluorescence based methods for the detection of Pb2' ions and thiols are highly demanded.19

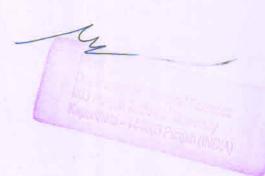
Herein, we have designed and synthesized PDI-HQ (Scheme 1) and Fig. 1 shows the overall detection strategy of Pb2s and thiols using PDI-HQ. (1) In an aqueous solution, PDI-HQ forms H-aggregates of diameter 50-160 nm and these aggregates are non-fluorescent due to the combined effect of H-aggregation and PET from the 8-hydroxyquinoline moiety. (2) When the Pb2+ ions are added to the aqueous solution of PDI-HQ, strong interaction between 8-HQ and Pb2+ results in the formation of metallo-supramoiecular aggregates having interlocked hexagonal self-assembly with diameter in the range of 0.5-1.7 µm and thickness ~250 nm. As a result of binding of Pb2* with PDI-HQ, the PET process is effectively inhibited and a consequent increase in the fluorescence intensity of aggregated species is observed at 660 nm. (3) When a thiol is added to the PDI-HO-Pb2 ensemble.



Scheme 1 Synthesis and chemical structure of PDI-HQ (also showing the ary designation of protons); Reagents and conditions: (a) Pd(PPh.). Na₂CO₃, Toluene: H₂O (2: L, v/v), 70 °C, 10 h, (b) 10 M solution of tetra-butylammonium fluoride (TBAF), THF, RT, 24 h.

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[&]quot;Department of Chemistry, UGC Centre for Advanced Studies-U Guru Nanak Dev University, Amritsar 143 000, India.

E-mail: prohipmer.chem@gndu.ac.in; Tel: +97-84271-01534

*Department of Chemical Sciences, IK Gupul Punjab Technical University, Kapurthala-144601, Punjah, Italia

[†] Electronic supplementary information (ESI) available: Synthesis, spectral date. and photophysical data of PIN-IN) with Pb and Cys. See INE 10.1039/c8cr05814b

■ Organic & Supramolecular Chemistry

[2+2] Cycloadditions of Sorbyl Tosylate with Imines/1-Azadienes: A One-Pot Domino Approach for α -Alkylidene- β -lactams and Their Computational Studies and Antimicrobial Evaluation

Yogesh Kumar, [a, b] Preet Mohinder Singh Bedi, [d] Prabhpreet Singh, [d] Adebayo A. Adeniyi, [e] Ashona Singh-Pillay, [f] Parvesh Singh, [f] and Gaurav Bhargava*[a]

(Dedicated to Prof. M. P. Mahajan on the occasion of his 71st Birthday)

The manuscript describes a straightforward and atom-efficient method for the synthesis of a-alkylidene β -lactams using sorbyl tosylate and imines/1-azadienes at high temperature (80 °C). The Density functional theory calculations have shown the prevalence of the first order kinetics in these [2+2] cyclo-

additions to produce mixture of 3-butadienyl-azetidin-2ones and 3-but-2-enylidene-azetidin-2-ones in good yields. The 3-but-2-enylidene-azetidin-2-ones have also shown antimicrobial activity against the E. coli. S. aureus, P. aeruginosa, B. cereus and B. subtilis.

Introduction

Ketenes are versatile intermediates in organic synthesis. ^[1] There are numerous reports on synthesis and cycloadditions of functionalized ketenes for the synthesis of heterocyclic systems of biological relevance. ^[2-3] The [2+2] cycloadditions of ketenes with alkenes or iminic systems have widespread been utilized for the synthesis of carbo– and heterocyclic systems respectively. ^[3] There has been significant interest and controversies over the mechanism of ketene-imines cycloadditions. Experimental work as well as theoretical studies have been made on the reactions of the imines with simple ketenes. ^[4] It is well understood that the Staudinger reactions involving [2+2]

cycloadditions of ketene and imines are proceeded via zwitterionic intermediates as shown in the Figure 1,^[5]

Figure 1. (A) Orbital Interactions in concerted cycloaddition. (B) Stepwise mechanism, con= connotatory

 [a] Dr. Y. Kumar, Dr. G. Bhargavo
 Department of Chemical Sciences, L.K. Gujral Punjab Technical University, Kapurthala, Punjab-144603, India
 E-mail: gauravorganic@mail.com

 UNAM-National Nanotechnology Research Center, institute of Materials Science and Nanotechnology, Department of Chemistry, Bilkent University, 06800 Turkey

[c] Dr. P. M. S. Bedi Department of Pharmaceutical Sciences, Guru Nanak Dev University, Amritsar-143005, India

[d] F. Singh
 Department of Chemistry, Guru Nanak Dev University, Amritsar-143005, India

[e] A. A. Adeniyi Pharmacy Department, University of KwaZulu Natal, Westville Campus, Durban, Chemistry Department, University of Oye-Ekiti, Ekiti State, Nigeria 10, 4, Signit-Politor, Or. P. Stoch.

[f] A. Singh-Pilloy, Dr. P. Singh School of Chemistry and Physics, University of KwaZulu Natal, P/Bag X54001, Westville, Durban 4000, South Africa E-mail: singhp4@ulgnac.za

Supporting information for this article is available on the WWW under https://doi.org/10.1002/sicz.201801605 However, the corresponding studies on [2+2] cycloaddition reactions involving conjugated ketene such as butadienyl-ketene still need to be explored. Earlier Mahajan et. al. have explored the [2+2] and [4+2] cycloaddition reactions of butadienylketene generated in situ from sorbyl chloride and triethylamine, with imines and 1,3-diazabuta-1,3-dienes respectively. The reaction resulted in the formation of cis- and transbutadienyl-azetion-2-ones. However, studies on cycloaddition reactions involving butadienylketene with iminic systems at high temperature and using alternative methods for in situ generation of butadienylketene still need to be studied. On the other hand, α -alkylidene- β -lactamase inhibitors such as Ro 15–1903, asparenomycins, 6-(2'-pyridyl)methylene penem sulfone,

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Coronene diimide-based self-assembled (fibre-to-disc) fluorescent aggregates for visualization of latent fingerprints

Kapil Kumar^a, Harminder Singh^{a,b}, Vanita Vanita^c, Ramesh Singh^d, Khashti Ballabh Joshi^d, Gaurav Bhargava^e, Subodh Kumar^a, Prabhpreet Singh^a

Department of Chimistry, UGC Centre for Advanced Studies, Guru Nanak Dev University, Amritian, 143005, India

*Department of Chemistry, Indian Institute of Technology, Kampur, UP, India
*Department of Chemistry, Indian Institute of Technology, Kampur, UP, India
*Department of Human Gouetics, Guru Nanak Dev University, Auriture Panjab, 143 005, India
*Department of Chemistry, H.S. G. Central University Sugar, M. P., 470003 India
*Department of Chemistry, H.S. G. Central University Sugar, M. P., 470003 India
*Department of Chemical Sciences, IK Gujral Punjah Technical University, Kapurthala, 144 601, Punjah, India

ARTICLEINFO

Keywords. Coronenediimides Fibrillar self-assembly Latent Engorpriess Powder dusting Biolabeling

ABSTRACT

Coronenedilmide (CDI 3) forms long fibrillar nonostructure when dissolved in DMF. On addition of > 70% water in DMF, CDI 3 undergo fibre-to-disc morphological transformation alongside onset of different emission colors in water-DMF binary mixtures. These self-assembled red emitting aggregates of CDI 3 applied for visualization of latent fingerprints using solution spray and powder dusting methods on porous and non-porous surfaces commonly found in home such as aluminium, stainless steel, glass, metal coins, compact discs (CDs), OHP films, ceramic tiles and paper sheets. In an attempt to explore the application of CDI 3 in the forensic cases, 2rd level of information was used for distinguishing different flagerprints from different subjects. These disc-like fluorescent red aggregates of CDI 3 were also used for staining of HeLa cells.

1. Introduction

Many n-conjugated molecules with extended polycyclic framework have been developed by research groups [1-4]. Among them, coronene diimides (CDIs), [5] which can be considered as bay-extended perylene diimides (PDIs), [3-4] have gained interest due to their excellent thermal and oxidative stabilities, high molar extinction coefficient, fluorescence quantum yields and self-organization properties [6-13]. The benzene and heterocycle ring annulation at the bay-region of the perylene core leads to coronenediimide derivatives which offers applications in optoelectronics such as organic field effect transistors (OFETs), fluorescent solar collectors, organic light emitting diodes (OLEDs) and optical lasers [6-13]. Despite many synthetic efforts of modification in CDIs, little progress has been made in making CDIs water soluble [1-2] and preserving its fluorescence (far-red to nearinfrared region) [1-2] in solid and solution state for application in the fields of designing AlEgens, visualization of latent fingerprints and biolabeling.

Latent (invisible) finger prints (LFPs), which are made up of papillary ridges and furrows present on fingers and palm offer a unique identity to each individual [14-15]. Such LPPs collected from crime scenes are considered as useful physical evidence and comerstone to identify individuals. Currently, powder dusting method, quantum dots, nanoparticles and chemical methods were applied in visualization of LFPs, however potential toxicity and destructive nature, short pre servation time, low resolution and expensive procedures limit their wider applications [16-17].

In continuation of our interest in PDIs, [18-27] herein, we report CDI 2 and CDI 3 as new additions in the rylene family. Due to onset of aggregation of molecules of CDI 3 with increasing water fraction (fw). the nano-fibres of CDI 3 in DMF are eventually transformed into disclike aggregates in > 70% water-DMF mixtures. These disc-like aggregates of CDI 3 both in water-dispersed and solid-state are fluorescent red and were used for visualization of LFPs imprinted on common household items and for staining HeLa cells.

2. Experimental section

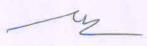
2.1. Materials and mediods

Chemicals and solvents were of reagent grade and used without further purification unless otherwise stated. Chromatographic purification was done with silica gel 60-120 mesh. TLC was performed on aluminium sheets coated with silica gel 60 F254 (Merck, Darmstadt).

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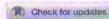


^{*} Corresponding author. E-mail address: probbpreet chem@gndu.ac.in (P. Singh).



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REVIEW



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Transition metal catalyzed [6 + 2] cycloadditions

Amit Anand,** Prabhpreet Singh, * Vipan Kumar* * and Gaurav Bhargava* *

The [6+2] cycloaddition reactions are one of the important synthetic transformations to construct eight membered carbo-/heterocyclic systems. The present review is an attempt to update readers on transition metal catalyzed [6+2] cycloaddition reactions of various 6π -contributing substrates such as cycloheptatrienes (CHT), cycloactatetrenes (COTT), allenals, vinylcyclobutanones, futurene etc. employing rhodium, cobalt, titanium, copper, platinum, ruthenium, rhenium and diphenylprolinolsityl ethers etc. as catalysts. The transition metal catalyzed [6+2] cycloaddition reactions with a variety of functionalized substrates provide straightforward access to eight membered cyclic and/or 5/8, 6/8 etc. condensed carbo-/heterocyclic molecules in moderate to good yields.

Introduction

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Cycloaddition reactions are considered to be one of the most powerful methods for the construction of cyclic skeletons' with atom economy. There are ample reports on different variants of [4+2] and [3+2] cycloadditions which provide an easy access to a variety of five or six membered carbo-/heterocycles.23 However, the parallel methodologies such as [5 + 2], [6 + 2], [8 + 2]sb etc. for the synthesis of seven, eight or ten membered carbo-/heterocycles are still uncommon in the literature. Eightmembered rings are found in a wide variety of natural products and are useful ring systems available in variety of drugs, drug leads, or biological probes.6 Taxol, the well-known natural product in this family, is now among the most potent anticancer drug in clinical use. In 2008, another member of this family, pleuromutilin was approved for use as an antibiotic by Food and Drug Administration (FDA)' (Fig. 1). Several methods8-12 including ring-closing metathesis and Cope rearrangement have been reported for the synthesis of eight-membered carbocycles,13 However, the intolerance to substrate substituent in conventional procedures along with complexities observed in natural products makes these techniques ineffective and instigated the development of alternate protocols. Recent years have seen a significant upsurge in this direction with relatively few reports appearing on the synthesis of eight membered carbocycles using [6 + 2]5.14 and [4 + 2 + 2]11 cycloaddition reactions. Concerted [6 + 2] cycloaddition reactions are forbidden thermally as per Woodward-Hoffmann rules, when the two reacting π-systems add in a suprafacial fashion. However, transition metal-catalyzed [6 + 2] cycloadditions constitute one such protocol for the synthesis of monocyclic or condensed functionalized cyclooctadienes. The present review summarizes various reports on the transition metal catalyzed inter- and intra-molecular [6 + 2] cycloaddition reactions appeared in literature since 2000.¹⁴

Rhodium catalyzed [6 + 2] cycloadditions

Rhodium complexes plays a pivotal role in organic synthesis as major catalytic contributors to C-C bond formation reactions.15 A number of carbo-/heterocyclic systems have been generated employing rhodium complexes as efficient catalyst in a variety of cycloaddition/cyclization reaction* Literature rationale suggests an inspiring advancement in this field and many research groups are extensively using rhodium complexes to efficiently conceive these reactions.12 Considering the importance of rhodium catalysed cycloadditions, there should have been a number of reports on generation of eight membered carbo-/heterocycles engaging rhodium complexes in [6 + 2] cycloadditions, however only a few reports are available which successfully comprehends [6 + 2] cycloadditions. Till now, different substrates such as cycloheptatriene, 2-vinylcyclobutanones, allenals, allenylcyclobutanes, cyclooctatrienes etc. have been reported as 6π-components in Rh(i) catalysed [6+

Fig. 1 Carbocycles in natural products.

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[&]quot;Department of Chemistry, Khalsa College, Amritsas, 743005, Punjah, India. E-mail: aa. wiz@yahov.com

^{*}Department of Chemistry, Guru Nanak Dev University, Ameritsus-143005, India

^{*}Department of Chemical Sciences, IKG Punjab Technical University, Kapurthala, 144603, Punjab, India. E-mail: gaurav@ptu.ac.in: gauravorganic@gmail.com

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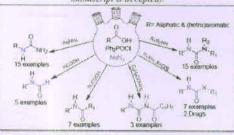
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Direct Conversion of Carboxylic Acids to Various Nitrogen Containing Compounds in One-Pot Exploiting Curtius Rearrangement

Arun Kumar¹, Naveen Kumar¹, Ritika Sharma², Gaurav Bhargava² and Dinesh Mahajan¹*

¹Drug Discovery Research Center, Translational Health Science and Technology Institute, Faridabad, 121001, India ²Department of Chemical Sciences, I. K. Gupral Punjab Technical University, Kapurthala, Punjab-144603, India

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ABSTRACT: Herein we report, a single pot multi-step conversion of inactivated carboxylic acids to various N-containing compounds using a common synthetic methodology. The developed methodology rendered the use of carboxylic acids as direct surrogate of primary amines, for the synthesis of primary ureas, secondary/tertiary ureas, O/S-carbamates, benzoyl ureas, amides and N-formyls, exploiting Curtius reaction. This approach has a potential to provide a diversified library of N-containing compounds starting form a single carboxylic acid, based on selection of the micleophile.

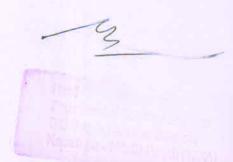
Nitrogen containing compounds such as amines, amides ureas, acyl ureas, and carbamates are ubiquitous in natural products, agrochemicals as well as pharmaceutical agents or molecules of biological and commercial interests. Hence, synthesis of these various N-containing compounds is an important objective in organic synthesis. Carboxylic acids are favorable substrates because of their broad structural diversity, ease in availability, high stability, low toxicity along with ease in storage and handling.1 Curtius rearrangement of acyl azides is a powerful method for construction of a new C-N (sp2C-N or sp³C-N) bond using azide as a source of nitrogen and carboxylic acid as source of carbon.2 The conversion of carboxylic acids to nitrogen containing compounds, avoiding use of any organic amme as a primary source of nitrogen is very alluring In a typical Curtius reaction, an acyl azide is thermally degraded to rearrange as an isocyanate.3 The acvl azide is often synthesized in a different pot as a separate step from a pre-activated carboxylic acid and sodium azide.24 The isocyanate thus formed can be reacted with water, alcohol or amine as one of the nucleophile to yield corresponding amine carbamate or urea as a final product. This makes whole process multi-pot and practically cumbersome, if one aims for conversion of carboxylic acids to N-containing compounds. One pot multi-step conversion provides an effective approach

to infuse economic and environmental efficiencies because several synthetic transformations and bond-forming steps can be carried out in a single pot. Yamada et al., reported the first one-pot conversion of inactivated carboxylic acids to urethanes involving Curtius reaction using DPPA (diphenyl phosphoric azide). The generation of DPPA needs an additional step in a different pot involving sodium azide and diphenyl phosphoryl chloride. Recently, Lebel et al. reported one-pot conversion of carboxylic acids employing, either chloroformate or di-tert-butyl dicarbonate along with sodium azide as a replacement of DPPA. However, these methodologies are confined to conversion of carboxylic acids to corresponding amines, ureas or carbamates only.

Herein, we want to report a direct, multi-step, single pot conversion of carboxylic acids to corresponding primary ureas, mixed ureas (secondary/tertiary). O-carbamates, Scarbamates, acyl ureas, N-formyls and amides exploiting a single methodology.

Initial optimization efforts were focused on conversion of benzoic acid to corresponding 1-phenylurea (Table 1). Primary ureas are not only found to be bioactive molecules but they also serve as starting materials for molecules of biological and commercial interests's Often, primary ureas are synthesized from corresponding organic amines and ammonia involving phosgene, carbamovl chloride or isocvanates.9 Interestingly. there is no report for direct conversion of carboxylic acids to corresponding primary ureas in one pot. This instigated us to start our optimization efforts for synthesis of 1-phenylurea (1b), as a model reaction from corresponding benzoic acid (1) In a typical procedure, benzoic acid was mixed with DPPA (entry 1, Table 1) at 0°C, using toluene as a solvent. An equivalent amount of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) was added and reaction was stirred for 30 minutes at room temperature to facilitate the formation of acvl azide. The conversion of the acvl azide was monitored as well as quantified, before heating the reaction to 90°C. Aqueous ammonium hydroxide was added to this heated solution and reaction was monitored by TLC and ¹H NMR for the formation of 1-phenylurea as a final product. Use of DPPA as a reagent of choice provided 32% isolated yield of 1phenylurea, albeit corresponding benzoyl azide intermediate was formed in reasonably good amount (Entry 1, Table 1) DPPA is a commercial grade reagent which is generally obtained by vacuum distillation of the crude product obtained after a reaction of diphenyl phosphoryl chloride with sodimu azide.7 We tried to replace DPPA by in situ use of diphenyl phosphoryl chloride and sodium azide (entry 2). The reaction of benzoic acid was repeated and replacement of DPPA was compensated by in situ addition of diphenyl phosphoryl chloride, sodium azide (as a solution in dry DMF) and catalytic amount of DMAP. In new reaction involving sodium azide, the conversion to corresponding benzoyl azide was less however it was pleased to see similar isolated yield for final product 1b (entry 2 vs entry 1). As clear from data tabulated in table 1, a quick screening of few other similar reagents concluded, diphenyl phosphinic chloride (Ph₂POCI) as a reagent of choice for better yields of benzoyl azide as well as 1-phenylurea (entry 3). Decent success in overall conversion with methanesulfonyl chloride (entry 5) and reaction feasibility with POCl₃ (entry 4) needs special mention as these two are common reagents in organic chemistry laboratories Ammonium chloride or ammonium formate (entry 6 and 7) were also found to be two other potential sources of ammonia for the desired conversion. Influenced by the outcome of this

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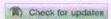


RSC Advances



REVIEW

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Transition metal catalyzed [6 + 2] cycloadditions

Cite this: RSC Adv 2019 9 25554

Amit Anand,*a Prabhpreet Singh, ab Vipan Kumar *b and Gaurav Bhargava *c

Received 15th April 2019 Accepted 2nd August 2019 DOI: 10.1039/c9ra02839k rsc.ii/rsc-advances The 6+2l cycloaddition reactions are one of the important synthetic transformations to construct eight membered carbo-/heterocyclic systems. The present review is an attempt to update readers on transition metal catalyzed 16+2l cycloaddition reactions of various 6π -contributing substrates such as cycloheptatrienes (CHT), cycloactatetrenes (COTT), allenats, vinylcyclobutanones, fulvene etc. employing rhodium, cobalt, titanium, copper, platinum, ruthenium, rhenium and diphenylprolinolsityl ethers etc. as catalysts. The transition metal catalyzed 16+2l cycloaddition reactions with a variety of functionalized substrates provide straightforward access to eight membered cyclic and/or 5/8, 6/8 etc. condensed carbo-/heterocyclic molecules in moderate to good yields.

Introduction

Cycloaddition reactions are considered to be one of the most powerful methods for the construction of cyclic skeletons1 with atom economy. There are ample reports on different variants of [4+2] and [3+2] cycloadditions which provide an easy access to a variety of five or six membered carbo-/heterocycles.23 However, the parallel methodologies such as [5+2], [6+2], [8+2]36 etc. for the synthesis of seven, eight or ten membered carbo-/heterocycles are still uncommon in the literature. Eightmembered rings are found in a wide variety of natural products and are useful ring systems available in variety of drugs, drug leads, or biological probes.º Taxol, the well-known natural product in this family, is now among the most potent anticancer drug in clinical use. In 2008, another member of this family, pleuromutilin was approved for use as an antibiotic by Food and Drug Administration (FDA)2 (Fig. 1). Several methods8-12 including ring-closing metathesis and Cope rearrangement have been reported for the synthesis of eight-membered carbocycles.14 However, the intolerance to substrate substituent in conventional procedures along with complexities observed in natural products makes these techniques ineffective and instigated the development of alternate protocols. Recent years have seen a significant upsurge in this direction with relatively few reports appearing on the synthesis of eight membered carbocycles using $[6+2]^{5,44}$ and $[4+2+2]^{14}$ cycloaddition reactions. Concerted [6 + 2] cycloaddition reactions are forbidden thermally as per Woodward-Hoffmann rules, when the two reacting π -systems add in a suprafacial fashion. However, transition metal-catalyzed [6 + 2] cycloadditions constitute one such

protocol for the synthesis of monocyclic or condensed functionalized cyclooctadienes. The present review summarizes various reports on the transition metal catalyzed inter- and intra-molecular [6 + 2] cycloaddition reactions appeared in literature since 2000.¹⁹

Rhodium catalyzed [6 + 2] cycloadditions

Rhodium complexes plays a pivotal role in organic synthesis as major catalytic contributors to C-C bond formation reactions.35 A number of carbo-/heterocyclic systems have been generated employing rhodium complexes as efficient catalyst in a variety of cycloaddition/cyclization reaction36 Literature rationale suggests an inspiring advancement in this field and many research groups are extensively using rhodium complexes to efficiently conceive these reactions.15 Considering the importance of rhodium catalysed cycloadditions, there should have been a number of reports on generation of eight membered carbo-/heterocycles engaging rhodium complexes in [6 + 2] cycloadditions, however only a few reports are available which successfully comprehends [6 + 2] cycloadditions. Till now, different substrates such as cycloheptatriene, 2-vinylcyclobutanones, allenals, allenyleyclobutanes, cyclooctatrienes etc. have been reported as 6x-components in Rh(1) catalysed [6 +

Department of Chemistry, Gura Nanak Dev University, Amritsar 143065, India Department of Chemical Sciences, IKG Punjab Technical University, Kapurthala, 144603, Punjab, India. E-mail: gaurav@pru.ac.la; gauravorganic@gmail.com

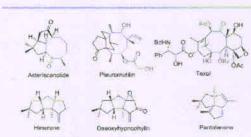
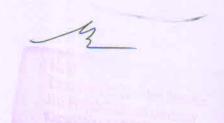


Fig. 1 Carbocycles in natural products.

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Department of Chemistry, Khalsa College, Amritsar, 143005, Punjab, India. E-mail: aa. wix@yahoo.com

Journal of Materials Chemistry B



PAPER

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Perylene diimide—Cu²⁺ based fluorescent nanoparticles for the detection of spermine in clinical and food samples: a step toward the development of a diagnostic kit as a POCT tool for spermine†

Kapil Kumar,^a Sandeep Kaur,^b Satwinderjeet Kaur,^b Gaurav Bhargava,^c Subodh Kumar [®] and Prabhpreet Singh [®] *

The sustainable development of point-of-care testing (POCT) for spermine detection is important to check for food spoilage, early diagnosis of various malignancies and diminished anticonvulsant drug carbamazepine response in chronic epilepsy. Herein, the synthesis, characterization and spectroscopic properties of perytene dimide EA-PDI \(\text{Cu}^{2-\text{s}}\) complex based nanoparticles towards spermine were studied in detail. This EA-PDI \(\text{Cu}^{2+\text{c}}\) complex can be used for the ultrasensitive detection of spermine as low as 86.3 nM (UV-vis) and 90 pM (fluorescence) in aqueous medium, in urine and blood serum samples (recovery 99 \(\pm \) 3) and in the solid state (0.1 \(\mu \) g L⁻¹), and EA-PDI shows minimal cytotoxicity to cells and can easity enter into Human Osteosarcoma MG-63 cells for bio-imaging of Cu²⁺ and spermine. This EA-PDI \(\text{Cu}^{2+\text{c}}\) complex can be established as a cost-effective method to develop a diagnostic kit for POCT of spermine in terms of a solution-based test kit for real time detection of spermine in vapor and solution form released from fermented food samples.

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rsc.li/materials-b

Introduction

Supramolecular synthetic receptors having molecular biorecognition properties are important in biology for monitoring enzymatic activity, signaling and regulation. ^{1,2} In this context, perylenediimide (PDI), ^{3,4} which was first synthesized in 1913 by Kardos² as an industrial colorant, ^{1s} now-a-days attracting increasing research attention as a key chromophore for applications in the field of fluorescent sensors and bio-imaging. ⁶⁻¹² besides its application in organic electronic devices. ^{13,14} The wide spread applications of PDIs may be attributed to their strong absorption and fluorescence properties, and chemical, thermal and photochemical stability. One of the main obstacles in exploring PDIs for sensing applications seems to be poor solubility, however their optical properties and solubility can be readily tuned through functionalization at the bay-, N-terminal-

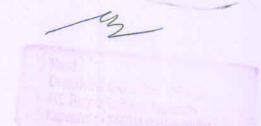
ortho- and peri-positions. $^{3.4.15}$ To circumvent the solubility issues, ionic PDIs were studied as fluorescent sensors, whereas to achieve sensing properties using neutral PDI derivatives in >50% water is still a challenging task. $^{6-12}$

Biogenic polyamines such as putrescine, spermidine and spermine, essentially found in all eukaryotic cells, play many crucial roles in cellular growth and proliferation, transcription, translation and modulation of gene expression. 16,17 Spermine detection in biofluids such as urine and blood serum is clinically used for diagnosis of early stages of various types of malignancies, to determine the efficiency of long term cancer chemotherapy and to check the development of anticonvulsant carbamazepine (CBZ)-resistant Na⁺ currents in chronic epilepsy. 18,19 Biogenic polyamines have also been produced as a physiological component in many protein rich food products by microbial proteolysis followed by decarboxylation of amino acids. 20,21 They can lead to production of nitrosoamines in meat products by reacting with nitrites22 and high levels of biogenic amines in spoiled food can cause headaches, nausea and changes in blood pressure.21 Therefore, the detection of spermine has become highly paramount as it is directly linked with food safety and human health

Now-a-days, point of care testing (POCT), which involves diagnostic testing at the time and place of patient care, is in

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[&]quot;Department of Chemistry, UGC Centre of Advanced Studies - H. Own Nanak Dev University. Amritsar 143 005, India. E-mail: prahhpreet.chera@gudu.ac.in

Department of Botonical and Environmental Sciences, Guru Nanak Dev University, Amritsar, India

Department of Chemical Sciences, IK Gujrai Punjab Technical University, Kapurthala 144601, India

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Near-IR discriminative detection of H₂S and Cysteine with 7-nitro-2,1,3benzoxadiazole-perylenediimide conjugate in water, live cells and solid state: Mimicking IMP, INH and NOR/OR complimentary logic

Poonam Sharma, * Kapil Kumar, * Sandeep Kaur, * Satwinderjeet Kaur, * Gaurav Bhargava, * Subodh Kumar * and Prabhpreet Singh, ***

*Department of Chemistry, UGC Centre for Advanced Studies, Guru Nanak Dev University, Amritsar 143 905, India.

e-mail: prabhpreet.chem@gndu.ac.in; Tel; +91-84271-01534

Department of Environmental and Botanical Sciences, Guru Nanak Dev University, Armitsar 143 005, India.

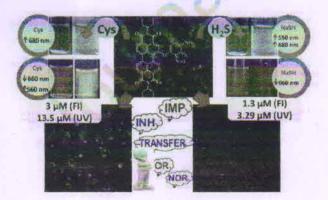
Department of Chemical Sciences, IK Gujral Punjab Technical University, Kapurthala-144601, Punjab, India.

*Corresponding author. Tel: +91-84271-01534

Email: prabhpreet.chem@gndu.ac.in:

Graphical abstract

Near-IR ratiometric chromo-fluorescent probe for detection of H₂S and Cys was developed based on differential optical response in water and live cells.



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A multifunctional perylenediimide-based dual-analyte chemodosimeter for specific and rapid detection of H₂S and Pd⁰ in water, biofluids, live cells and solid state

Kapil Kumar^a, Sandeep Kaur^b, Satwinderjeet Kaur^b, Gaurav Bhargava^c, Subodh Kumar^a, Prabhpreet Singh^{a*}

*Department of Chemistry, UGC Centre for Advanced Studies, Guru Nanak Dev University, Amritsar 143005 (India) e-mail: prabhpreet.chem@gudu.ac.in

Department of Botanical and Environmental Sciences, Guru Nanak Dev University, Amritsar, India.

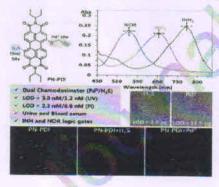
Department of Chemical Sciences, IK Gujral Panjab Technical University, Kapurthala-144601, India

*Corresponding author. Tel: +91-84271-01534

Email: prabhpreet.chem@gndu.ac.in;

Graphical abstract

A fast-responsive, near-IR colorimetric fluorescent chemodosimeter for individual and simultaneous detection of H₂S and Pd⁰ at nanomolar level in water, biofluids and A549 cells is reported.



Highlights

 A novel perylenediimide-based dual-analyte chemodosimeter PN-PDI have been synthesized.

- 3





Rhodium-catalysed chemo- and regio-selective [3+2+2] cycloadditions of bis(methylenecyclopropanes) and alkynes: Synthesis of spirocyclic 5–7 condensed cycloheptenes

Bilash Kuila^a, Priyanka Sharma^a, Dinesh Mahajan^b, Prabhpreet Singh^c, and Gaurav Bhargava^a

^aDepartment of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, India; ^bDrug Discovery Research Center, Translational Health Science and Technology Institute, Faridabad, India; ^cDepartment of Chemistry, Guru Nanak Dev University, Amritsar, India

ABSTRACT Rhodium-catalyzed intermolecular [3+2+2] cycloaddition reactions of bis(methylenecyclopropanes) with different alkynes are described. The rhodium catalyzed [3+2+2] cycloadditions resulted in the formation of functionalized 5-7-3 spirocyclic carbocycles in moderate yields with excellent regio- and chemo-selectivity. GRAPHICAL ABSTRACT 10 mark (RhiCOD)O(L) 25md % P(OPh)₃ Toksene RO₂C [3+2+2] cycloaddition ta: R = Me 3a; R. R. = Me, R2 = H 2a: R .= Mo.R .= H 2b; R₁= Et R₂ = Me 2c; R₁, R₂ = Et 2d; R₁= Et R₂ = Ph 16: R = Et 3b; R= Me; R = Et R = Me 3c; R= Me, R₁, R₂ = Ét 3d; R= Me, R₁= Ét R₂ = 2e: Phenyt acetyler 2f: DMAD 2g: Butyne⁻¹ 4-dist of; R= Et. R₁= Me.R₂ = H 3f, R= 'Pr, R₁= Me.R₂ = H 2h Propargyl OTBS

ARTICLE HISTORY Received 10 October 2019

KEYWORDS

[3 + 2+2] Cycloadditions; [Rh(COD)CI]₂; bis(methylenecyclopropanes); multicomponent reaction; spirocyclic cycloheptenes

Introduction

Transition metal-catalyzed cycloaddition reactions are one of the few sustainable methods which lead to synthesis of complex molecules with high atom economy exploiting simple synthetic precursors. The two-component cycloadditions such as [4+2] or [3+2] transformations are synthetically most explored reactions for the synthesis of six and five-membered carbo-/heterocycles. Contrary to this, synthesis of higher-membered carbo-/heterocycles using [m+n+o] cycloadditions still is infancy and needs the attention of synthetic chemists. The [3+2+2] cycloadditions of functionalized methylenecyclopropanes and alkenes and/or alkynes provide an easy access to functionalized cycloheptanes and their analogs. The procedure of the synthesis of the synthesis of higher-membered carbo-/heterocycles using [m+n+o] cycloadditions of functionalized methylenecyclopropanes and alkenes and/or alkynes provide an easy access to functionalized cycloheptanes and their analogs.

CONTACT Gaurav Bhargava gauravorganic@gmail.com; gaurav@ptu.ac.in Department of Chemical Sciences, I, K Gujral Punjab Technical University, Kapurthala, Punjab 144603, India.

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Journal of Materials Chemistry B



PAPER

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Self-assembled nanofibers of perylene diimide for the detection of hypochlorite in water, bio-fluids and solid-state: exogenous and endogenous bioimaging of hypochlorite in cells†

Kapil Kumar,^a Sandeep Kaur,^b Satwinderjeet Kaur,^b Gaurav Bhargava,^c Subodh Kumar (p^a) and Prabhpreet Singh (p) *^a

A fluorescent probe PDI-DAMN based on perylenediimide containing diaminomaleonitrile at the bay-position was designed and synthesized for the detection of CIO*. PDI-DAMN self-assembled as nanofibers with diameters in the range of 100-200 nm in H2O CH3CN (1:1). The addition of CIO into PDI-DAMN resulted in the disintegration of nanofibers into flake-like aggregates of smaller size (50-80 nm) as supported by SEM and DLS data. The addition of CtOT to HEPES buffer-CH₃CN solution (1.1. v/v, pH 7.4) of PDI-DAMN caused a hypochromic effect on the ICT band at 528 nm and turn-on fluorescence enhancement at 508/554 nm due to the oxidative cleavage of -C=N- bond. A linear correlation plot between the concentration of CIO versus fluorescence intensity (R2 = 0.9968)/ absorbance ($R^2 = 0.9988$) in the concentration range 0-7 nM (fluorescence)/0-90 nM (absorbance) could determine CIO" with the detection limits of 1 and 10 nM, respectively. Optical studies performed on spiked urine and blood serum samples showed good estimation and recovery of CIO+ (100 ± 5%). TLC-based test-strips coated with PDI-DAMN changed colour upon the addition of CtO- with detection as low as 7.44 ng cm⁻². The application of PDI-DAMN for the bio-imaging of both exegenous and endogenous CIO⁻ in MG-63 cells with good biocompatibility has also been demonstrated. The detailed mechanism of the interactions of CIO" with PDI-DAMN using ¹H NMR titration, DFT studies and response mechanism of pH are also discussed.

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1. Introduction

Hypochlorous (HOCl) or hypochlorite (ClO⁻; conjugate base), besides being a bioactive molecule, is an important reactive oxygen species (ROS), which can be endogenously generated by the reaction of H₂O₂ and chloride ion catalyzed by myeloperoxidase (MPO) enzyme in leukocytes.³ Hypochlorous acid helps the human immune system by killing the invading bacteria and pathogens.³ At physiological pH, HOCl could partially dissociate to hypochlorite (ClO⁻) in water and can be used as a bleaching agent and as a disinfectant for drinking water.³ The uncontrolled level of hypochlorous acid (oxidative agent) can lead to the oxidation of proteins, cholesterol, DNA and RNA in living

cells, resulting in pathological diseases such as atherosclerosis, cardiovascular disease, osteoarthritis, cystic fibrosis cancers and pulmonary lesions. Therefore, real-time direct visualization or in situ cellular imaging of ClO in a living system is highly desirable to better understand its role in the biological system.

Many detection techniques such as electrochemical, electron paramagnetic resonance (EPR),5 chromatographic methods,6 and chemiluminescence,7 are available in the literature. However, fluorescence and colorimetry-based methods offer several advantages in comparison to other analytical techniques such as non-destructive nature, sensitivity, selectivity, chemical modification and spatiotemporal resolution in living cells. In recent years, many fluorescent probes have been designed for the detection of CIO with an underlying mechanism involving the oxidation of oxime, a acyl nitroso group, p-methoxyphenol, 10 selenide, 11 arylboronate, 12 thioester/thioether 13 and -C=N- bond.14 In this context, fluorescent probes based on a red-to-NIR dye such as cyanine,15 rhodamine,16 BODIPY,17 porphyrin¹⁸ have been reported. However, their use is often limited due to fast photobleaching, low quantum yield, interference from other ROS, incompetency in detecting of CIO in

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Department of Chanistry, UGC Centre for Advanced Studies-II, Guru Nanak Dev University, American, (Fb) 143 005, India. E-mail: prabhpreet.chem@gndu.ac.in; Tel: 491-84221-01534

b Department of Botanical and Environmental Sciences, Guru Nanak Dev University, Amritsur, India

Department of Chemical Sciences, IK Gujrul Punjab Technical University, Kapurthala 144601, Punjab, India

[†] Electronic supplementary information (ESI) available. See DOI: 10.1039/e9tb01902b

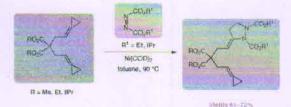
Nickel(0)-Catalyzed [3+2] Cycloadditions of Bis(alkylidenecyclopropanes) with Diazenes: A Facile Synthesis of Functionalized Pyrazolidine-1,2-dicarboxylates

Bliash Kulla Ravees Natkoon Dinesh Mahalan^b Prabhpreet Singh Gaurav Bhargava*

[®] Department of Chemical Sciences, J. K. Cujtal Punjab Technical University, Kapurthalia, Punjab-144603, India дашаv≜phuac.in

gate averprise...n Drag Discovery Research Centre (DDRC), Translational Health Sciences and Technology Institute (THSTI), Faridabad-121001, india

ent of Chemistry, Guru Nanak Dev University, Amritsar,



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Abstract A nickei(0)-catalyzed intermolecular [3+2] cycloaddition of bis(alkylidenecyclopropanes) with diazenes such as diethyl or diisopropyl azodicarboxylate gave pyrazolidine-1,2 dicarboxylates in moderate to good yields (61-72%).

Keywords bisalkylidenecyclopropanes, dicarboxylates, [3+2] cycloaddition, nickel catalysis

Metal-catalyzed [m+n] cycloaddition reactions are effective tools for the synthesis of carbo- and heterocyclic systems. Functionalized cyclopropanes have been exploited in metal-catalyzed [m+n] cycloadditions for the synthesis of monocyclic or condensed carbocycles.2-4 There are also reports on cycloadditions of activated cyclopropanes with heterodienophiles for the synthesis of monocyclic or condensed carbo- and heterocycles.2-4 De Meijere and coworkers explored the Lewis acid-catalyzed [3+2] cycloadditions of 2-arylcyclopropane-1,1-dicarboxylates with diazenes to afford functionalized pyrazolidines.5 However, [m+n] cycloadditions of functionalized nonactivated cyclopropanes, especially with heterodienophiles, have rarely been reported in the literature.20,4,6,3

Pyrazolidines have been evaluated as antibacterial, antifungal, anticancer, antidepressant, antiinflammatory, antituberculosis, antioxidant, and antiviral agents in various pharmacological studies.8 Several pyrazolidine-based drugs have been marketed, including the antiinflammatory drug celecoxib, rimonabant for the treatment of obesity, fomepizole as an effective alcohol dehydrogenase inhibitor, and sildenafil as a phosphodiesterase inhibitor.9 Pyrazolidines are also useful as chiral auxiliaries and as synthetic re-

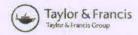
agents in multicomponent reactions.10 In addition, natural products containing pyrazolidine moieties have been shown to have pharmacological properties.31

Conventional approaches, such as the condensation of 1,3-dicarbonyl compounds with hydrazines or [3+2] cycloadditions of 1,3-dipoles have been used in syntheses of simple pyrazolines.12 However, there are few reports on synthesis of functionalized pyrazolidines. Chaudhry et al. recently reported acid-catalyzed cyclizations using allylic hydrazines for the synthesis of pyrazolidines.13

In view of these results and our ongoing interest in the cycloaddition chemistry of functionalized cyclopropanes, we wish to report an extension of our nickel(0)-catalyzed [3+2]-cycloadditions of bis(alkylidenecyclopropanes) to the preparation of pyrazolidines by using diazenes (Scheme 1).7

Scheme 1 [3+2] Cycloadditions of bis(alkylidenecyclopropanes)

In the present work, we examined the [3+2] cycloadditions of bis(alkylidenecyclopropanes) with diazenes such as diethyl azodicarboxylate (DEAD) or diisopropyl azodicarboxylate (DIAD) as dienophiles (Scheme 1). The reaction



SYNTHETIC COMMUNICATIONS REVIEWS

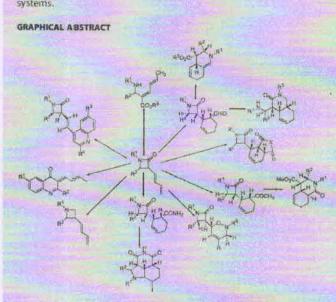
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3-Butadienyl- β -lactams: A useful synthon for functionalized heterocycles

Gaurav Bhargava, Maninderjeet K. Mann, and Rayees Ahmad Naikoo Department of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, India

ABSTRACT

3-Butadienyl-\(\beta\)-lactams are useful organic synthon which have efficiently been explored for the synthesis of a variety of heterocyclic systems. The present review compiles the various reports on the synthesis of 3-butadienyl-\(\beta\)-lactams and their explorations in the development of variety of monocyclic as well as condensed heterocyclic systems.



ARTICLE HISTORY Received 6 July 2020

KEYWORDS

amidic bond hydrolysis; Dienyl- β — Lactam; cycload-ditions; β -lactams, β -lactam synthon methodology

Introduction

 β -Lactams are versatile synthons in organic synthesis. [1-5] Functionalized lactams have also been explored as important pharmacophore having plethora of biological applications [1-10] β -lactam based antibiotics, such aspenicillin, cephalosporins, carbapenems, nocardicins, monobactams, clavulanic acid are used therapeutically for controlling

CONTACT Gaurav Bhargava agauravorganic@gmail.com, gaurav@ptu.ac.in Department of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, Punjab, 144603, India.

Dedicated to Professor M. P. Mahajan for his work on 3-butadienyl-\(\beta\)-lactams

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7-Endo-trig Pictet-Spengler type cyclization of 5-alkylidene/arylidene-amino-3H-pyrimidin-4-ones: An efficient and diastereoselective synthesis of pyrimido[4,5-b] [1,4]benzodiazepines

Rayees Ahmad Naikoo^a, Rupesh Kumar^a , Parvesh Singh^b, and Gaurav Bhargava^a

^aDepartment of chemical sciences, I. K. Gujral Punjab Technical University Kapurthala, Punjab, India,

bSchool of Chemistry and Physics, University of KwaZulu Natal, Durban, South Africa

ABSTRACT

The manuscript describes an efficient, atom economical synthesis of pyrimido(4,5-b)[1,4]benzodiazepin-4-ones by relatively unexplored 7-endo-trig Pictet-Spengler type cyclisations. The synthetic methdodlogy involves the synthesis of different variants of 5-arylidene-amino-3H-pyrimidines and their p-toluene sulfonic acid mediated 7-endo-trig Pictet-Spengler type cyclisations to afford biologically relavent functionalized benzodiazepine condensed pyrimidinones such as pyrimido[4,5-b)[1,4]benzodiazepines in good to excellent yields (82–94%),

GRAPHICAL ABSTRACT

ARTICLE HISTORY

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KEYWORDS

Pictet—Spengler; 7-endo-trig cyclization; benzodiazepine; pyrimidinones; benzodiazepines condensed pyrimidinones

Introduction

Pictet-Spengler cyclizations are among the useful synthetic methodologies for the synthesis of a variety of heterocycles. [1] There are ample reports on the synthesis of biologically relevant heterocycles using 5/6-endo-dig Pictet-Spengler cyclisations. [1] However, the reports on the synthesis of corresponding seven-membered heterocycles via 7-endo-trig Pictet-Spengler cyclizations are still scarce in the literature. [2] Hence, exploration on atom economical Pictet-Spengler cyclizations using appropriately

CONTACT Gaurav Bhargava agaurav@ptu.ac.in, gauravorganic@gmail.com Department of Chemical Sciences, I. K. Gujral Punjab Technical University Kapurthala, Punjab 144603, India.

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Recent development in synthesis of pyrrolin-4-ones/pyrrolin-3-ones

Priyanka Sharma, Rupesh Kumar, GauravBhargava**

^aDepartment of Chemical Sciences, I.K.Gujral Punjab Technical University, Kapurthala-144603, Punjab, India

E-mail: gauravorganic@gmail.com, gaurav@ptu.ac.in

Abstract: Pyrrolin-4-ones/Pyrrolin-3-ones and its derivatives are important heterocyclic systems which are observed in vast variety of natural products, pharmaceuticals and biologically important compounds. Different researchers all across the world have developed different synthetic methodologies for the construction of functionalized pyrrolin-4-ones/Pyrrolin-3-one scaffolds such as the transition-metal catalyzed/mediated cycloisomerizations of 1-amino ynones and dimerization of enaminones or σ -diazo- β -oxoamidesetc. The present review article summarizes the various reports on the synthesis of various simple and functionalized pyrrolin-4-ones/pyrrolin-3-ones from 2000 onwards.

Keywords: Pyrrolin-4-ones, Pyrrolin-3-ones, Enaminones, Ynones, Diketones, Acetylenic esters, Diazadienes, Formyl-2*H*-azirines.

1. INTRODUCTION

Pyrrolin-4-ones/Pyrrol-3-ones are prevalent in a vast variety of pharmaceutical active molecules, biologically important natural products, and a variety of vital materials having immense applications. [1-3] Functionalized pyrrolin-3/4-oneshave received incredible attention, as privileged N-heterocycles, in the development of various molecules having anticancer, antithrombotic, and antimalarial activities. [4-5] The pyrrolin-4-ones are also valuable intermediates in the development of functionalized pyrrolidines and other natural products. Pyrrol-3/4-ones occur extensively in natural products as fundamental units with useful biological activities (Figure-1). [6-8]

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SYNTHETIC COMMUNICATIONS REVIEWS

Recent developments in the synthesis of tricyclic condensed pyrimidinones

Rayees Ahmad Naikoo^a, Rupesh Kumar^a @, Vipan Kumar^b, and Gaurav Bhargava^a

^aDepartment of Chemical Sciences, IKG Punjab Technical University, Kapurthala, Punjab, India; ^bDepartment of Chemistry, Guru Nanak Dev University, Amritsar, Punjab, India

Functionalized tricyclicpyrimidinones, a class of condensed heterocyclic systems, are found in numerouspotentially active molecules having both innate as well as synthetic origin. Different researchers across the world have explored different synthetic methods for the construction of appropriately functionalized tricyclicpyrimidinones. The present review article recapitulates various reports pertaining to substituted and functionalized tricyclic pyrimidinones reported since 2000. GRAPHICAL ABSTRACT ORDINATION OF THE PROPRIES OF THE PR

ARTICLE HISTORY Received 7 January 2021

KEYWORDSCondensed pyrimidinones; fused pyrimidinones; tricyclic pyrimidinones

Introduction

Pyrimidinones are important heterocyclic compounds which are predominant innumerous biologically active molecules. Functionalized pyrimidinones have been employed as a useful template in the exploration of therapeutically active molecules. The synthetic explorations for devising new routes for the formation of condensed pyrimidinones have been made comprehensively due to their wide spectrum of biological activity. Pyrimidinones and its derivatives are prevalent in DNA, RNA, vitamins, coenzymes and

CONTACT Gaurav Bhargava gauravorganic@gmail.com, gaurav@ptu.ac.in Department of Chemical Sciences, IKG Punjab Technical University, Kapurthala, Punjab, India.

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BRIEF REPORT



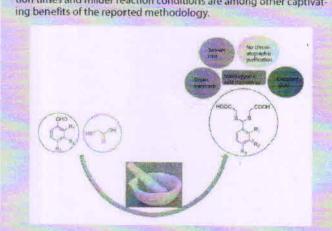
Solvent-free mechanochemical synthesis of bisthioglycolic acid derivatives: an efficient and versatile strategy for carbon-sulfur bond formation

Rayees Ahmad Naikoo^a, Parvesh Singh^b, Rupesh Kumar ^{@a} and Gaurav Bhargava^a

^aDepartment of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, India; ^bSchool of Chemistry and Physics, University of KwaZulu Natal, Durban, South Africa

ABSTRACT

An eco-friendly, easily achievable and efficient strategy has been explored to synthesize functionalized bisthioglycolic acids using a variety of aldehydes and thioglycolic acid. The employed protocol is solvent free and provides the desirable products in excellent yields (90–99%) with atom economy. Besides, cost effectiveness, short reaction times and milder reaction conditions are among other captivating benefits of the reported methodology.



ARTICLE HISTORY

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KEYWORDS

bisthioglycolic acid; mechanochemical synthesis; carbon–sulfur bond; formation; solvent-free synthesis; Green synthesis

Introduction

The exploration of atom economical greener protocols with easily accessible final products without any additional extraction or chromatographic procedures is among the significant aims in organic chemistry. Solvent-free mechanochemical strategies for synthetic transformations have created a niche at the forefront of sustainable green chemistry. These strategic

CONTACT Gaurav Bhargava gaurav@ptu.ac.in; gauravorganic@gmail.com; Rupesh Kumar rupesh.manak@gmail.com Department of chemical sciences, I. K. Gujral Punjab Technical University, Kapurthala, Punjab 144 603, India

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Oxa-Michael Addition Reactions of 3-hydroxy-2-azetidinones: Synthesis of 1, 3, 4-Trisubstituted-2-Azetidinones

Priyanka Sharma¹, Maninderjeet Kaur Mann¹, and Gaurav Bhargava^{*1}

Department of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, Punjab. 144 603, India

ARTICLE HISTORY

Received: June 28, 2020 Revised: January 16, 2021 Accepted: January 24, 2021

DOM: 10.2174/15701786186662/0610155744 Abstract: The manuscript describes a facile and an efficient methodology for the synthesis of 1, 3, 4 trisubstituted- β -lactams by base mediated oxa-Michael addition reactions of 3-hydroxy-2-azetidinones with acetylenic esters under different reaction conditions. These functionalized 1, 3, 4-trisubstituted-azetidin-2-ones are useful organic synthens for the synthesis of various heterocyclic compounds having diverse pharmacological applications.

Keywords: Lactams. Oxa-Michael reaction, 3-Hydroxy lactams, Acetylenic ester.

1. INTRODUCTION

The reactions of Michael addition and its modified forms are versatile methods in organic synthesis [1-3]. Michael addition reactions offer tremendous synthetic potential since the products available are valuable intermediates in organic synthesis [4-10]. Extensive progress has been made in this field over the last decades. However, hetero-Michael addition, such as aza- Michael [11, 12], sulfa-Michael [13, 14] phospha-Michael [15] and oxa-Michael reactions have gained considerably less attention [16, 17]. Oxa-Michael reactions are efficient tools, which afford a variety of pharmacologically important molecules, such as ten hydropyrans, chromenes, xanthones, etc. [18]. Functionalized 1, 3, 4-trisubstituted-azetidin-2-ones have been evaluated for their authorities and the properties. antibacterial, anti-tubercular, anti-hyperglycemic, anticancer, anti-HIV, anti-parkinson [19-21]. Antidepressant and anti-elastase activity and cholesterol absorption inhibitors, etc. [22, 23]. As a part of our ongoing interest in heterocyclic synthesis, we have earlier reported the synthesis of functionalized 1, 3, 4-trisubstituted azetidinones and their transformations to a variety of heterocyclic systems [24-26], such as proline esters [27, 28], benzodiazepinones and thiazolidinones [29], and pyrrolidines [30]. In continuation of this interest, we have recently studied aza-Michael addition reactions of 3-amino-2-azetidinones with different acetylenic esters for the formation of 4-oxo-dihydro-1H-pyrroles via tandem aza-Michael additions and intramolecular amidic ring-opening reactions (Fig. 1) [31]. Since Michael additions of functionalized azeticlin-2-ones, such as cis-3-hydroxyazetidin-2-ones [32] with different Michael acceptors, can provide easy access to 1, 3, 4-trisubstituted-2-azetidinones, hence, if was thought worthwhile to explore the oxa-Michael addition reactions of cis-3-hydroxy-1-aryl-4-styrylazetidin-2-ones with a variety of acetylenic esters. The present manuscript describes a simple methodology for the preparation of functionalized 1, 3, 4 trisubstituted azetidin-2-ones via oxa-Michael addition reactions of functionalized 3-hydroxy-azetidin-2-ones with acetylenic esters.

2. RESULTS AND DISCUSSION

The starting cis-3-hydroxy-2-azetidinones were synthesized by [2+2] cycloaddition reactions of functionalized 1-azadienes with acetoxyacetic acid in the presence of triethylamine and p-TsCl (Scheme 1) according to the

Reported procedure [33-35], followed by the hydrolysis of cts-3-acetoxy-2-azetidinones using lithium hydroxide in THF/water (Scheme 1) [35]. Then, oxa-Michael addition reactions of the cis-3-hydroxy-2-azetidinones (1a-c) were explored for the synthesis of functionalized 1, 3, 4-trisubstituted-2-azetidinones. Initially, oxa-Michael addition reactions with acetylenic esters, such as DMAD and methyl propiolate, were optimized for best yields and selectivity. The results are presented in Table 1.

The reactions of 1a-c with acetylenic esters 2 resulted in the formation of oxa-Michael adducts 3, and the corresponding 4-oxo-5-(3-aryl-1-arylaminoally1)-4. 5-dihydro-1H-furan-2, 3-dicarboxylate esters (4) were not observed (Scheme 2) The reaction of 1b with 2a gave better yields of oxa-Michael adducts using NaH as a base. The reaction did not occur in the absence of any base or by using weak bases, such as K₂CO₃, Na₂CO₃, etc. (Table 1, Entries 5-14). The

*Address correspondence to this author at the Department of Chemical Sciences, I. K. Gujral Punjab Technical University, Kapurthala, Punjab. 144 603, India; E-mail: gaurav@ptu.ac.m

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Regioselective Copper(I)-Catalyzed Ullmann Amination of Halopyridyl Carboxylates using Sodium Azide: A Route for Aminopyridyl Carboxylates and their Transformation to Pyrido[2, 3-d]pyrimidin-4(1H)-ones

Nisha, *1st Mr. Chetan Sharma, the Rupesh Kumar, and Yogesh Kumar*16, cl

Destroted to Dr. Couray Bhargova on the accasion of his 40th birthday

We report began an efficient, straightforward and ligand free synthesis of amicopyridy carboxylates, an important building black used in pharmaceuticals and agrochemicals. The Cisp' N band formation utilize a readily available Cu catalyst, NaN, as the ansaro source in ethanol and the corresponding onhoturnitionalized atomatic arrives were synthesized in good to excellent neids. This ligand free one-pot domino methodology counts through Ullmann-type coupling of halopyridyl carpornistes with sodium ande followed by reduction with stratial. These tunchonalized ammopyridyl carboxylates provides an easy access to biologically potent pyridol2,3-til promouse algebrane hybrids.

Designance of narrogen containing heterocyclic compounds in synthetic intermediates, natural products, pharmaceutical and accordenace agents and functional materials has provoked synthetic themists to develop expedient and mild amination extens. Over the last decades, transition metal catalysts provides venicate strategies to construct Cisp²). NH, bond and present a great opening to derivative uncommon chemicals with modest functionality to synthetically novel compounds to st in this context, several catalytic methods have been used, particular, the palladium and copper catalyzed amination of ey stades has pained increasing interest the Since all these med with involve the use of prefunctionalized arene starting recent efforts are focused on the count of new mithids by the use of parent arenes as the starting materials. Attracted the waisting methods are quite useful," however,

developing a direct route to achieve amination of o-functional ized haloheteroarenes with structural diversity from the readily accessible simple substrates would be enthralling while exigent at the same time.

On the other hand, aminopyridyl carboxylates are highly useful and valuable compounds that have numerous applications." 2-aminonicotinic acid (vitamin 83) and 2-aminonicotinamide moieties appear frequently in medicinal chemistry programs and, are valuable intermediates for the grounding of compounds with prospective as chemotherapeutic agents is a In view of the advances made in the formation of CC, CO, CN. and C-halogen bonds by modern ortho-CH functionalizations, a similar approach would be interesting also for the regionaler tive synthesis of aminoheteroarenes.

Based on our previous efforts on developing the synthetic methods for heterocycles, to in herein we report a novel ligand free one-pot copper-catalyzed direct amination of halopyridyl carboxylates using sodium azide in the presence of etharwil at moderate remperature. This newly exposed reaction is simple. straightforward, utilizing inexpensive copper catalyst and converts the readily available substrates i.e. 2-halomorphic acid, 2-halonicotinamide into important aminoheteroarenes vio a ligarid free one-pot-Ullmann type coupling with socium acide followed by reduction. These aminoheteroarenes are novel synthons for the synthesis of highly functionalized pyrido[2, 3. dipyrimidin-4(1H) ones.

Initially, to find an optimized reaction condition for the synthesis of 2 aminonicotinic acid 2a 2 chibronicotinic acid was chosen as the model substrate including the catalysts, bases, and solvents under nitrogen atmosphere as shown in Table I We started our experiment with 2 chloronications acid La and NaN₁ (4 mmol) in the presence of Cu8r (10 mol %) and

Scheme 1. Copper Catalyzed Amination of 2-Chlusopicetime acid with Nan-

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Paper

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Concentrated solar radiation promoted unconventional greener approach: solvent-free benign synthesis of functionalized benzimidazoles

Simran Harsh, Mohamad Yusuf, Rohit Sharma, Yogesh Kumar, and Rupesh Kumar*

Department of Chemical Sciences, I. K. G. Punjab Technical University, Kapurthala, Punjab, India-144603

Department of Chemistry, Punjabi University, Patiala, Punjab, India

Department of Medicinal Chemistry, National Institute of Pharmaceutical Education and Research, Mohali
Punjab, India

*UNAM-National Nanotechnology Research Center, Institute of Materials Science and Nanotechnology,

Department of Chemistry, Bilkent University, 06800 Ankara, Turkey

E-mail: rupesh.manak@gmail.com

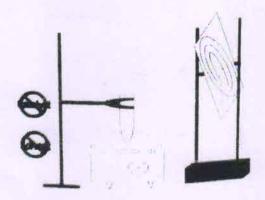
eceived 07-10-2018

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bstract

enewable concentrated solar-radiation (CSR) offered a promising en route for the development of practic ighly efficient, scalable, catalyst free and solvent-free clean process leading to the synthesis of functionaliz enzimidazoles. Developed protocol has a very good substrate scope, involves mild reaction conditions a roducts obtained in good to excellent yields. Method presented the observations in which light alone confect the remarkable changes with more than 85% energy saving and 75% less reaction time in listed organizations.



eywords: Concentrated solar radiation, photocatalysis, solar synthesis, C-N bond formation, benzimidazole

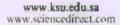
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King Saud University

Arabian Journal of Chemistry





Chlorophyll triggered one-pot synthesis of 3,4-dihydropyrimidin-2(1H)-ones via photo induced electron transfer reaction



Simran Harsh a, Sanjay Kumar b, Rohit Sharma c, Yogesh Kumar d, Rupesh Kumar d

Received 19 July 2019; accepted 10 November 2019 Available online 22 November 2019

KEYWORDS

Chlorophyll 3.4-dihydropyrimidin-2(1H)-Photochemical reaction. Biginelli reaction; Concentrated solar radia-Photo-induced electron transfer (PET)

Abstract The photocatalytic potential of chlorophyll has been investigated for the facile synthesis of dihydropyrimidinones utilizing concentrated solar irradiation towards sustainable energy solutions. This, one-pot, multicomponent Biginelli reaction, which involves a photoinduced electron transfer (PET) mechanism, affords a green and efficient approach for the transformation of the commercial aldehydes, \(\beta\)-keto ester and urea into valuable 3,4-dihydropyrimidin-2(1H)-ones with wide substrate scope and diversity. These improved reaction conditions allow the formation of a variety of substituted dihydropyrimidinones with high yields and purity in a short duration of time

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1. Introduction

(Sarrophyll is a complex, natural and abundant organic pigment that sets as photocatalyst in green plants. It behaves as the proceptal photoreceptor in the green plants that transfers

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energy to the reaction centres to form bigger sugar molecules from simple carbon dioxide and water through the initiation of photoinduced electron transfer (PET) (Bassham and 1950). Illuminating photochemical synthesis is greatly motivated by photosynthesis, one of the most captivating natural phenomena (Gust et al., 2009; Kim et al., 2015). The perfect harvesting of the solar light and photoelectron transfer through a series of steps to the redox centres is the theme of the photosynthesis (Depenheter and Michel 1989, Flemmy et al., 1985). limitating of the photosynthesis in the simulating environment is the key idea behind the use of solar energy for chemical research in the photocatalysis (Shaw et al. 3016 Xer and Sanlu(1) 2018. Priori et al., 2018) Some of the organic chemists used photoredox catalysis in visible light for organic

11-11. Assistant Paralistical by Chevner B.V. on Johall of King Sand University.

Department of Chemical Sciences, I. K. G. Punjab Technical University, Kapurthala, Punjab 144603, India

Department of Chemistry, M.M. Modi College, Patiala, Punjab, India

^{*} College of Horticulture and Forestry, Dr. Y.S. Parmar University of Horticulture and Forestry, Neri, Hamirpur, HP, India

Department of Chemistry, Durham University, Durham, UK

REVIEW



Recent development in the synthesis of pyrrolin-4-ones/ **Pyrrolin-3-ones**

Priyanka Sharma | Rupesh Kumar | Gaurav Bhargava

Department of Chemical Sciences, I K Claimst Founds to bankal University Adjournals, India

Correspondence

Souther Bhargava, Department of Chemical Sciences, LK Guiral Puniab Technical University, Kapurthala, Puniab,

Email gautavorganic@gmail.com. COLL TON SEPTEMBER OF

Abstract

Pyrrolin-4-ones/pyrrolin-3-ones and its derivatives are important heterocyclic systems which are observed in vast variety of natural products, pharmaceuticals, and biologically important compounds. Different researchers all across the world have developed different synthetic methodologies for the construction of functionalized pyrrolin-4-ones/pyrrolin-3-one scaffolds such as the transition-metal catalyzed/mediated cycloisomerizations of 1-aminoynones and dimerization of enaminones or α -diazo- β -oxoamides etc. The present review article summarizes various reports on the synthesis of various simple and functionalized pyrrolin-4-ones/pyrrolin-3-ones from 2000 onward.

INTRODUCTION

Pytrolin-4-ones/pytrol-3-ones are prevalent in a vast variety of pharmaceutical active molecules, biologically important natural products, and a variety of vital matemals having immense applications. 1-3 Functionalized pyrrolin 3 4-ones have received incredible attention, as privileged N-heterocycles, in the development of various molecules having anticancer, antithrombotic, and antimalarial activities.45 The pyrrolin-4-ones are also valuable intermediates in the development of functionalized period dines and other natural products. Pyrrol-3/4-ones occur extensively in natural products as fundamental unit with useful biological activities (Figure 1).6-8

Different methods are reported in the literature Figure 2) for the convenient synthesis of functionalized pyrrolin 4-ones pyrrol 3-ones such as condensation of functionalized amidines and 1.4-diazabutadienes with auctyleric esters and drazo compounds, cyclization reactions of readily available enaminones, and metal-catalyzed intramolecular cyclization of a aminoynones which are discussed later in the different sections of this review Other diverse methods for the synthesis of substituted pyrrolin-4-ones are also discussed thoroughly in the present review which involves dimerization of enamides, oxidative CH/NH cross-coupling of tralkenoyl ketene N,S-acetals, reactions involving formyl-2H-azirines and 3-amino-2-azetidinones highlighting their various important aspects of these reported protocols. As a whole, the present review summarizes all such reports associated with the developments pertaining to the synthesis of functionalized pyrrolin-4-ones/pyrrol-3-ones, since 2000.

CONDENSATION OF \a-AMINO ACID ESTERS WITH ALDEHYDES

Smith III et al have disclosed an effective, three-step iterative solid support-based synthesis of polypyrrolinones for the formation of a wide variety of pyrrolinone scaffolds such as monopyrrolinone (-)6 and bispyrrolinone (-)-10 (Scheme 1).8d The above methodology involves the initial synthesis of Wang resin-bound amino ester 3 using Teoc-protected amino acid (+)-2 via Mitsunobu reaction. Removal of the Teoc-protecting group (TBAF) followed by condensation of amino ester obtained 4 with hydrocinnamaldehyde providing (PhCH2CH2CHO) functionalized imines. Imines 5 undergo cyclization to yield functionalized pyrrolinones (Scheme 1). The above synthetic methodology is further extended for the synthesis of bispyrrolinone (-)10 and tetrapyrrolinone (-)17 using modified resin-bound amino esters. The ozonolysis

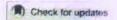
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ARTICLE



Development of a sustainable ternary magnetic nanocomposite GCNI for efficient and synergistic photodegradation of Rhodamine B under solar irradiation: kinetic and mechanistic studies

Monika Thakura, Harminder Singhb, Jaspreet Kaur Rajputc and Rupesh Kumar (5)a

^aDepartment of Chemical Sciences, I.K. Gujral Punjab Technical University, Kapurthala, INDIA; ^bDepartment of Chemistry, DAV University, Jalandhar, INDIA; Department of Chemistry, Dr. B.R. Ambedkar National Institute of Technology, Jalandhar, INDIA

ABSTRACT

In this study, a novel sunlight-active ternary magnetic nanophotocatalyst GCNI (Nanozero valent Iron @ Graphene oxide and Chitosan support) was fabricated. This easily retrievable magnetic nanocomposite was prepared by depositing NZVI (Nano Zero Valent Iron, Fe⁰) over binary GO-CS in seven different compositions for its best optimisation in the solar-light-promoted degradation of RhB. NC (Nano Composite) was characterised by SEM demonstrating their surface morphology. Further characterisation and properties of NC were also studied using XRD, FT-IR, VSM and BET and EDS techniques. The efficacy of the NC was investigated for the removal of RhB in comparison to binary GO-CS and bare NZVI. GCNI 1:1:3 was found to be the best suitable photocatalyst showing synergistic effect in the photodegradation of RhB under given conditions of pH, concentration and time. The mechanism of dye removal and degradation was further discovered by UV-VIS and mass spectroscopy techniques. FT-IR and XRD results of fresh and treated GCNI successively corroborated the mechanism. The sorption kinetics of RhB on to GCNI was found to be described by the pseudo-second-order kinetic equation. The parameters such as dye concentration, pH and the reaction time are varied to understand the effective removal of RhB in wastewater. The outcomes exhibited the highly efficient; up to 99.4% of dye removal, easy magnetic separation of photocatalyst, excellent reusability up-to 87.5% till six consecutive cycles, synergistic effect in adsorption and photo-degradation of dye.

ARTICLE HISTORY

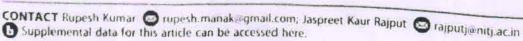
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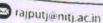
KEYWORDS

Magnetic nanocomposite; photodegradation; ternary nanocomposite; Rhodamine B; water remediation

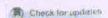
1. Introduction

Organic synthetic dyes are well-known water pollutants, as these dyes are used in textile, plastic, leather tanning and food processing industry. Discharge of huge amount of effluents from such industries causes water pollution [1,2]. These dyes are mutagenic and carcinogenic in nature which also prove to be toxic to both human and aquatic life. Also undesirable change in the colour of water causes aesthetic problem [3,4]. Various









7-Endo-trig Pictet-Spengler type cyclization of 5-alkylidene/arylidene-amino-3H-pyrimidin-4-ones: An efficient and diastereoselective synthesis of pyrimido[4,5-b] [1,4]benzodiazepines

Rayees Ahmad Naikoo^a, Rupesh Kumar^a , Parvesh Singh^b, and Gaurav Bhargava^a

Department of chemical sciences, I. K. Gujral Punjab Technical University Kapurthala, Punjab, India;
School of Chemistry and Physics, University of KwaZulu Natal, Durban, South Africa

ABSTRACT

The manuscript describes an efficient, atom economical synthesis of pyrimido[4,5-b][1,4]benzodiazepin-4-ones by relatively unexplored 7-endo-trig Pictet-Spengler type cyclisations. The synthetic methdodlogy involves the synthesis of different variants of 5-arylidene-amino-3H-pyrimidines and their p-toluene sulfonic acid mediated 7-endo-trig Pictet-Spengler type cyclisations to afford biologically relavent functionalized benzodiazepine condensed pyrimidines such as pyrimido[4,5-b][1,4]benzodiazepines in good to excellent yields (82–94%).

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KEYWORDS

Pictet-Spengler; 7-endo-trig cyclization; benzodiazepine; pyrimidinones; benzodiazepines condensed pyrimidinones

GRAPHICAL ABSTRACT

Introduction

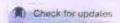
Pictet-Spengler cyclizations are among the useful synthetic methodologies for the synthesis of a variety of heterocycles. There are ample reports on the synthesis of biologically relevant heterocycles using 5/6-endo-dig Pictet-Spengler cyclisations. However, the reports on the synthesis of corresponding seven-membered heterocycles are redo-trig Pictet-Spengler cyclizations are still scarce in the literature. Hence, exploration on atom economical Pictet-Spengler cyclizations using appropriately

CONTACT Courses Briangers (2) gautavarptu at in, gauravorganic@gmail.com (2) Department of Chemical Sciences, I. K.

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SYNTHETIC COMMUNICATIONS REVIEWS



Recent developments in the synthesis of tricyclic condensed pyrimidinones

Rayees Ahmad Naikoo^a, Rupesh Kumar^a, Vipan Kumar^b, and Gauray Bhargaya^a

^aDepartment of Chemical Sciences, IKG Punjab Technical University, Kapurthala, Punjab, India: Department of Chemistry, Guru Nanak Dev University, Amritsar, Punjab, India

ABSTRACT

Functionalized tricyclicpyrimidinones, a class of condensed heterocyclic systems, are found in numerouspotentially active molecules having both innate as well as synthetic origin. Different researchers across the world have explored different synthetic methods for the construction of appropriately functionalized tricyclicpyrimidinones. The present review article recapitulates various reports pertaining to substituted and functionalized tricyclic pyrimidinones reported since

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KEYWORDS

Condensed pyrimidinones: fused pyrimidinones: tricyclic pyrimidinones

GRAPHICAL ABSTRACT

Introduction

Examplements are important beterocyclic compounds which are predominant innumer this histogically active molecules functionalized pyrimidinones have been employed as a sected template in the exploration of the apeutically active molecules. The synthesis explorations for deviang new routes for the formation of condensed perimidingness have been made comprehensively due to their wide spectrum of biological activity. Per-modulation cand trealerts alives are prevalent in DNA, RNA, vitamins, coenzymes and

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VANDANANAITHANI', ANIRUDH P. SINGH' and M.K. NAUTIYAL'

Department of Applied Sciences, Department of Research and Development, IKG Punjab Technical University, Jalandhar- Kapurthala Highway, Kapurthala-144603, Department of Genetics and Plant Breeding, College of Agriculture, G.B. Pant University of Agriculture and Technology, Pantnagar-263 145 (U.S. Nagar, Uttarakhand)

ABSTRACT: Clay samples separated from agricultural soil of Ahmadgarh, Faridkot and Zira of Punjab, India were selected for systematic study. In this paper three samples of clay which were separated from soil of Punjab, India. Clay samples were characterized by XRD to know different minerals present in them. Kaolinite, smectite, montinorillonite, illite were analyzed by XRD which were further substantiated by FTIR and TG/DTA/TGA analysis.

Key words: Clay, FTIR, illite, Kaolinite, montomorillonite, texture, TG/DTA/TGA, smectite, XRD, whole soil

Clay is an important component of soil system. The formation of clay minerals in the soil involves slow processes which depend upon the weathering environment and that is mainly controlled by climate and drainage conditions of the landscape (konta 2005; Emadadul, 2013). Depending upon the structure developed in the course of soil formation different clay minerals will exhibit different physical, chemical and mechanical behaviors. Study of the clay minerals present in soils is very important to know the role of clay minerals in various soil processes. Due to their physical and chemical properties like abundance, high specific surface area, high layer charge, laminar morphology and chemical reactivity with both neutral and charge species, these are of great importance to industry, agriculture and the environment (Murray 1999; Hoidy et al., 2009; Njoka et al., 2015). They are hydrated aluminum silicates with very fine particle size, usually <2 mm (Moore and Reynolds, 1989). Structurally, clay minerals consists of two basic units, an octahedral sheet and a tetrahedral sheet which is joined by sharing the apical oxygens or hydroxyls to form the 1:1 clay mineral layer e.g., kaolinite or the 2:1 clay mineral layer e.g. illite, smecfite, chlorite, vermiculite, gibbsite. The structure and composition of the major clays, i.e. kaolins, smectites and palygorskite-sepolite, are very different even though they are each comprised of octahedral and tetrahedral sheets as their basic building blocks. In this study, three regions of Punjab namely Ahmadgarh, Faridkot and Zira which are situated in different region of Puniab were selected. Samples of soil from these three regions were

collected from agricultural fields. In the present communication the clays have been separated from soil collected from agriculture fields which has both organic and inorganic components. These have been studied by XRD, FTIR and TG/DTA/TGA. Clay samples used were not commercial and were separated from soil of agricultural fields of Punjab, India. Analytical studies of clays of soil of Punjab have not been done earlier. The objective of the present study is to identify the mineralogy in samples of clay separated from agricultural soils of different texture. This study will help in identification of clays and understanding of their interaction behaviour with different chemicals used in agriculture.

MATERIALS AND METHODS

Three samples of whole soil were collected from different agricultural fields of Ahmadgarh, Faridkot and Zira. The soil samples were taken from the upper horizons 0-20 cm, air dried, crushed and passed through 2 mm sieve and stored in plastic bags for analysis and further treatment. The soil samples collected from Ahmadgarh, Faridkot and Zira were designated code as AHM-S, FDK-S and ZRA-S respectively. Texture analyses of the two soil samples were done by Bouyoucos hydrometer method (Bouyoucos, 1962). Clay was separated from these three samples of whole soil by centrifugation method (Tan, 1996) and was designated as AHM-C, FDK-C and ZRA-C respectively. In order to identify the minerals from clays after separation, different samples of clay were analyzed by X-ray diffraction (X-ray

Spectroscopic Characterization of Humic Acids Extracted From Different Type of Soils of Punjab

Vandana Naithani*, Anirudh P. Singh and M.K. Nautiyal1

Punjab Technical University Jalandhar, Jalandhar- Kapurthala Highway, Kapurthala, Punjab

The humic acids isolated from soils of different places of Punjab have been characterized by ultraviolet visible (UV-VIS) and Fourier transforms infrared (FTIR) spectrophotometers. The texture and chemical analyses of the whole soils were also carried out by X-ray fluorescence (XRF) spectroscopy. The molecular weight of extracted humic acids was determined by colour ratio (E₂₆₅/E₄₆₅) from absorbance values recorded in UV-VIS region. The FTIR spectra revealed the presence of different functional groups of -COOH, -OH, C-O, C=N, C-OH, etc. The texture and the chemical characteristics of the whole soils have been correlated with their humic acids. It was observed that humic acids and soils of similar textures have almost identical characteristics and molecular weights.

Key words: Humic acid, FTIR, UV-VIS, soil texture

Soil consists of inorganic components, organic matter and living organisms. The inorganic components consist of the minerals and clay. The organic matter (OM) present in soils is a mixture of products formed at decomposition stages, resulting from chemical and biological degradation of plant and animal residues and synthetic activity of microorganisms. The soil organic matter has two components, humified (humic acid, fulvic acid and humin) and non-humified (carbohydrates, lipids, amino acids, nucleic acids and lignin). Humic substances are the major constituents of soils, and generally known as biologically refractory degradation components (Chefetz et al. 2002). The humic substances are dark-coloured amorphous polymers. They are formed in the environment with biomass constituents (Stevenson 1982). It is considered that humic substances initially lose those components which are easily decomposed, and thereafter condensation-polymerisation and oxidative reactions take place which results in a more stable structure (Kumada 1987). This process corresponds to the progression of humification, and it is identified on the basis of the degree of darkening of colour of humic acids, alkali-soluble and acid-

insoluble fraction (Watanabe and Takada 2006). These substances are usually fractionated on the basis of their solubility characteristics and the pH value of solvents as well. The humic substances are noncrystalline organic elements. They are composed with 40-60% carbon, 30-50% oxygen, 3-7% hydrogen, and 1-5% nitrogen. They consist of a very complex series of carbon chains and ring structures, with numerous chemically active functional groups like phenolic, carboxylic and alcoholic hydroxyl. Martin et al. (1998) characterized the humic substances extracted from soils of varying climate and altitude. They found the chemical and structural composition of humic acids (HAs) and fulvic acids (FAs) are influenced by climate and altitude of the region. The dominant effect of annual rainfall and vegetation on the composition of HAs and FAs has also been reported by Arshad and Schnitzer (1989). They found that humic substances of higher, wetter and cooler altitudes were lesser humified than those extracted from lower, warm and drier sites. At highest altitude, the nitrogen contents in HAs from both cultivated and natural forest soils were lowest perhaps due to accelerated leaching during the rainy period (1417 cm annual rainfall). Studies revealed that the characteristics of humic substances like functional group, molecular weights (Mukhopadhyay and Banerjee 1985) are too influenced by cultivation and vegetation type.

Clay and sand sediments have many differing physical and chemical characteristics, and

¹College of Agriculture, G.B. Pant University of Agriculture and Technology, Pantnagar, 263 145, Udham Singh Nagar, Uttarakhand



^{*}Corresponding author (Email: naithanivandana@gmail.com)
Present address



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Influence of microwave processing and sintering temperature on the structure and properties of Sr/Zr doped hydroxyapatite



Ravinder Kumar Chadha^a, Anirudh P. Singh^{a,*}, Kanchan L. Singh^{b,**}, Chetan Sharma^a, Vandana Naithani^a

ABSTRACT

HIGHLIGHTS

- Sr-Zr hydroxyapatite has been sintered by microwave and conventional methods.
- · The lattice parameters increase with temperature during conventional sintering.
- Reverse trend for lattice parameters is observed during microwave sintering.
- Microwave sintered products have higher density and hardness.
- Microwave sintered product at 1100 °C has a less percentage of secondary phase.

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Keywords:
Calcium hydroxyapatite
Microwave sintering
Conventional sintering
X-ray diffraction

Among the different cations that are substituted to modify the properties of calcium hydroxyapatite, Sr and Zr have gained interest because of their effect on bone formation and mechanical properties, respectively. In the present work, Sr and Zr substituted hydroxyapatite, Ca_{9.38}Zr_{0.11}Sr_{0.5}(PO₄)₆(OH)₂ are synthesized by conventional and microwave sintering at different temperatures. The effect of different sintering techniques and temperatures are investigated on the structural and morphological properties of as-synthesized samples. The X-ray diffraction (XRD) and Fourier transform infrared spectroscopy (FT-IR) is used to study the variation in structural properties and scanning electron microscopy (SEM) is used to investigate the morphology with the variation in sintering conditions. The changes in lattice parameters clearly show that Sr and Zr are incorporated into hydroxyapatite lattice. Microwave sintered samples retain the higher amount of hydroxyapatite and experience higher density levels as compared to conventional sintering. As compared to conventional sintering, the grain size is more uniform and low percentage of secondary phase is observed in microwave sintered sample.

1. Introduction

Increasing number of accidents, trauma, arthritis, and tumors have increased the interest of the scientific community to find new biomaterials [1–3]. Many research groups are working for the development and evaluation of new synthetic biomaterials for bone grafting. The main challenge in biomaterial engineering is the quest to find a material which has good biocompatibility, resorbability and mechanical strength [4–6]. Many materials are tried and tested, but they are found inappropriate regarding the above-mentioned properties. The most appropriate material to date is calcium hydroxyapatite (HA) [7–9]. HA is considered as a better option as compared to other calcium phosphates

due to its biocompatibility and osteoconductive properties. It also exhibits excellent chemical and biological affinity with the bone tissue [10]. HA is basically calcium hydroxyapatite which makes most of the mineral part of bone and tooth enamel. But its stability in vivo and low mechanical strength as compared to bone has limited its use in load-bearing applications [11,12]. To improve its mechanical properties, many approaches like the addition of dopants and different synthesizing routes have been investigated [13]. The best structural aspect of HA is its ability to accommodate a great variety of cations and anions while retaining its hexagonal structure. These substitutions can alter the thermal stability, surface reactivity and biological properties of HA [14–16].

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[&]quot; I.K.G. Punjab Technical University, Jalandhar, India

b D.A.V. Institute of Engineering and Technology, Jalandhar, India

^{*} Corresponding author.

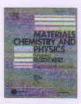
^{**} Corresponding author.



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Comparison of microwave and conventionally sintered manganese and niobium doped lanthanum germanate based apatites by micro Raman Spectroscopy



Chetan Sharma ^a, Singh Kanchan L.^{b,**}, Singh Anirudh P.^{a,*}, Payal Sharma ^{a,b}, Sonia Mago ^a, Vandana Naithani ^a, Ravinder Kumar Chadha ^a

HIGHLIGHTS

- · Sintering of apatites have been done by conventional and microwave processing.
- The XRD revealed that the sintered products had single Phase hexagonal oxy apatite.
- Raman Spectroscopy revealed that monoclinic phase was also present in the apatites.
- · Microwave sintered products had uniform grain growth and higher density.
- · Microwave sintered products had higher hardness and conductivity value.

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ABSTRACT

Keywords: Micro Raman spectroscopy XRD SEM and hardness In this paper, the effect of microwave and conventional sintering on doping behaviour of Mn²⁺ and Nb⁵⁺ in lanthanum germanate based apatite have been compared. The precursor of apatite has been prepared by mixed oxide method and conventionally calcined at 1100 °C for 6 h. The resulting powders have been sintered by microwave energy at 1400 °C for 30 min and by conventional heating 1400 °C for 4 h. The products prepared have been characterized by X-ray diffraction, scanning electron microscope, Micro Raman spectroscopy, and Vicker's hardness and the results are compared. The apatites of similar compositions sintered by microwave energy have higher density and hardness than the apatites sintered by electric heating.

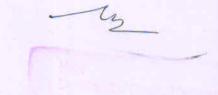
1. Introduction

The ideal general formula of apatite is $A_{10}M_6O_{24}X_2$ (A = alkaline earth, rare earth; M = P, Si Ge) and (X = OH, O and halide). Its applications ranging from biomaterial to electrolytes have been of interest and therefore highly researched [1]. Hassan et al. have written a review on microwave-assisted techniques for the synthesis of nano hydroxyl apatite but they did not cover the sintering of hydroxyapatite [2]. Transparent hydroxyapatite was prepared first time in the Microwave Processing & Engineering Center of Material Research Institute, Penn State by the group of Agrawal. They had used fine crystalline material prepared by hydrothermal process and microwave sintering of the fine powder at 1150 °C within 5 min of soaking time [2–5]. In another work,

the group used microwave processing in the preparation of HAP/ZrO₂ composites to avoid loss of calcium phosphate. They could prepare HAP/ZrO₂ by microwave processing at the temperature as low as 1200 °C within 20 min of sintering time. Conventionally HAP/ZrO₂ was prepared at 1200 °C in 240 min which led to the loss of calcium phosphate [6]. The sintering of hydroxyapatite to be used as a biomaterial has been reported by the group of Singh [7].

In the year 1995, Nakayama et al. investigated lanthanum silicate-based apatites $La_{10}(SiO_4)_6O_3$, (Ln=La, Nd, Sm, Gd, Dy, Y, Ho, Er and Yb) for its applications as an ionic conductor [8] and later lanthanum germanates was also studied for the applications as ionic conductors [9] to be used at intermediate temperature in the solid oxide fuel cells (SOFC). Since the $La_{9.33+2x/3}(Si/Ge)_6O_{26+x}$ apatites have a conductivity

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[&]quot; L.K.G Punjab Technical University, Kapurthala, 144601, India

Department of Applied Sciences, D.A.V Institute of Engineering and Technology, Jalandhar, 144003, India

^{*} Corresponding author.

^{**} Corresponding author.

E-mail address: deanric@ptu.ac.in (A.P. Singh).



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Phytoremediation of Pulp and Paper Mill Effluents by Salvinia molesta and its comparison with Nanoparticles Inclusive Phytoremediation

Punit Bhardwaj*, Jyotsna Kaushal**, Vandana Naithani*** Anirudh P Singh****

* Research Scholar, I.K.G. Punjab Technical University, Jalandhar, Punjab

** Dean & Professor in Chemistry, Chitkara Institute of Engg. & Technology, Jhansla, Rajpura, Punjab.

***Asstt. Professor in Department of Chemical Science, Main campus, Chemistry, I.K.G. Punjab

Technical

University, Jalandhar, Punjab.

****Dean (R & D), I.K.G. Punjab Technical University, Jalandhar, Punjab.

Abstract

Due to industrialization, there is increased disposal of pollutants such as heavy metals, various types of organic and inorganic compounds into the environment. These toxic pollutants affect soil, surface water, and ground water which can cause serious risks to health of human beings. Many physicochemical processes such as screening, coagulation, activated carbon treatment, ion-exchange, electrodialysis, trickling filteration, reverse osmosis and activated sludge digestion have been used for the treatment of polluted water. But all these methods involve high energy and large capital investments. Therefore, phytoremediation has been employed successfully to clean up pollution which is emerging technology. In this paper, the Phytoremediation of Pulp and Paper Mill Effluent using Salvinia molesta has been assessed in terms of reduction in pH, EC,BOD, COD, TS,TSS and TDS and compared the results with nanoparticles inclusive phytoremediation. The effluent has been treated for 45 days. All parameters exhibited exponential decrease from the start up to 45 days and thereafter showed negligible decrease till the termination of the experiment. The proposed nanoparticles inclusive phytoremediation method can be profitably employed for the abatement of pollution from industrial waste water.

Keywords: Phytoremediation, Pulp And Paper Mill Effluent, Salvinia molesta, Silver Nanoparticles.

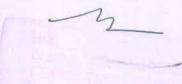
Introduction

Pulp and paper industry is the major industry of our country. It cosumes a large amount of water [1]. The ministry of environment and forest, Government of India has categorized the pulp and paper industry as one of the twenty most polluting industries[2]. Approximately 75 to 95% of total pollution load is contributed by small paper mills. The discharged effluents contain organic, inorganic pollutants and colouring materials. These chemicals may affect soil and in turn the growth and development of plants[3]. Marwaha et al reported that the pulp and paper industry waste water might cause oxygen deficiency, acute toxicity, eutrophication in recipient waste water bodies[4]. Various investigators had studied the effect of paper mill effluent on various crops [5-9]. There is decline in germination of rice seeds and seedling growth with paper mill effluent treatment in comparison to control[10]. Rajannan and Oblismai [11] reported that the germination of rice, black gram and tomato seeds was effected by paper mill effluent. However, the diluted form of effluent (25 and 50%) enhance the growth.

Pulp and paper mill effluents contain heavy metals and other organic toxicants which are hazardous to the environment[12]. Studies on assessment of Phytoremediation potential of water hyacinth and Trapa bispinosk of pulp and paper mill waste water have been carried

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Effect of microwave and conventional processing techniques on mechanical properties of Strontium substituted hydroxyapatite

Ravinder Kumar Chadha^a, Kanchan L. Singh^b, Chetan Sharma^a, Anirudh P. Singh^{a,*}, Vandana Naithani^a

ARTICLEINFO

Keywords: Microwave and conventional sintering XRD Micro Raman spectroscopy Microhardness

ABSTRACT

The effect of different sintering routes: conventional and microwave sintering along with different sintering temperatures has been studied on Sr-doped hydroxyapatite bioceramics. The physical and mechanical properties of different samples synthesized at different sintering temperatures and sintering techniques have been evaluated by X-ray diffraction (XRD), Raman spectroscopy, Fourier transform infrared spectroscopy (FT-IR), scanning electron microscopy and microhardness. X-ray diffraction studies showed that microwave sintering suppresses the formation of secondary phases. The density, grain size along with microhardness was higher for samples sintered by microwave processing as compared to samples sintered by conventional technique. The results from different characterization techniques show that microwave sintering technique is not only efficient in terms of energy-saving and time reduction but also improves the formation of phases and mechanical properties.

1. Introduction

Biomaterials are those materials which are meant to interact with biological systems and are used in medical devices [1]. Advancements in biomaterials have significantly increased the lifetime of patients suffering from trauma, injuries, large bone defects, and osteoporosis. But, still, enhancements are required to improve the physicochemical and biological properties of biomaterials to mimic the natural bone [2,3].

For implantation and other orthopedic issues, Bioceramics are widely accepted due to their useful properties like non-toxicity and excellent biocompatibility [4]. Among different bioceramics, calcium phosphate-based bioceramics constitute a great promise for bone tissue engineering as they resemble the mammalian bone and teeth [5]. Single-phase hydroxyapatite (HA) or biphasic calcium phosphate (BCP) comprising β -tricalcium phosphate (TCP) is an interesting candidate for synthetic bone substitution applications [6,7]. Hydroxyapatite provides a scaffold for osteoconduction and β - TCP oversaturates the local environment with Ca²⁺ and PO₄³⁻ ions to speed up the new bone formation [8]. HA/ β -TCP ratio predicts the extent and rate in vivo resorption of biphasic calcium phosphate, higher β -TCP content allows rapid and substantial resorption [9]. However, the solubility of HAp is

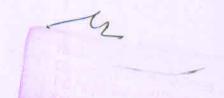
poor as compared to other bioceramics which affect the rate of biodegradation in vivo and increases patient's rehabilitation time [10]. The solubility of bioceramics is affected by chemical composition, crystallinity, particle size, surface area, pore size, density/porosity and extent of ionic substitution in HAp lattice [11–13]. To improve the mentioned properties, several methods like variation of sintering techniques and temperature, granule size, and electrical polarity have been proposed [14]. Besides these methods, ionic substitution is also considered to improve the solubility of HAp and hence osteoconductivity [15].

The biological apatite contains trace ions like Na⁺, Sr²⁺, Mg²⁺, Zn²⁺, SiO₄⁴⁻, SO₄²⁻, F etc. and is non -stoichiometric [16]. These ions play a vital role in bone metabolism. But, synthetic hydroxyapatite is stoichiometric and deficient of these ions which make it a poor bone repair material [17]. But, the properties of synthetic hydroxyapatite can be improved by ionic substitution. After ionic substitution, HAp more closely mimics the biological HAp [18]. From the above-mentioned ions, Sr²⁺ ion helps to maintain the balance between the bone formation and resorption process in vivo and low dosage of Sr²⁺ helps in increasing bone mass and volume. It also accelerates the osteogenic differentiation of mesenchymal stem cells and enhances the functionality of osteoblasts [19].

Synthesis of hydroxyapatite has been reported by several different

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^a I.K.G. Punjab Technical University, Jalandhar, India

b D.A.V. Institute of Engineering and Technology, Jalandhar, India

^{*} Corresponding author.

E-mail address: deanricptu@gmail.com (A.P. Singh).

Estimation of Genetic Variability in *Jatropha Curcas* L. Genotypes for Vegetative Traits and Seed yield

Bhawesh Joshi¹, MK Nautiyal¹, Leela Bhatt², Vikas Mangal¹* and Vandana Naithani³

¹Department of Genetics and Plant Breeding, G.B, Pant University of Agriculture and Technology, Pantnagar-263145 Uttarakhand, India

²Department of Vegetable Science, G.B, Pant University of Agriculture and Technology, Pantnagar-263145 Uttarakhand, India
³Department of Applied Sciences, IKG Punjab Technical University, Jalandhar, Kapurthala Highway, Kapurthala-144603

Abstract

Twenty five Jatropha curcas genotypes collected from the foothills of the Shivalik range of the Himalayas in a narrow belt called Tarai were grown in Randomized Block Design with three replications at Medicinal Plant Research and Development Centre, G. B. Pant University of Agriculture and Technology, Pantnagar, Uttarakhand and evaluated for 8 characters. The genotypes showed significant differences in most of traits. The phenotypic coefficient of variation was the highest for seed yield/ plant, followed by 100 seed weight (gm) and collar diameter (cm). Similarly, the genotypic coefficient of variation was the highest for seed yield/plant, followed by 100 seed weight (gm) and collar diameter (cm). Very high level of broad sense heritability was observed for seed yield /plant, 100 seed weight, Number of Secondary branches/plant, Collar diameter, Flower initiation (days) after 31st July, Number of Primary branches/plant, Plant Height, Oil content. High estimates of genetic gain as per cent mean were recorded for Flower initiation (days) after 31st July, Oil content, while low estimates of genetic gain as per cent were recorded for plant Height.

Selection for Number of Primary branches/plant, Number of Secondary branches/plant, collar diameter, seeds/fruit and 100 seed weight would result in some improvement in yield. Thus, the ideotype to achieve high yield in Jatropha should have more number of flowering bunches/plant and fruits per plant and moderate to high value for seeds/fruit and 100 seed weight.

Keywords: Genetic Variability, Vegetative Traits, *Jatropha curca*, Seed yield

*Correspondence

Author: Vikas Mangal

Email: vikas90bharatpur@gmail.com

Introduction

Jatropha curcas Linn. (Physic nut or Ratanjot) is a crop tree-borne oil seed crop. It is native to Tropical America and belongs to family Euphorbiaceae. Global increase in demand for renewable energy to combat the greenhouse effect and rapid depletion of ozone layer as a result of discharge of harmful gases into the environment, couple with the depletion of reserved fossil fuel has mandated the use of biomass energy feedstock for sustainable production of biofuel. Biofuel has been known to be a good alternative to fossil fuel due to its cheap, sustainable and environmental friendly properties [1]. This newly introduced crop which grows abundantly in wild and abandoned land has its seed and oil yield unpredictable especially in tropical climate. Favorable environmental conditions that affect its production has yet to be known [2] and [3]. In spite of the great potentials and attributes of Jatropha as a biodiesel crop, the full potentials of Jatropha have not been realized. One of the reasons for this, apart from the agronomic, social economic and institutional constraints is the facts that there is presently no planned rational conventional breeding and genetic programs. [4] reported that, for the fact that Jatropha has adapted itself to wide range of environmental and ecological conditions suggests that, there exists considerable amount of genetic diversity yet to be detected for potential realization. [5] reported variability observed in J. curcas in Central India which was mainly limited to seed source variation in morphology, germination and seedling growth. [6] have reported divergence in seed oil traits of J. curcas from a limited number of locally collected accessions. Since heritability estimate in perennial plants decrease with age because of compounded environmental effects masking genotypic differences, there is an urgent need for the knowledge of these estimates in the progenies.

However, one of the limitations to this conventional breeding programme work is the wide genetic diversity, which has to be evaluated before selection of suitable accessions could be possible. Also breeding of high yielding variety is not possible without knowing the extent of variation that exists among the available germplasm in India, the growth, yield and extent of environmental influence on these factors, heritability and genetic gain of the material.

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Microwave and conventional processing of niobium and manganese doped lanthanum germanate based apatites



Chetan Sharma^a, Kanchan L. Singh^{b,*}, Anirudh P. Singh^{a,**}, Vandana Naithani^b, Payal Sharma^{a,b}, Sonia Mago^a, Ravinder Kumar Chadha^a

A.K.G Punjab Technical University, Kapurthala, 144601. India

HIGHLIGHTS

- In this paper, apatites have been prepared by microwave heating as well as conventional heating.
- The single phase of the apatite could be prepared at 1100 °C within 1 h by microwave heating.
- By the conventional heating, the single phase of the apatite could not be prepared at 1100 °C within 1 h.
- The grain growth in the microwave sintered products was more uniform than the conventionally sintered products.
- The density of the sintered products was as high as 96% of the theoretical density.

ARTICLEINFO

Keywords: Apatite Electrolyte Microwave processing XRD SEM

ABSTRACT

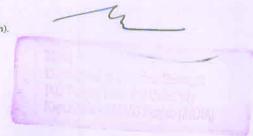
The paper reported calcination and sintering of apatites of compositions $La_{9.5}Ge_{5.8}Nb_{0.5}O_{26.5}$ and $La_{9.5}Ge_{5.8}Nn_{0.5}O_{26}$ using microwave energy for the first time from the precursor prepared by the mixed oxide method. The calcination was done conventionally under an identical condition to compare the results. The single phase of $La_{9.5}Ge_{5.5}Nb_{0.5}O_{26.5}$ and $La_{9.5}Ge_{5.5}Mn_{0.5}O_{26}$ could be obtained by microwave processing at temperature $1100\,^{\circ}$ C within 1 h. But the single phase of aptites could not be prepared by conventional heating under the identical conditions. The calcined powder obtained by microwave heating was sintered by microwave heating as well as conventional heating. The products obtained were characterized by XRD and SEM. It was observed that the microwave sintered products had higher density and uniform microstructure compared to the sintered products obtained by the conventional heating.

1. Introduction

Yttria-stabilized zirconia (YSZ) exhibits high oxygen ion conductivity at high temperatures in the range of 850–1000 °C. Therefore, YSZ is used as an electrolyte in SOFC. Since YSZ shows appreciable conductivity only at elevated temperature consequently, the operation of SOFC causes problems related to the degradation of materials and mechanical and chemical compatibility in oxidizing and reducing atmospheres [1]. Many other oxide electrolytes are being studied to lower the operating temperature of the SOFC. Some of the oxide ion conductors are fluorite, perovskite, pyrochlore, brownmillerite, and apatite [2–7]. Lanthanum oxide, due to its multivalent valence state, is used as a catalyst, photocatalyst [8,9], and in the fabrication of supercapacitor [10]. Lanthanum has also attracted attention because lanthanum doped

hydroxyapatite has improved performance as dental implants in osteoporotic cases [11]. Further, lanthanum-based apatite, $\rm Ln_{10-x}Si_6O26 \pm y~(Ln=La)~(LSA)$, has found its application as an electrolyte in the intermediate temperature solid oxide fuel cell (IT-SGFC) because of its higher conductivity (of the order of $10^{-3}~\rm Scm^{-1}$) at 500 °C which is higher than that of YSZ at an intermediate temperature [12–16]. Lanthanum silicate based apatite requires higher sintering temperature which is usually higher than 1700 °C and a long soaking time of more than 4 h before they can be used as an electrolyte in SOFCs [17]. The substitution of Si ions by Ge ions has been found to decrease sintering temperature and improve electrical conductivity [18,19] at the intermediate temperature. But during the preparation of lanthanum germanate based electrolyte, the GeO2 gets evaporated at a higher temperature which changes the La/Ge stoichiometric ratio resulting in

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Department of Applied Sciences, D.A.V Institute of Engineering and Technology, Jalandhar, 144003, India

^{*} Corresponding author.

^{**} Corresponding author.

E-mail addresses: kanchan_69@rediffmail.com (K.L. Singh), anips123@rediffmail.com (A.P. Singh).



ORIGINAL RESEARCH

Vermicompost and vermiwash as supplement to improve seedling, plant growth and yield in *Linum usitassimum L*. for organic agriculture

Cinny Makkar3 - Jaswinder Singh3 - Chander Parkash3

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Abstract

Purpose To study the organic production in two varieties of Linuas (Linseed) crop using different proportions of vermicomposts and combination of vermiwash as an environment friendly substitute to chemical fertilizers for organic farming of Linseed.

Method Vermicompost prepared from cattle flung was mixed in different proportions, i.e. 0, 20, 40, 60, 80, and 100% vermicompost v/v with soil and commercial porting media (Perlite, Peat and Coconut corr). To study the germination, vegetative growth, reproductive growth and yield in two varieties of Linum using somem L. LC-54 and LC-2063 were grown in a polyhouse for a 2 years trial. A total of 960 seedlings and 240 plants were studied for assessment of result in 24 treatments. Additional treatments with total application of 1.1 v/v vermiwash and recommended dose of inorganic fertilizer were done to determine and compare the role of vermicomposi, vermiwash and morganic treatment.

Results Substituting soil with 60% (v/v) vermicompost in LC-54 and with 40% (v/v) vermicompost in LC-2063 improved the performance of seeds, root morphology and stem growth. Life cycle of crop was shortened in two

varieties of Linseed raised under different treatments of vermicompost and vermiwash. Performance of LC-54 was better at 60% and that of LC-2063 was at 40%, which can be attributed to their genotypic differences. Vermicompost proportion above 60% was of no use at any stage of life cycle in both the varieties of Linum. Germination and yield were further improved when foliar application of vermiwash was integrated with vermicompost for nutrient management when compared with recommended chemical fertilizer treatment and control.

Conclusion Bio-efficacy of vermicompost and vermiwash for production of organic Linseed crop has been well established with enhanced yield.

Keywords Agriculture Earthworm Linseed Organic farming Plant growth Vermicompost

Introduction

Organic foods have recently guined importance in developed, as well as developing countries with the growing awareness about adverse effects of agricultural chemicals on human health. Demand for organic food has increased worldwide which has led to increased interest in organic agriculture (Follet et al. 1981). Sinha et al. 2009). Crops with a label of 'organically grown' fetch much higher value to farmers in domestic and international market (Lim et al. 2015, Crowder and Regunold 2015). Organic agriculture also promotes ecological conservation due to sustainable use of natural resources (Regunold et al. 1993; Letourneau and Goldstein 2001; Mäder et al. 2002). In cultivation of organically grown crops, chemical usage is excluded at all stages so organic agriculture broadly provides the dual benefit of soil quality improvement and chemical free

Cinny Makkar consymathetra@gmail.com

Chander Parkash chandes plu@gmail.com

- Department of Applied Sciences, LK G Penjab Technical University, Kapurthala, Punjab, India
- Department of Zoology, Kladsa College, American Punjab, India





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RESEARCH ARTICLE



Modulatory role of vermicompost and vermiwash on growth, yield and nutritional profiling of Linum usitatissimum L. (Linseed): a field study

Cinny Makkar 1 - Jaswinder Singh 2 - Chander Parkash 1.3

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Abstract

Vermicorripost, recommended inorganic fertiliser, vermiwash and their combinations were used in the present study to know their impact on the germination, growth, yield, oil content and nutritional status in seeds of two varieties of Linum unturnsylmans L. LC-54 and LC-2063. Eight treatments, in 48 plots with three replicates, were studied and compared with control. Application of vermicompost enhanced growth and yield in both varieties of Limin. Vermicompost modulated the ratio of to-3 and to-6 fatry acids in seeds of Linseed. Growing degree days (GDD) indicated the significantly shortened of crop life cycle with vermiwash. Vermiwash particularly increased the yield indices. Fatty acid profile through GC-FID showed the increase in polyunsaturated fatty acids (PUFA), monounsaturated fatty acids (MUFA) and saturated fatty acids (SFA) with Integrated Nutrient Management (INM). Different effects on nutritional status of LC-54 and LC-2063 seeds can be attributed to their unique genotypes. INM proved to be nutritionally balanced strategy with enhanced yield leading to better soil health. INM can be hypothesised as a transitional step to the organic agriculture after the green revolution, when compared with recommended inorganic fertiliser treatment and control. In terms of oil yield, highest oil content 34,85% and 33,67% was obtained with Integrated Nutrient Management in both the varieties, whereas treatments with vennicompost and venniwash produced modulated and most suited ratio of omega-3 and onwga-6 in Linseed seeds.

Keywords Agriculture - Earthworm - Organic farming - Omega-3 fatty acids

Introduction

During green revolution, the land of Punjab (India) had been left hunggier due to excessive use of chemical fertilisers (Shiva 1991). Some regions of India had been affected by serious diseases like caucer due to excessive use of chemical fertilisers, insecticides, pesticides etc. (Halder 2009). People are now getting aware about the harmful effects of conventional morganic produce since past few decades. Thus, the demand of organic products at consumer outlets has drastically increased (Crowder and Reganold 2015). Organic agriculture promotes human health with ecological conservation and sustainable use of natural resources (Leiourneau and Goldstein 2001; Maeder et al. 2002). Land and water resources need to be restored by adopting traditional or other organic practices (Hardin 2011). In organic produce, no chemicals are used at any stage of agricultural practices, thus, soil health is improved in addition to providing organic foods. (Thouastiff 2014).

Use of vermicompost as an organic soil amendment is one such practice in current scenario. Studies are going on to establish the positive impact of vermicompost on plant growth under greenhouse conditions or field conditions (Jashi et al. 2013; Zuo et al. 2018). Studies establish that vermscompost has enhanced the morpho-physiological parameters and plant growth in pea (Pisum sanvum) (Maji et al. 2017); chickpea (Civer arietinum) (Yadav and Garg 2015) and lettuce (Khosravi et al. 2018). Vermscompost addition significantly increased the crop yield indices in groundant (Arachis

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- 154 Jasveinder Single singlipassi75@yabdo.co.iii
- Department of Applied Sciences, LK-tr. Punjah Technical University Johnshie, Punjab, India
- Department of Zoology, Khaisa College, Amritian, Puejab, India
- Department of Chemical Sciences, LK G. Purjub Technical University, Johandhar, Puritio, India

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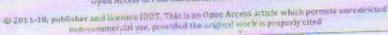


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Research Article

Synthesis, Characterization and Evaluation of Anticancer Activity of Nanoresveratrol in B16 Melanoma Cell Line

Arora Reena 1 & Parkash Chander 19

Department of Chemical Sciences, 18th Punjah Technical Sinderesty, Raporthala, Funjah

ABSTRACT

Reservation, a member of the stilleone family, is a naturally occurring polyphonolar compound in grapes and other hotanicals. It carries a Reservation, a member of use someone family, is a caturally occurring polyposites of composed in grapes and object including chemogree-rative potential to prevent or treat several chross deserves, including cancers. Despite virtues promising health benefits, including chemogree-rative potentials to prevent or treat several chross transfer to prevent of the poor brownstability, and and chemotherapsuite benefits against cancer chross translation of reservative is a improving the resveration profile. Calcium phosphate leaded stately, Here we present the study to prepare nano-formulation of resverative for improving the resveration profile. sanstray, sere or present the smay as people transfer and the artifered for store effects were examined in a B14 marries molanisms cell ben. The sanstrayer artifects of name overallost were directly compared with free resocrated and chemotherapy doing 5-Fit. Nameresversire) induced anti-rancer effects of name overallost were directly compared with free resocrated and chemotherapy doing 5-Fit. Nameresversire) induced appears to the victorius more officiently as compared to reaver and amore and \$ FU. A nuclear staining profile following Buschst staining of Bis cells is indicative of apoptonic and shows the the amicroscope effect of nanocessveratrol is due to apoptonic induction. Since antications activities of separated discording and catalage activities of separated discording and catalage activities of separated discording and catalage. amountedant exagines. We found that, resourcest with the anti-tancer activity the automobile activity of nanovesserated increased Collectively. these findings clearly show that encountered is superior to free resperatrol in suppressing cell growth in \$16 musine melanoms cell line, and that the anti-cancer selectly is primarily through apoptoess suduction.

Keywords: Resperated Nanoresverated Formulation, 816 cell line, Cancer, Mekinsuma, Chemographenters, Chemotherapeonic

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*Address for Correspondence:

Or Chander Parkard, Institution Name, Department of Chemical Sciences, LK, C. Pumpab Technical University, Kapurthala, Punjab 144603, India

1. INTRODUCTION

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(chemically: 3.5.4-trihydroxystilliene) member of stableno family is a naturally occurring polypherotic compound present in grapes, berries, peanuts and red wine. It is a very well-known antioxidant and exhibits various biological offects including anticancer, antiangiogenic, immunomodulatory and cardioprotective activities [1,2,3], its nemeficial potential in a variety of the chemoproxective indications includes chemotherapeutic effects in several cancers [4]. Rescentitrol has been reported to have a role in reducing the growth of human cancers originating from diverse organs/sites. including breast, cervix, ovary, uterine, kidney, liver, eye, bladder thyroid oesophagus prostate, brain, long, skin, gastric, colon, lisad and neck, buse and blood [5-13]. Several studies indicate that resvecatrol shows multiple anticancer effects such as protection against currour initiation and progression, apoptous, antioxidant, anti-CHREET angiogenesis, anti-inflammatory, and ambumor migration [6-13]. Evidence show that resversitral exerts its effects through different signalling pathways and several molecular players. Various studies report that susverstrol has a range of cellular effects, including cell cycle, invasion/metastasis.

angiogenests, apoptosis and autophagy. Several molecular players, such as BCL-2/BAX and AMPK, PI3K/ART/mFOR WNT, MEE, ERK pathways mediate resveratrol-induced cellular effects in different cancers [14]. However, the precise mechanisms remain to be delineated. There are several reports indicating the promising effects of resveratral in treatment of slan cancer [15,16,17]. A recent report shows that resveratrol induces mitochondrial apoptosis in melanoma cell line through downregulating Bci-Z, activating BAX leading to apoptosis [13]. Additionally, this report also showed that pyruvate kinase PKM2 played a role in apoptosis through interacting with Rel-2.

Despite widely established promising health benefits of resveratrol and its potential as a chemo preventive and chemotherspeutic agent in cancer based on in vitro and preclinical studies against several human cancers, its clinical translation remains challenging. Key challenges associated with the therapeutic application of resveration include poor solubility, stability and limited bioavailability [19,20].

Nanotechnology offers a wide range of innevative strategies to address the challenges associated with resveratrol. Plant metabolites can be incorporated into nane structures so that

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Probing interactions of neurotransmitters with twin tailed anionic surfactant: A detailed physicochemical study



Rajwinder Kaura, Reshu Sananb, Rakesh Kumar Mahajana,*

^a Department of Chemistry, Guru Nanak Dev University, Amritsar 143005, India ^b P.G. Department of Chemistry, Khalsa College, Amritsar 143001, India

G R A P H I C A L A B S T R A C T



Region I: Formation of ion-pairs

Region II: Formation of induced micelles

Region III: Formation of micelles with adsorbed NTs

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ABSTRACT

Keeping in view the role of neurotransmitters (NTs) in central nervous system diseases and in controlling various physiological processes, present study is aimed to study the binding of neurotransmitters (NTs) such as norepinephrine hydrochloride (NE) and serotonin hydrochloride (5-HT) with twin tailed surfactant sodium bis(2-ethylhexyl)sulfosuccinate (AOT). Spectroscopic and electrochemical measurements combined with microcalorimetric measurements were used to characterize the interactions between AOT and NTs. Meteoric modifications to emission profile and absorption spectra of NTs upon addition of AOT are indicative of the binding of NTs with AOT. Distinct interactional states such as formation of ion-pairs, induced and regular micelles with adsorbed NTs molecules have been observed in different concentration regimes of AOT. The formation of ion-pairs from oppositely charged NTs and AOT is confirmed by the reduced absorbance, quenched fluorescence intensity and decrease in peak current ($f_{\rm pa}$) as well as shifts in peak potential ($E_{\rm pa}$) values. The stoichiometry and formation of the NTs-AOT complexes has been judged and the extent of interactions is quantitatively discussed in terms of binding constant (K) and free energy of binding (ΔG°). The enthalpy ($\Delta H^{\circ}_{\rm mic}$) and free energy of micellization ($\Delta G^{\circ}_{\rm mic}$) for AOT in presence and absence of NTs are determined from the enthalpy curves.

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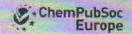


* Corresponding author. E-mail address: rakesh_chem@yahoo.com (R.K. Mahajan).

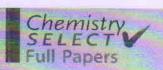
1. Introduction

Mental health issues are now more common in our society and is in the throes of a virtual epidemic of depression whose numbers are quite staggering. Major depressive disorder (MDD) is a

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Electro, Physical & Theoretical Chemistry

Biamphiphilic Ionic Liquids-Drug Mixtures: Interactional and Morphological Aspects

Rajni Vashishat, [a] Reshu Sanan, [c] Debes Ray, [d] Vinod Kumar Aswal, [d] and Rakesh Kumar Mahajan*[a, b]

The interactional and morphology behavior of mixed system embracing chlorpromazine hydrochloride (CPZ) drug and surface active biamphiphilic ionic liquids (BAILs); 1-hexyl-3-methylimidazolium dodecylsulfate ([Comim][C12OSO3]) and 1-butyl-3methylimidazolium octylsulfate ([Camim][CaOSOa]) at pH 7 have been investigated to develop new strategies to use BAILs as carriers of pharmaceutical active compounds. UV-vis absorption spectroscopy and cyclic voltammetry (CV) methods are used to shed light on the foremost interactions governing CPZ + BAILs mixtures. The higher value of binding constant (Ka) and Gibbs free energy change (ΔG) for CPZ + [C₆mim][C₁₂OSO₃] mixture, determined from absorbance measurements depict the existence of more attractive interactions than CPZ + [Camim] [C₈OSO₃]. The composition mediated self-assembly of CPZ + BAIL mixed aggregates in bulk has been investigated by small angle neutron scattering (SANS), dynamic light scattering (DLS) and zeta potential measurements.

Introduction

Chlorpromazine hydrochloride (CPZ) the nitrogen-containing heterocyclic drug, belongs to phenothiazine family. The CPZ widely used as psychopharmacological agent, as a peripheral vasodilator, tranquilizer, to control aggression, anxiety, tension, excitement and psychomotor disturbances in patients suffering from schizophrenia.[1-3] However, CPZ has some minor side effects such as movement problems, sleepiness, and low blood pressure upon standing and dry mouth. Along with minor side effects, the major side effects of CPZ are, permanent movement disorder, neuroleptic malignant syndrome and low level of white blood cells.[4,5] The amphiphilicity of CPZ is essentially governed by hydrophobic tricyclic ring linked to hydrophilic portion propyl chain having charged amino group which make them good enough molecules to bind membrane systems causing disruption and solubilization, similar to surfactants. [6,7]

Due to its amphiphilicity, CPZ forms micelles in solution which would have an effect on the nature of its interactions in biological systems. A delicate balance lies between the stabilization of amphiphiles by micelle formation and the degree of steric, repulsion that is encountered. [8,9] This balance manifests itself in the size and shape of the micelle. CPZ is often regarded as a model drug for the investigation of interactions between drugs and biological or model membranes. Phenothiazine drugs aggregate in a micelle-like manner with the value of N_{agg} (aggregation number) being of the order of 6 to 15.[10,11] The binding of CPZ with membranes is an issue of major concern because it depicts the mechanism of drug action. In organisms the pharmacodynamics mechanism involves the interactions of small amount of drug with large amount of biomacromolecules or biomembranes. Thus, the amphiphilic drugs are beneficial for understanding the association between molecular architecture and the physicochemical properties. The biological activity of CPZ is the consequence of its binding with biological molecules. [12,13]

In pharmacy, the interactions between drugs and surfactants are important as the undesirable side effects of drugs can be eliminated when the drugs are used along with surfactants as mixed micelles. [14] Surfactants molecules mimic the complex biomembrane environments due to their surface and structural properties. The importance of drug + surfactant interactions based on the fact that an extreme amount of medication causes excess stimulation and psychotic illness.[15,16] Consequently, targeted drug delivery in body organs is compulsory. Moreover the oppositely charged drug + surfactant mixed system possess great synergism and have been shown to form various molecular assemblies such as, spherical micelles, rods, worm like micelles, bilayer, vesicles, ribbons disks etc. [17,18]

There are number of reports published based on the interactions of drugs with surfactants.[19-24] Ghosh et al. have investigated the interactions of common nonsteroidal anti-

[b] R. K. Mahajan Vice Chancellor **DAV** University Jalandhar - 144001, India

[c] R. Sanan Department of Chemistry Punjab Technical University Kapurthala - 144601, India

[d] D. Ray, V. K. Aswal Solid State Physics Division Bhabha Atomic Research Centre Mumbai 400085, India

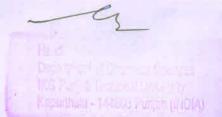
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[[]a] R. Vashishat, R. K. Mahajan Department of Chemistry UGC-Centre for Advanced Studied Guru Nanak Dev University Amritsar - 143005, India E-mail: rakesh_chem@yahoo.com

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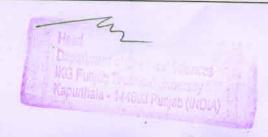


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